

## TCHEA4: TCS High Entropy Alloy Database

<i>Database name:</i>	TCS High Entropy Alloy Database	<i>Database acronym:</i>	TCHEA
<i>Database owner:</i>	Thermo-Calc Software AB	<i>Database version:</i>	4

TCHEA4 is a thermodynamic database for high entropy alloys (HEA) [2004 and 2006, Yeh], which are also known as multi-principal element alloys (MPEA) [2013, Wang; 2015, Senkov]. TCHEA4 is developed in a CALPHAD spirit based on the critical evaluation of all the binary systems and many ternary systems. A hybrid approach of experiments, first-principal calculations and CALPHAD modeling has been used to obtain reliable thermodynamic descriptions of the BCC, FCC and HCP solutions. That enables predictions to be made for multi-component alloy systems, especially for HEAs.

### Included Elements (26)

The database has been developed in a 26-element framework:

Al	B	C	Co	Cr	Cu	Fe	Hf	Ir	Mn	Mo	N	Nb	Ni	Re
Rh	Ru	Si	Sn	Ta	Ti	V	W	Y	Zn	Zr				

Almost all binary systems in this 26-element framework (307 in total) are assessed to their full range of composition and temperature and can be calculated with the BINARY Module in Thermo-Calc. In total, 493 ternaries are assessed, and 192 of them to the full range of composition and temperature. These can be calculated with the TERNARY Module in Thermo-Calc.

The database can be used to calculate various phase diagrams and property diagrams in the assessed systems and higher-order systems.

- ▶ [TCHEA4 Assessed Binary Systems](#)
- ▶ [TCHEA4 Critically Assessed Ternary Systems](#)

The extrapolation to higher-order systems helps to understand the phase equilibria in HEAs, so as to predict the phase formation, phase fractions and phase compositions or to calculate the driving force of forming a phase. The database can also be used to predict solidification behaviour of HEAs with the SCHEIL\_GULLIVER module in Thermo-Calc. All necessary molar volume data and thermal expansion data have been assessed or estimated for most of the phases.

TCHEA4 contains nearly all stable phases in all assessed binary systems and most ternary systems. In total, 500 solution and intermetallic phases are modeled.

[Phases Included in TCHEA4](#) lists all phases and [Models for the Phases Included in TCHEA4](#) includes a detailed description of the models, e.g. number of sublattices and constituents on each sublattice.

▶ Also see [TCHEA4 Calculation Examples](#) for a variety of validation information.

## Limits

As in the spirit of the CALPHAD method, predictions can be made for multicomponent systems by extrapolation into multicomponent space of data critically evaluated and assessed based on binary, ternary and in some cases higher order systems. However, critical calculations must always be verified by equilibrium experimental data; it is the user's responsibility to verify the calculations but Thermo-Calc Software AB is interested to know about any significant deviations in order to improve any future release.

## Database Revision History

If you are interested in the revision history for this or other databases, the information is available in the online help (from Thermo-Calc go to **Help>Online Help**) or in the release notes on our [website](#). For the [TCFE \(TCS Steel and Fe-alloys\) database](#) there is a dedicated page with the history of its development.

## TCHEA4 Assessed Binary Systems

These are the assessed binary systems (307 in total) in the full range of composition and temperature.

	Al	B	C	Co	Cr	Cu	Fe	Hf	Ir	Mn	Mo	N	Nb	Ni	Re	Rh	Ru	Si	Sn	Ta	Ti	V	W	Y	Zn	Zr
B	x	B																								
C	x	x	C																							
Co	x	x	x	Co																						
Cr	x	x	x	x	Cr																					
Cu	x	x	x	x	x	Cu																				
Fe	x	x	x	x	x	x	Fe																			
Hf	x	x	x	x	x	x	x	Hf																		
Ir	x	x	x	x	x	x	x	x	Ir																	
Mn	x	x	x	x	x	x	x	x	x	Mn																
Mo	x	x	x	x	x	x	x	x	x	x	Mo															
N	x	x		x	x	x	x	x		x	x	N														
Nb	x	x	x	x	x	x	x	x	x	x	x	x	Nb													
Ni	x	x	x	x	x	x	x	x	x	x	x	x	x	Ni												
Re	x	x	x	x	x	x	x	x	x	x	x				Re											
Rh	x	x	x	x	x	x	x	x	x	x				x	x	Rh										
Ru	x	x	x	x	x	x	x	x	x	x	x		x	x	x	x	Ru									
Si	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x		x	Si								
Sn	x	x	x	x	x	x	x	x	x	x			x	x	x	x	x	x	Sn							
Ta	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x		x	x	x	Ta						
Ti	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	Ti					
V	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x		x	x	x	x	x	V				
W	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x		x	x	x	x	x	x	W			
Y	x	x	x	x	x	x	x	x	x		x		x	x	x		x	x	x	x	x	x	x	Y		
Zn	x	x	x	x	x	x	x	x	x	x	x		x	x	x	x	x	x	x	x	x	x	x	x	Zn	
Zr	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	x	Zr									

updated

added

## TCHEA4 Critically Assessed Ternary Systems

These are the assessed ternary systems in the full range of composition and temperature.

Al-B-Ti	Al-Nb-Ni	C-Co-W	C-Nb-Ti	Cr-Cu-Nb	Cu-Fe-Ni
Al-C-Co	Al-Nb-Ti	C-Cr-Fe	C-Nb-W	Cr-Cu-Si	Cu-Fe-Si
Al-C-Fe	Al-Ni-Ru	C-Cr-Hf	C-Ni-Ta	Cr-Cu-Sn	Cu-Fe-Sn
Al-Co-Ni	Al-Ni-Si	C-Cr-Si	C-Ni-Ti	Cr-Cu-Zr	Cu-Fe-Ti
Al-Co-Ti	Al-Ni-Ta	C-Cr-Ti	C-Ni-W	Cr-Fe-Mn	Cu-Fe-V
Al-Co-W	Al-Ni-Ti	C-Cr-V	C-Ni-Zr	Cr-Fe-Mo	Cu-Mn-Ni
Al-Co-Zr	Al-Ni-W	C-Cr-Zr	Co-Cr-Cu	Cr-Fe-N	Cu-Mn-Si
Al-Cr-Ni	Al-Ni-Zn	C-Cu-Fe	Co-Cr-Fe	Cr-Fe-Ni	Cu-Mn-Sn
Al-Cr-Sn	Al-N-Ti	C-Fe-Mn	Co-Cr-Ni	Cr-Fe-Si	Cu-Mn-Zn
Al-Cr-Ti	Al-Ru-Ti	C-Fe-Mo	Co-Cr-Ti	Cr-Fe-V	Cu-Mo-Ni
Al-C-Ti	Al-Si-Sn	C-Fe-N	Co-Cr-W	Cr-Mn-N	Cu-Ni-Ti
Al-Cu-Fe	Al-Si-Ti	C-Fe-Ni	Co-Cu-Fe	Cr-Mn-Ti	Cu-Ni-Zn
Al-Cu-Mn	Al-Si-Zn	C-Fe-Si	Co-Cu-Mn	Cr-Mo-Ni	Cu-Ti-Zr
Al-Cu-Ni	Al-Sn-Ti	C-Fe-Ti	Co-Cu-Nb	Cr-Nb-Ni	Fe-Mn-N
Al-Cu-Si	Al-Sn-Zn	C-Fe-V	Co-Cu-Ni	Cr-Ni-Re	Fe-Mn-Si
Al-Cu-Sn	Al-Ta-Ti	C-Fe-W	Co-Fe-Mo	Cr-Ni-Si	Fe-Mo-Ni
Al-Cu-Zn	Al-Ti-V	C-Hf-Mo	Co-Fe-N	Cr-Ni-Ta	Fe-Nb-Ni
Al-Fe-Mn	Al-Ti-W	C-Hf-Ni	Co-Fe-Ni	Cr-Ni-Ti	Fe-Ni-Ru
Al-Fe-N	Al-Ti-Y	C-Mn-Si	Co-Fe-W	Cr-Ni-W	Fe-Ni-Si
Al-Fe-Si	Al-Ti-Zr	C-Mo-Ni	Co-Mo-Ti	Cr-Ni-Zr	Fe-Ni-Ti
Al-Fe-Ti	C-Co-Cr	C-Mo-Ta	Co-Ni-Si	Cr-N-Ni	Fe-Ni-W
Al-Mn-Ni	C-Co-Fe	C-Mo-Ti	Co-Ni-V	Cr-Si-Ti	Fe-N-Nb
Al-Mn-Si	C-Co-Mo	C-Mo-V	Co-Ni-W	Cr-Ti-Zr	Fe-N-Ni
Al-Mn-Ti	C-Co-Nb	C-Mo-W	Co-Sn-Ti	C-Ta-W	Fe-N-Ti
Al-Mn-Zn	C-Co-Ni	C-Mo-Zr	Co-Ta-Ti	C-Ti-W	Fe-N-V
Al-Mo-Ni	C-Co-Ta	C-Nb-Ni	Co-W-Zr	Cu-Fe-Mn	Fe-Ti-V
Al-Mo-Ti	C-Co-Ti	C-Nb-Re	Cr-Cu-Mo	Cu-Fe-N	Hf-Nb-Si

---

Hf-Ni-Ti
Ir-Rh-Ru
Mn-Si-Zn
Mo-Nb-Ti
Mo-Ni-Ta
Mo-N-Ni
Mo-Ta-Ti
Mo-Ti-V
Mo-Ti-W
Mo-Ti-Zr
Nb-Ni-Ti
Nb-Sn-Ti
Nb-Ta-Ti
Nb-Ti-V
Nb-Ti-W
Nb-Ti-Zr
Ni-Si-Ti
Ni-Ta-Ti
Ni-Ta-W
Ni-Ti-W
Ni-Ti-Zr
N-Ni-Ti
Re-Ta-W
Si-Ti-W
Ta-Ti-V
Ta-Ti-W
Ta-Ti-Zr
Ti-V-W
Ti-V-Zr
Ti-W-Zr

## TCHEA4 Tentatively Assessed Ternary Systems

Al-C-Cr	Al-Hf-Ru	C-Cr-Mn	Co-Cr-Ru	Co-Re-W	Cr-Mo-N
Al-C-Ni	Al-Hf-Si	C-Cr-Mo	Co-Cr-Si	Co-Ru-Ta	Cr-Mo-Nb
Al-Co-Cr	Al-Hf-Ti	C-Cr-N	Co-Cr-Ta	Co-Ru-W	Cr-Mo-Re
Al-Co-Hf	Al-Mo-Nb	C-Cr-Nb	Co-Cr-V	Co-Si-Ta	Cr-Mo-Ru
Al-Co-Mo	Al-Mo-Re	C-Cr-Ni	Co-Cu-Ti	Co-Si-Ti	Cr-Mo-Si
Al-Co-Nb	Al-Mo-Ru	C-Cr-Re	Co-Fe-Hf	Co-Si-W	Cr-Mo-Ta
Al-Co-Ru	Al-Mo-Si	C-Cr-Ta	Co-Fe-Nb	Co-Si-Zr	Cr-Mo-Ti
Al-Co-Si	Al-Mo-W	C-Cr-W	Co-Fe-Ta	Co-Ta-W	Cr-Mo-W
Al-Co-Ta	Al-Mo-Zr	C-Fe-Nb	Co-Fe-Ti	Co-Ti-Zr	Cr-Mo-Zr
Al-Cr-Fe	Al-Nb-Re	C-Fe-Re	Co-Fe-Zr	Cr-Cu-Fe	Cr-Nb-Re
Al-Cr-Mo	Al-Nb-Ru	C-Fe-Ta	Co-Hf-Ni	Cr-Cu-Ni	Cr-Nb-Si
Al-Cr-Nb	Al-Nb-Si	C-Hf-Ta	Co-Hf-Ti	C-Re-Ta	Cr-Nb-Ta
Al-Cr-Re	Al-Nb-Ta	C-Hf-Ti	Co-Mo-Nb	C-Re-V	Cr-Nb-Ti
Al-Cr-Ru	Al-Nb-W	C-Hf-W	Co-Mo-Ni	C-Re-W	Cr-Nb-W
Al-Cr-Si	Al-Ni-Re	C-Mn-V	Co-Mo-Re	Cr-Fe-Hf	Cr-Nb-Zr
Al-Cr-Ta	Al-Ni-V	C-Mo-N	Co-Mo-Ru	Cr-Fe-Nb	Cr-Ni-Ru
Al-Cr-W	Al-Ni-Zr	C-Mo-Re	Co-Mo-Ta	Cr-Fe-Re	Cr-Ni-V
Al-Cr-Zr	Al-Re-Ru	C-Mo-Si	Co-Mo-V	Cr-Fe-Ta	Cr-N-Nb
Al-C-Si	Al-Re-Ta	C-Nb-V	Co-Nb-Ni	Cr-Fe-Ti	Cr-N-V
Al-Fe-Hf	Al-Re-Ti	C-Nb-Zr	Co-Nb-Si	Cr-Fe-W	Cr-Re-Ru
Al-Fe-Mo	Al-Re-W	C-Ni-Ti	Co-Nb-Ta	Cr-Fe-Zr	Cr-Re-Ta
Al-Fe-Nb	Al-Ru-Ta	C-Ni-V	Co-Nb-Ti	Cr-Hf-Mo	Cr-Re-V
Al-Fe-Ni	Al-Ru-W	C-N-Nb	Co-Nb-W	Cr-Hf-Nb	Cr-Re-W
Al-Fe-Re	Al-Ru-Zr	C-N-Ti	Co-Ni-Ru	Cr-Hf-Ni	Cr-Re-Zr
Al-Fe-Ta	Al-Si-Zr	Co-Cr-Hf	Co-Ni-Ta	Cr-Hf-Re	Cr-Ru-Ta
Al-Fe-W	Al-Ta-W	Co-Cr-Mo	Co-Ni-Ti	Cr-Hf-Si	Cr-Ru-Ti
Al-Fe-Zr	C-Co-Re	Co-Cr-Nb	Co-Ni-Zr	Cr-Hf-Ta	Cr-Ru-W
Al-Hf-Ni	C-Co-V	Co-Cr-Re	Co-Re-Ta	Cr-Hf-W	Cr-Si-Ta

Cr-Si-W	Fe-Mo-Ti	Hf-Ni-Ta	Nb-Ni-Ta	Re-W-Zr
Cr-Si-Zr	Fe-Mo-W	Hf-Ni-W	Nb-Ni-V	Ru-Si-Ti
Cr-Ta-Ti	Fe-Mo-Zr	Hf-Re-Ta	Nb-Ni-W	Ru-Ta-Ti
Cr-Ta-W	Fe-Nb-Re	Hf-Re-W	Nb-Ni-Zr	Ru-Ta-W
Cr-Ta-Zr	Fe-Nb-Si	Hf-Ru-Ti	Nb-Re-Ta	Ru-Ti-Zr
Cr-Ti-V	Fe-Nb-Ta	Hf-Ru-Zr	Nb-Re-Ti	Si-Ta-Zr
Cr-Ti-W	Fe-Nb-Ti	Hf-Si-Ta	Nb-Re-W	Si-Ti-Zr
Cr-W-Zr	Fe-Nb-W	Hf-Si-Ti	Nb-Re-Zr	Si-W-Zr
C-Si-Ti	Fe-Nb-Zr	Hf-Si-W	Nb-Ru-Si	Ta-W-Zr
C-Ta-Ti	Fe-Ni-Ta	Mn-Ni-Si	Nb-Si-Ti	
C-Ti-Zr	Fe-Ni-V	Mn-Ni-V	Ni-Re-Ta	
Cu-Fe-Mo	Fe-Ni-Zr	Mo-Nb-Ni	Ni-Re-W	
Cu-Fe-Nb	Fe-Re-Ti	Mo-Nb-Re	Ni-Re-Zr	
Cu-Ni-Si	Fe-Re-W	Mo-Ni-Re	Ni-Ru-Ta	
C-V-W	Fe-Re-Zr	Mo-Ni-Ru	Ni-Ru-Ti	
C-W-Zr	Fe-Si-Ta	Mo-Ni-Si	Ni-Ru-W	
Fe-Hf-Mo	Fe-Si-Ti	Mo-Ni-W	Ni-Ru-Zr	
Fe-Hf-Nb	Fe-Si-W	Mo-Ni-Zr	Ni-Si-Ta	
Fe-Hf-Ni	Fe-Si-Zr	Mo-N-V	Ni-Si-V	
Fe-Hf-Re	Fe-Ta-Ti	Mo-Re-Ru	Ni-Si-W	
Fe-Hf-Si	Fe-Ta-W	Mo-Re-Ta	Ni-Si-Zr	
Fe-Hf-Ta	Fe-Ta-Zr	Mo-Re-Ti	Ni-Ta-Zr	
Fe-Hf-Ti	Fe-Ti-W	Mo-Re-V	Ni-W-Zr	
Fe-Hf-W	Fe-W-Zr	Mo-Re-W	N-Ti-V	
Fe-Hf-Zr	Hf-Mo-Ni	Mo-Re-Zr	Re-Ru-Ta	
Fe-Mn-Ni	Hf-Mo-Si	Mo-Ru-Si	Re-Ru-Ti	
Fe-Mo-N	Hf-Nb-Ni	Mo-Ru-Ta	Re-Ru-W	
Fe-Mo-Nb	Hf-Nb-Re	Mo-Ru-W	Re-Ta-Ti	
Fe-Mo-Re	Hf-Ni-Re	Mo-Si-Zr	Re-Ta-V	
Fe-Mo-Si	Hf-Ni-Ru	Nb-Ni-Re	Re-Ta-Zr	
Fe-Mo-Ta	Hf-Ni-Si	Nb-Ni-Si	Re-Ti-W	

# TCHEA4 Calculation Examples

## FCC Forming Ternary Systems

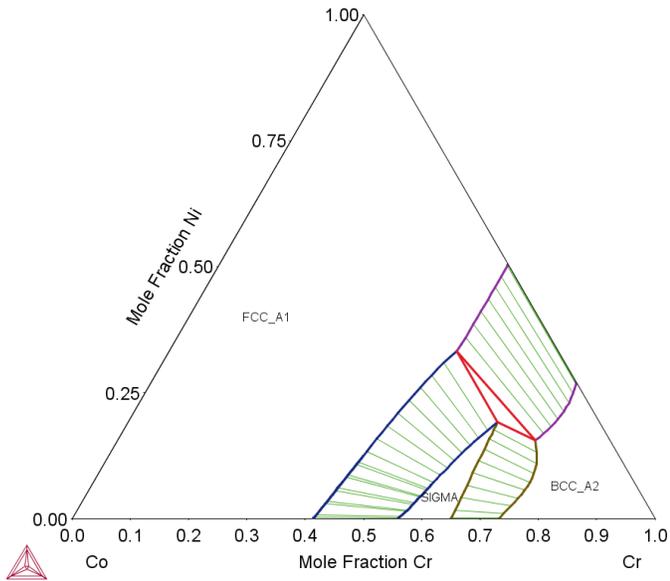


Figure 1: Calculated Co-Cr-Ni isothermal section at 1200 °C.

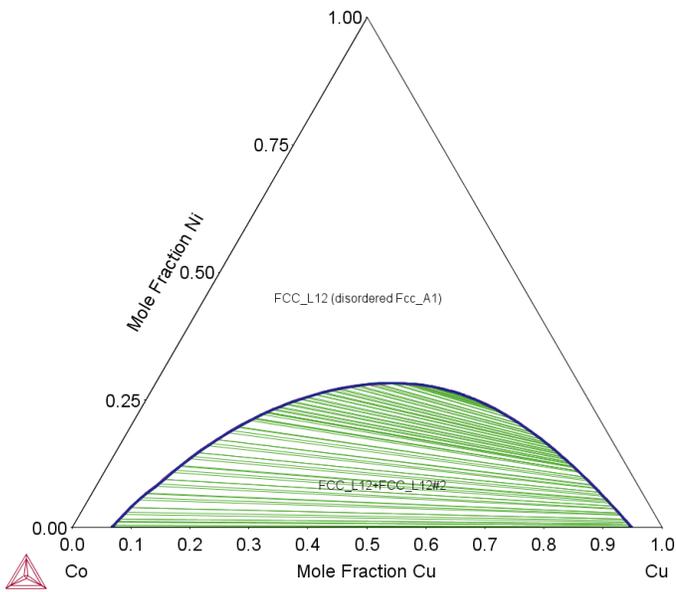


Figure 2: Calculated Co-Cu-Ni isothermal section at 1000 °C.

## BCC Forming Ternary Systems

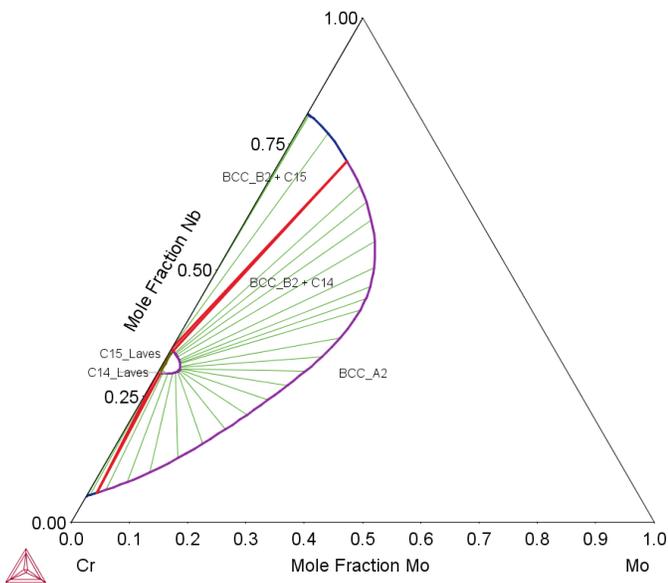


Figure 3: Calculated Cr-Mo-Nb isothermal section at 1500 °C.

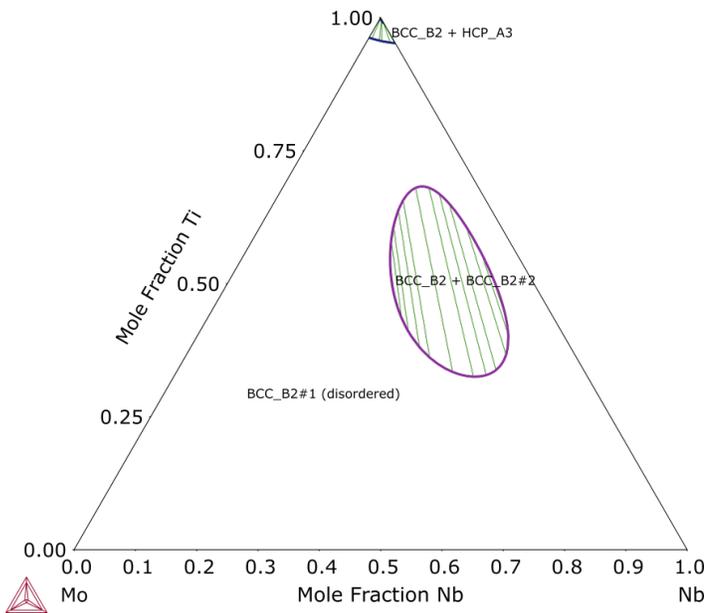


Figure 4: Calculated Mo-Nb-Ti isothermal section at 800 °C.

## FCC/FCC+BCC Forming Ternary Systems

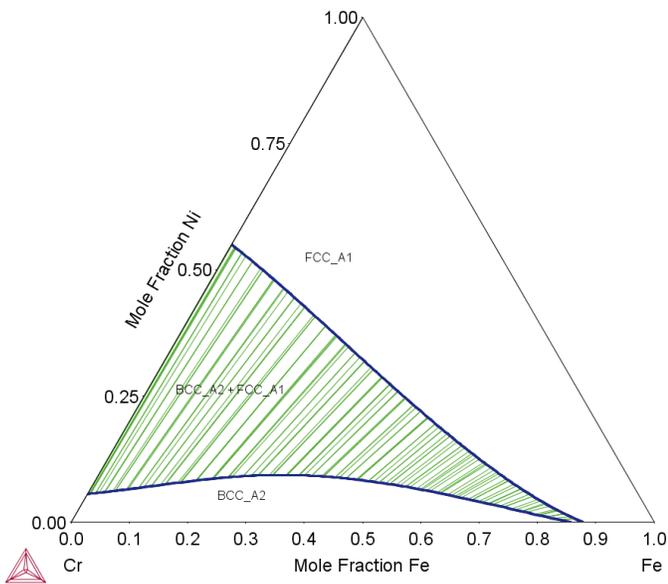


Figure 5: Calculated Cr-Fe-Ni isothermal section at 1000 °C.

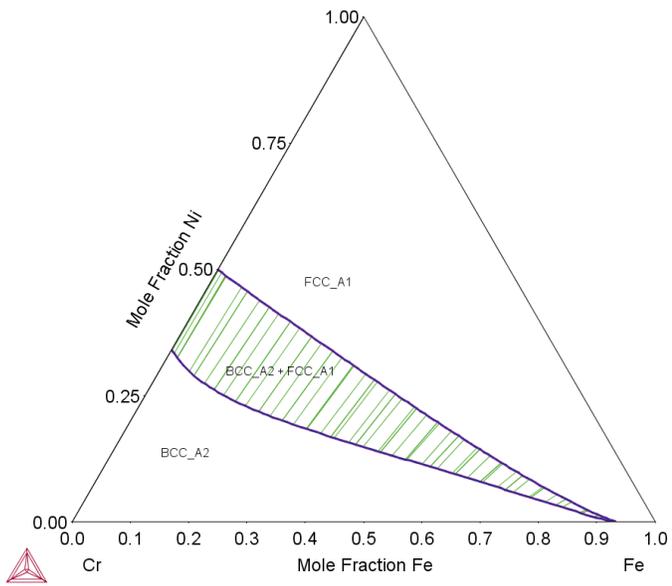


Figure 6: Calculated Cr-Fe-Ni isothermal section at 1300 °C.

## FCC+BCC Forming Ternary Systems

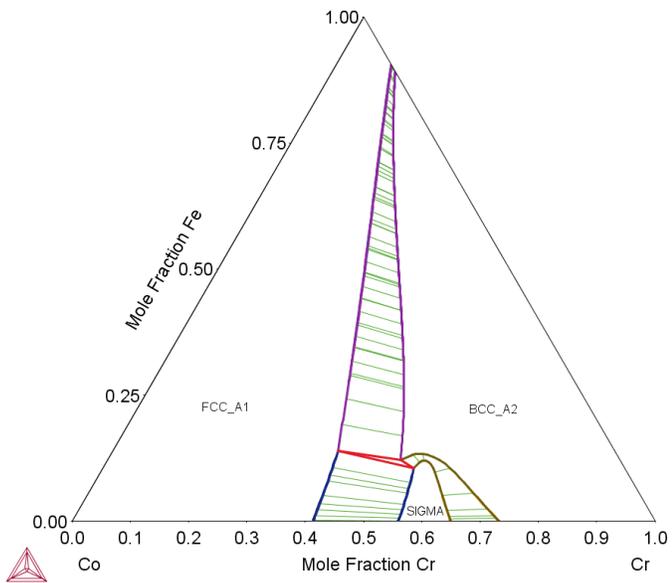


Figure 7: Calculated Co-Cr-Fe isothermal section at 1200 °C.

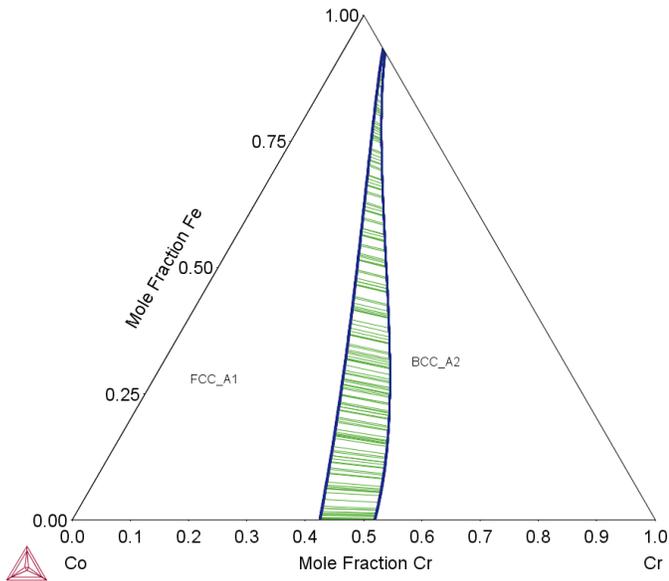


Figure 8: Calculated Co-Cr-Fe isothermal section at 1300 °C.

## Multi-Component Alloy Systems

[Figure 9](#) shows the effect of the Al addition on the phase formation in the CoCrFeNi-bases high entropy alloy. With a small amount of Al addition (<10 at.% Al), only FCC forms in as-cast alloys. The region roughly from 10 to 18 at.% Al is near the eutectic reaction, where both FCC and BCC can form in as-cast alloys. With an Al addition above 18 at.%, only BCC forms. It should be noted, however, that the phase formation also depends on the actual experimental conditions, especially the cooling rate and the heat treatment. The data from Wang [2012] indicate the compositions of the investigated as-cast alloys and the observed phases. The data from Kao et al. [2011] are the composition limits for forming BCC, FCC and BCC+FCC in as-cast alloys or alloys homogenized at 1000 °C.

[Figure 10](#) and [Figure 11](#) show the phase formation in the Co-Cr-Fe-Mn-Ni equiatomic alloy from the equilibrium calculation and the Scheil calculation, respectively. They indicate that a single FCC solid solution can be obtained during solidification at different cooling rates (from equilibrium solidification to extremely non-equilibrium solidification). The single FCC state remains during a wide temperature ranged down to 600 °C. Cantor et al [2004] and Otto et al. [2013] observed only FCC in as-cast alloys. The single phase state remained after the alloy had been annealed at 1000 °C or 850 °C for three days [2013, Otto].

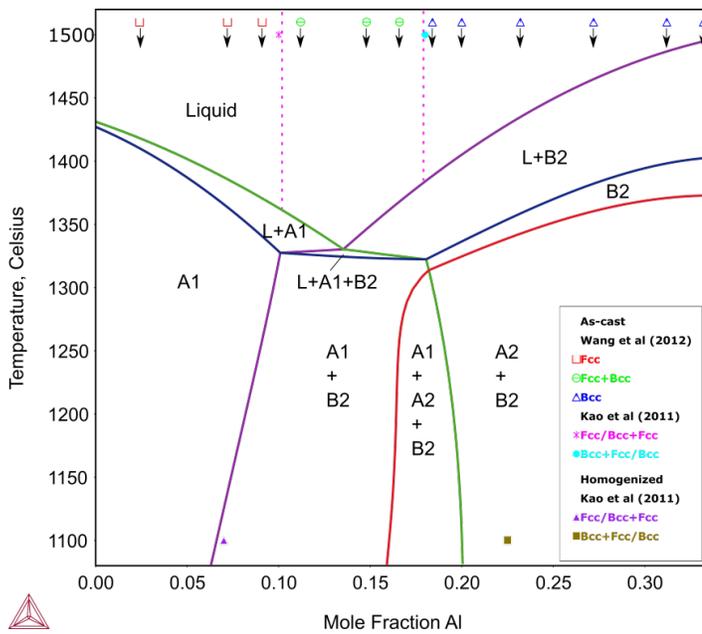


Figure 9: Calculated phase diagram along the composition line of CoCrFeNi-Al.

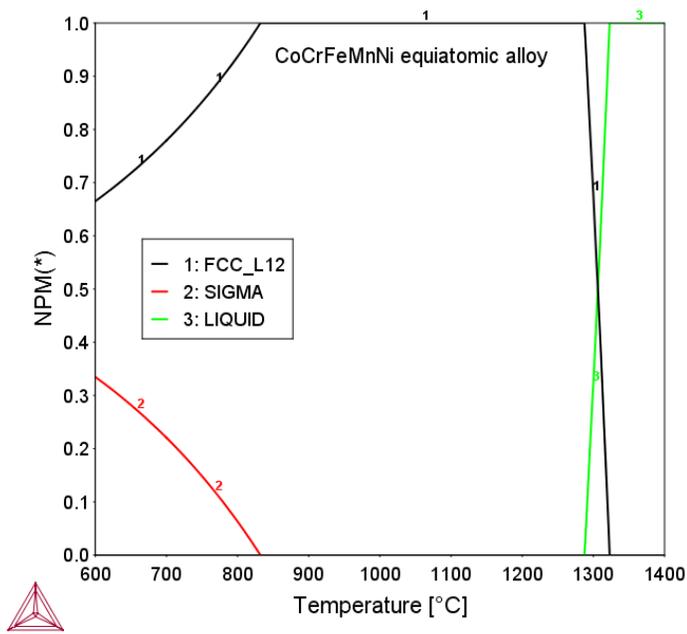


Figure 10: Equilibrium calculation of the phase formation in the Co-Cr-Fe-Mn-Ni equiatomic alloy.

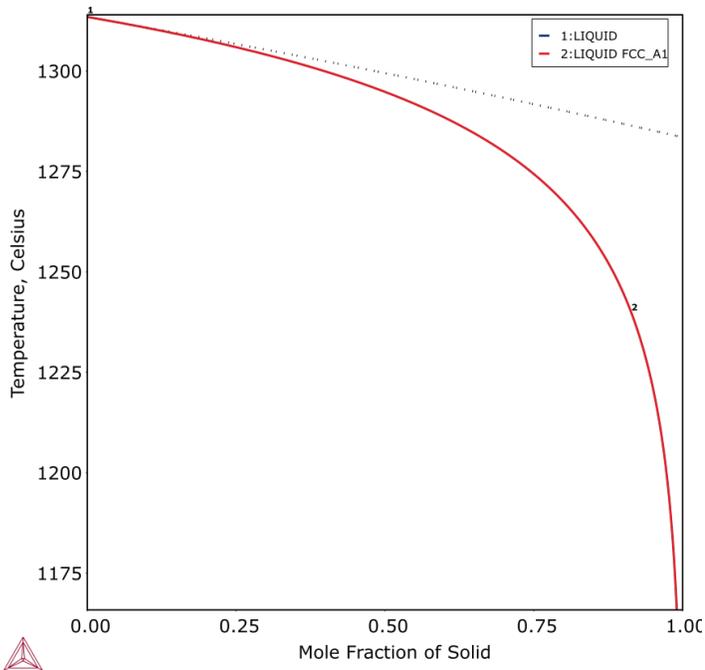


Figure 11: Scheil calculation of the phase formation in the Co-Cr-Fe-Mn-Ni equiatomic alloy.

Cantor et al. [2004] had investigated as-cast alloys prepared by melting spinning and reported that adding Cu into the Co-Cr-Fe-Mn-Ni alloy destabilized FCC dendrites and caused an interdendritic segregation with no second phase. According to the present calculation, as shown in [Figure 12](#), however, the interdendritic region actually corresponds to a Cu-rich secondary FCC phase. In other words, there is a miscibility gap of FCC. Note that the conclusion of no second phase by Cantor et al. [2004] had been drawn merely based on the poor XRD pattern, which was not sufficient to identify two sets of FCC phases. [Figure 12](#) suggested a peritectic type transformation from the primary FCC and liquid to the secondary FCC, but the amount of liquid involved in the transition is much more than that of the primary FCC. The phase formation was dramatically affected by the composition segregation during the fast cooling, the accumulation of Cr in the residual liquid caused the BCC phase to form. The amount of BCC was calculated to be less than 1 %, so it might be overlooked during the microstructure examination. Moreover, the formation of BCC might be suspended during melting spinning, due to insufficient diffusion in both liquid and FCC and/or inefficient nucleation of BCC.

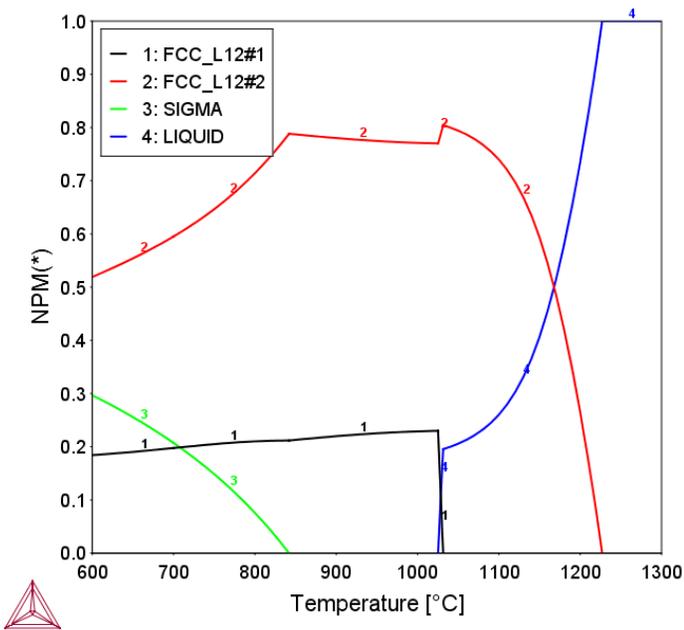


Figure 12: Equilibrium calculation of the phase formation in the Co-Cr-Cu-Fe-Mn-Ni equiatomic alloy.

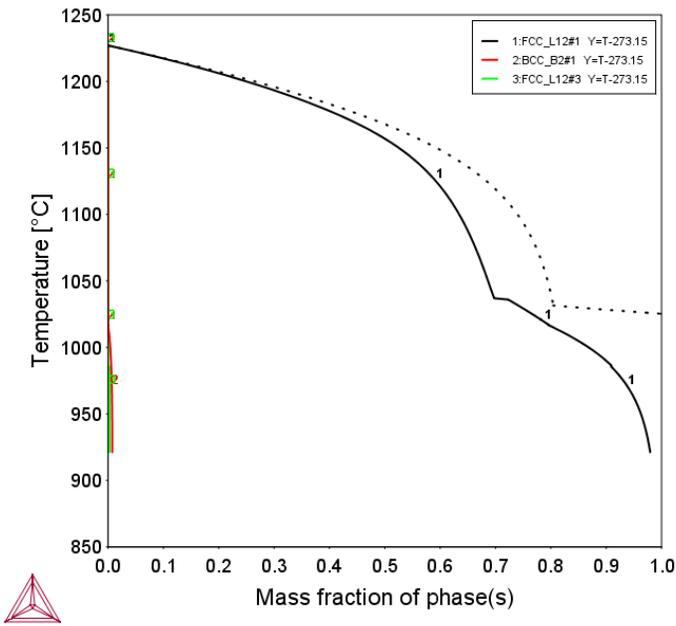


Figure 13: Scheil calculation of the phase formation in the Co-Cr-Cu-Fe-Mn-Ni equiatomic alloy.

Figure 14 shows that single BCC\_A2 solid solution can be obtained in as-cast equiatomic Al-Cr-Mn-Ti-V alloys. The BCC single phase remains in a wide temperature range down to 629 °C, where it starts to order to form the B2 phase.

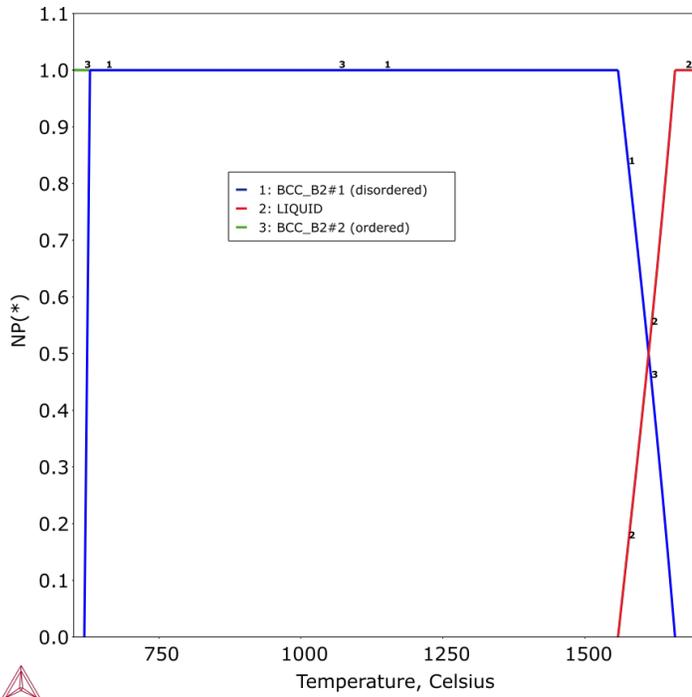


Figure 14: Equilibrium calculation of the phase formation in the Al-Cr-Mn-Ti-V equiatomic alloy.

---

## TCHEA4 References

---

### References in the text

- [2004, Cantor] B. Cantor, I. T. H. Chang, P. Knight, A. J. B. Vincent, Microstructural development in equiatomic multicomponent alloys. *Mater. Sci. Eng. A*. 375–377, 213–218 (2004).
- [2004, Yeh] J. W. Yeh et al., Nanostructured High-Entropy Alloys with Multiple Principal Elements: Novel Alloy Design Concepts and Outcomes. *Adv. Eng. Mater.* 6, 299–303 (2004).
- [2006, Yeh] J.-W. Yeh, Recent progress in high-entropy alloys. *Ann. Chim. Sci. des Matériaux*. 31, 633–648 (2006).
- [2011, Kao] Y.-F. Kao et al., Electrical, magnetic, and Hall properties of Al<sub>x</sub>CoCrFeNi high-entropy alloys. *J. Alloys Compd.* 509, 1607–1614 (2011).
- [2012, Wang] W.-R. Wang et al., Effects of Al addition on the microstructure and mechanical property of Al<sub>x</sub>CoCrFeNi high-entropy alloys. *Intermetallics*. 26, 44–51 (2012).
- [2013, Wang] S. Wang, Atomic Structure Modeling of Multi-Principal-Element Alloys by the Principle of Maximum Entropy. *Entropy*. 15, 5536–5548 (2013).
- [2013, Otto] F. Otto, Y. Yang, H. Bei, E. P. George, Relative effects of enthalpy and entropy on the phase stability of equiatomic high-entropy alloys. *Acta Mater.* 61, 2628–2638 (2013).
- [2015, Senkov] O. N. Senkov, J. D. Miller, D. B. Miracle, C. Woodward, Accelerated exploration of multi-principal element alloys with solid solution phases. *Nat. Commun.* 6, 6529 (2015).

### Suggested references for citing this database

- [2017, Chen] H.-L. Chen, H. Mao, Q. Chen, Database development and Calphad calculations for high entropy alloys: Challenges, strategies, and tips. *Mater. Chem. Phys.* (2017). <https://doi.org/10.1016/j.matchemphys.2017.07.082>.
- [2017, Mao] H. Mao, H.-L. Chen, Q. Chen, TCHEA1: A Thermodynamic Database Not Limited for “High Entropy” Alloys. *J. Phase Equilibria Diffus.* 38, 353–368 (2017). <https://doi.org/10.1007/s11669-017-0570-7>.

### Published research examples applying TCHEA database

- [He, 2016] J.Y. He, H. Wang, Y. Wu, X.J. Liu, H.H. Mao, T.G. Nieh, Z.P. Lu, Precipitation behavior and its effects on tensile properties of FeCoNiCr high-entropy alloys, *Intermetallics*, 79 (2016) 41-52.
- [Bracq, 2017] G. Bracq, M. Laurent-Brocq, L. Perriere, R. Pires, J.-M. Joubert, I. Guillot, The fcc solid solution stability in the Co-Cr-Fe-Mn-Ni multi-component, *Acta Mater.*, 128 (2017) 327-336.

- 
- [Butler, 2017] T.M. Butler, M.L. Weaver, Investigation of the phase stabilities in AlNiCoCrFe high entropy alloys, *J. Alloys Compd.*, 691 (2017) 119-129.
- [Chen, 2017] Hai-Lin Chen, Huahai Mao, Qing Chen, "Database development and Calphad calculations for high entropy alloys: challenges, strategies, and tips", *Materials Chemistry and Physics*. (2017).
- [Choudhuri, 2017] D. Choudhuri, B. Gwalani, S. Gorsse, C.V. Mikler, R.V. Ramanujan, M.A. Gibson, R. Banerjee, Change in the primary solidification phase from Fcc to Bcc-based B2 in high entropy or complex concentrated alloys, *Scrip. Mater.*, 127 (2017) 186-190.
- [Guruvidyathri, 2017] K. Guruvidyathri, K.C. Hari Kumar, J.W. Yeh, and B.S. Murty, Topologically Close-packed Phase Formation in High Entropy Alloys: A Review of Calphad and Experimental Results, *JOM*, Vol. 69, No. 11 P. 2113-2124, ( 2017)
- [Mao, 2017] H. Mao, H.-L. Chen, Q. Chen, TCHEA1: a thermodynamic database not limited for "high entropy" alloys, *J Phase Equilib. Diff.* 38 (2017) 353-368.
- [Sathiyamoorthi, 2017] Sathiyamoorthi Praveen and Hyoung Seop Kim, High-Entropy Alloys: Potential Candidates for High-Temperature Applications – An Overview, *Adv. Eng. Mater.* (2017) 1700645.
- [Sheikh, 2017] S. Sheikh, H.H. Mao, S. Guo, Predicting solid solubility in CoCrFeNiMx (M = 4d transition metal) high-entropy alloys. *J. Appl. Phys.*, 121 (2017) 194903.
- [Takeuchi, 2017] A. Takeuchi, T. Wada, Y. Zhang, MnFeNiCuPt and MnFeNiCuCo high-entropy alloys designed based on L10 structure in Pettifor map for binary compounds, *Intermetallics*, 82 (2017) 107-115.
- [Tancret, 2017] F. Tancret, I. Toda-Caraballo, E. Menou, P.E.J.R. Diaz-Del-Castillo, Designing high entropy alloys employing thermodynamics and Gaussian process statistical analysis, *Mater. Design.*, 115 (2017) 486-497.
- [Abu-Odeh, 2018] A. Abu-Odeh, E. Galvan, T. Kirk, H. Mao, Q. Chen, P. Mason, R. Malak, R. Arroyave, "Efficient exploration of the High Entropy Alloy composition-phase space", *Acta Materialia*, 152 (2018) 41-57
- [Christofidou, 2018] K.A. Christofidou, E.J. Pickering, P. Orsatti, P.M. Mignanelli, T.J.A. Slater, H.J. Stone, N.G. Jones, On the influence of Mn on the phase stability of the CrMnxFeCoNi high entropy alloys, *Intermetallics*, 92 (2018) 84–92.
- [Holmström, 2018] Erik Holmström, Raquel Lizarraga, David Linder, Armin Salmasi, Wei Wang, Bartek Kaplan, Huahai Mao, Henrik Larsson, Levente Vitos, "High Entropy Alloys: Substituting for Cobalt in Cutting Edge Technology", *Applied Materials Today* 12 (2018) 322–329.

- 
- [Klimova, 2018] Margarita Klimova, Nikita Stepanov, Dmitry Shaysultanov, Ruslan Chernichenko, Nikita Yurchenko, Vladimir Sanin, and Sergey Zherebtsov. "Microstructure and Mechanical Properties Evolution of the Al, C-Containing CoCrFeNiMn-Type High-Entropy Alloy during Cold Rolling." *Materials* 11 (2018) (1): 53.
- [Li, 2018] Qin Li, Weimin Chen, Jing Zhong, Lijun Zhang, Qing Chen, and Zi-Kui Liu. 2018. "On Sluggish Diffusion in Fcc Al–Co–Cr–Fe–Ni High-Entropy Alloys: An Experimental and Numerical Study." *Metals* 8 (2018) (1): 16.
- [Lu, 2018] Yiping Lu, Xiaoxia Gao, Yong Dong, Tongmin Wang, Hai-Lin Chen, Huahai Mao, Yonghao Zhao, Hui Jiang, Zhiqiang Cao, Tingju Li and Sheng Guo, Preparing bulk ultrafine-microstructure high-entropy alloys via direct solidification. *Nanoscale*. 10 (2018) 1912-1919.
- [Liu, 2018] W.H. Liu, T. Yang, C.T. Liu, "Precipitation hardening in CoCrFeNi-based high entropy alloys", *Materials Chemistry and Physics*. 210 (2018) 2-12.
- [Saal, 2018] James E. Saal, Ida S. Berglund, Jason T. Sebastian, Peter K. Liaw, Greg B. Olson, Equilibrium high entropy alloy phase stability from experiments and thermodynamic modeling. *Scripta Materialia* 146 (2018) 5–8.
- [Schuh, 2018] B. Schuh, B. Völker, J. Todt, N. Schell, L. Perriere, J. Li, J.P. Couzinie, A. Hohenwarter, "Thermodynamic instability of a nanocrystalline, single-phase TiZrNbHfTa alloy and its impact on the mechanical properties", *Acta Materialia* 142 (2018) 201-212.
- [Taylor, 2018] Christopher D. Taylor, Pin Lu, James Saal, G. S. Frankel, and J. R. Scully. "Integrated Computational Materials Engineering of Corrosion Resistant Alloys." *Npj Materials Degradation* 2 (2018) (1): 6
- [Xia, 2018] Songqin Xia, Cláudio M. Lousada, Huahai Mao, Annika C. Maier, Pavel A. Korzhavyi, Rolf Sandström, Yugang Wang and Yong Zhang, "Nonlinear Oxidation Behavior in Pure Ni and Ni-Containing Entropic Alloys", *Frontiers in Materials*. (2018) Vol. 5 Article 53.
- [Cory, 2019] Francisco Gil Cory, Paul Wilson, Kester D. Clarke, Michael J. Kaufman, and Amy J. Clarke. 2019. "High-Throughput Solid Solution Strengthening Characterization in High Entropy Alloys." *Acta Materialia* 167 (April): 1–11.
- [Klimova, 2019] M.V. Klimova, D.G. Shaysultanov, R.S. Chernichenko, V.N. Sanin, N.D. Stepanov, S.V. Zherebtsov, and A.N. Belyakov. "Recrystallized Microstructures and Mechanical Properties of a C-Containing CoCrFeNiMn-Type High-Entropy Alloy." *Materials Science and Engineering: A* 740–741 (2019) 201–10.
- [Lee, 2019] Je In Lee, Koichi Tsuchiya, Wataru Tasaki, Hyun Seok Oh, Takahiro Sawaguchi, Hideyuki Murakami, Takanobu Hiroto, Yoshitaka Matsushita, and Eun Soo Park. 2019. "A Strategy of Designing High-Entropy Alloys with High-Temperature Shape Memory Effect." *Scientific Reports* 9 (1): 13140.

- [Raturi, 2019] Abheepsit Raturi, Jaya Aditya C, N. P. Gurao and Krishanu Biswas, "ICME Approach to Explore Equiatomic and Non-equiatomic Single Phase BCC Refractory High Entropy Alloys", J. Alloy. Comp. 806 (2019) 587-595.
- [Wang, 2019] Wei Wang, Ziyong Hou, Raquel Lizarraga, Ye Tian, R. Prasath Babu, Erik Holmström, Huahai Mao, Henrik Larsson, "An experimental and theoretical study of duplex fcc+hcp cobalt based entropic alloys", Acta Materialia 176 (2019) 11-18.
- [Rahul, 2020] M. R. Rahul and Gandham Phanikumar. "Solidification Behaviour of Undercooled Equiatomic FeCuNi Alloy." Journal of Alloys and Compounds, 815 (2020) 152334.

## Scientific Bibliography

See the Thermo-Calc Software scientific bibliography at: <https://www.thermocalc.com/support/resources/>.

---

## About the Included Phases

---

TCHEA4 contains nearly all stable phases in all assessed binary systems and most ternary systems. In total, 500 solution and intermetallic phases are modeled.

[Phases Included in TCHEA4](#) lists all phases and [Models for the Phases Included in TCHEA4](#) includes a detailed description of the models, e.g. number of sublattices and constituents on each sublattice.

The ordered B2 and L1<sub>2</sub> phases, together with bcc\_A2 and fcc\_A1, respectively, have been modeled with the so-called partitioning model, which describes an ordered phase and its disordered counterpart using a single Gibbs energy curve. This type of description is of particular importance to be able to predict second order transformations between a disordered phase and its ordered structures. Also note that there may be several possible composition sets for the phases named FCC\_L12 and BCC\_B2 designated by #1, #2, and so on (e.g. FCC\_L12#1 and FCC\_L12#2), due to the co-existence of disordered and ordered structures or the presence of miscibility gap. The #n suffix (where n is an integer) is generated dynamically by Thermo-Calc when using global minimization and therefore the identification of the phases should be determined from their site occupations. It can be found by LIST\_EQUILIBRIUM with the VXNS option in the Console Mode or showing the site fraction in moles of the constituent elements in the Graphical Mode. When the site occupancies of the first and second sublattices are equal the phase is disordered.



In Console Mode, you can list phases and constituents in the Database (TDB) module and the Gibbs (GES) module. For some phases, supplementary information is included in the definitions. In order to show the information, it is recommended to use the command LIST\_SYSTEM with the option of `Constituents` in the Database (TDB) module.

## Phases Included in TCHEA4

LIQUID	AL23V4	AL3ZR5	AL8CR5_L	ALMO_A2	CHI_A12
BCC_A2	AL24MN5ZN	AL45IR13	AL8MN5	ALN_B4	CO10CU57TI33
BCC_B2	AL28CU4MN7	AL4C3	AL8MO3	ALNI2ZN	CO11ZR2
FCC_A1	AL28IR9	AL4CR	AL8SIC7	ALPHA_B19	CO17Y2
FCC_L12	AL2FE	AL4MN_R	AL8V5	ALRE	CO2SI_C23
HCP_A3	AL2MN2SI3	AL4MN_U	AL9CO2	ALRE2	CO3SI
AL10CU10FE	AL2MNSI3	AL4RE	AL9CR4_H	ALRH2	CO3V
AL10FEMN2	AL2TI_LT	AL4SIC4	AL9CR4_L	ALSI3TI2	CO3Y
AL10V	AL2TI3N2	AL4W	AL9IR2	ALTI_L10	CO3Y2
AL11CR2	AL2W	AL4ZR5	AL9MN2ZN	ALTI2N	CO3Y4
AL11CU5MN3	AL2ZR3	AL5CO2_D811	ALB12_ALPHA	ALTI3_D019	CO5Y_D2D
AL11MN3ZN2	AL31MN6NI2	AL5CU4ZN	ALCR2C	ALTI3N	CO5Y8
AL11MN4_HT	AL3CO	AL5FE2	ALCU_DEL	ALZR	CO7HF
AL11MN4_LT	AL3IR	AL5FE4	ALCU_EPS	ALZR2_B82	CO7M2
AL11RE4	AL3MN4SI2	AL5MN6SI7	ALCU_ETA	B3SI	CO7Y6
AL12MN	AL3MNSI2	AL5TI2_HT	ALCU_PRIME	B4C	COSN_B35
AL12W	AL3NI_D011	AL5TI3	ALCU_ZETA	B6SI	COSN3_OS32
AL13CO4	AL3NI2_D513	AL5W	ALCU3MN2	B82_OMEGA	COY_BF
AL13CR2	AL3NI5	AL62CU25FE13	ALFESI_ALPHA	BCT_A5	COZN_DELTA
AL13FE2MN2	AL3RH_HT	AL63MO37	ALFESI_BETA	BETA_RHOMBO_B	COZN_GAMMA_D82
AL13FE4	AL3RH_LT	AL6MN	ALFESI_DELTA	BN_B4	COZN_GAMMA1
AL13IR4	AL3TI_D022	AL77W23	ALFESI_GAMMA	BNSI	COZN_GAMMA2
AL13NI38ZN49	AL3TI_LT	AL7CU2FE	ALFESI_TAU1	C14_LAVES	COZN_HT
AL15SI2M4	AL3Y_HT	AL7CU4NI	ALFESI_TAU3	C15_LAVES	COZN_LT
AL16FEMN3	AL3Y_LT	AL7RH3	ALM3C_E21	C16_THETA	CR2B_ORTH
AL1MN1SI1	AL3ZR_D023	AL7V	ALMNSI_T6	C36_LAVES	CR2NI2SI
AL2_7IR	AL3ZR2	AL7W3	ALMNSI_T8	CBCC_A12	
AL23CUFE4	AL3ZR4	AL8CR5_H	ALMNTI_L12	CEMENTITE	

CR3MN5	GAMMA	FEZN_ZETA	IR3SI2	M4SI3	MN6SI
CR3NI5SI2	CU5MN4SI	G_PHASE	IR3SI4	M5B6	MN9SI2
CR3SI_A15	CU6NISI3	GAMMA_D03	IR3SI5	M5C2	MNNI_L10
CR5B3_D81	CU6SN5_HT_NIAS	GAMMA_D83	IR3Y	M5SI3_D88	MNNI2
CRMN3_HT_SIGMA	CU6SN5_LT	GAMMA_H	IR3Y5	M6C	MNNISI_T5
CRNBSI	CU7Y1	GAS:G	IR3ZR5	M6SI5	MNNISI_T6
CRNI2_OP6	CU7Y2	GPHASE_ALCOTI	IR4B3	M7C3	MNTA
CRSI2_C40	CU8HF3	GRAPHITE	IR4B5	MB_B27	MNTI_HT
CRZN13	CU8ZR3	HEUSLER_L21	IR4SI5	MB_B33	MNTI_LT
CRZN17	CUB_A13	HF2M	IR5B4	MB2_C32	MNZN9
CU10HF7	CUMNZN	HF3N2	IRMN_L10	MB4	MO2B5_D8I
CU10SN3	CUSI_ETA	HF3NI7	IRNB_L10	MC_ETA	MO3IR
CU10ZR7	CUTI_B11	HF3RH4	IRSI	MC_SHP	MO7IR3
CU15SI4_EPSILON	CUTI3	HF3RH5	IRSI3_HT	MN11SI19	MOB_BG
CU2TI	DIAMOND_A4	HF4N3	IRSI3_LT	MN12Y	MOIR_HT
CU2TIZR	DIS_MU	HF5IR3	IRV_L10	MN15NI45SI40	MOIR_LT
CU2Y_H	DIS_SIG	HF5SN4	IRV_RT	MN15NI50SI35	MOIR3
CU2Y_L	FE2SI	HF8NI21	IRY3	MN1NI1SI1	MONI_DELTA
CU33SI7_DELTA	FE3SN2	HFIR	IRZR_ALPHA	MN2B_D1F	MONI4_BETA
CU3SN	FE4N_LP1	HFIR3	IRZR_B2	MN2NISI	MOSI2_C11B
CU3TI2	FE5SN3	HFMN	IRZR3	MN3NI2SI	MOZN22
CU41SN11	FE8SI2C	HFN	M11SI8	MN3SI	MOZN7
CU46NI25SI29	FECN_CHI	HFNI_ALPHA	M12C	MN3SN2	MSI_B27
CU4MNSN	FENBZR_CFC2	HFNI3_ALPHA	M23C6	MN3TI	MSI2_C1
CU4TI1	FESI_B20	HFNI3_BETA	M2B_TETR	MN4TI	MU_PHASE
CU4TI3	FESI2_H	HFRE	M3B2_D5A	MN52NI29SI19	MZR3_E1A
CU4Y	FESI2_L	HFRH	M3B2_T	MN66NI4SI30	NB13NI75TI12
CU51HF14	FEZN_DELTA	HFSN2_C40	M3B4_D7B	MN6N4	NB15NI56TI29
CU51ZR14	FEZN_GAMMA_D81	IR2Y3	M3C2	MN6N5	NB15NI80TI5
CU56SI11_	FEZN_GAMMA_D82	IR2Y5	M3SI	MN6NI16SI7	NB2ZN3
		IR3SI	M3SI2_D5A	MN6NISI3	NB3RU5

NB5NI75TI20	NIT12	RU2B3	TA41IR59	Y15C19_H
NB8NI9TI3	NIZN_TP2	RU2SI_C37	TA43IR57	Y15C19_R
NBSN2	NIZN8_DELTA	RU2SI3	TA5SI3_D8L	Y2C3_H
NBZN	NIZR	RU2SN3	TAAL	Y2C3_R
NBZN15	O_PHASE	RU2Y3	TAAL2	Y2ZN17
NBZN2	O1_DIS	RU2Y5	TAN_EPS	Y3SI5_HT
NBZN3	P_PHASE	RU3SN7	TASN2	Y3SI5_LT
NBZN7	PI_A13	RUB	TAU_COSNTI	YB6
NI10ZR7	R_PHASE	RUSI	TI2ALC	YB66
NI11ZR9	RE2SI	RU3Y3	TI2N_C4	YC_GAMMA
NI17Y2	RE3B_E1A	SI3N4	TI3ALC	YC2_C11A
NI2SI_TETA	RE7B3_D102	SIC	TI3ALC2	YSI2_HT
NI2TA_C11B	REB2	SIGMA	TI3N2	YZN2_A
NI2V	RESI2_C11B	SN10Y11	TI3SIC2	YZN2_B
NI2Y	REZR2	SN2Y_C49	TI4N3	YZN3
NI2Y3	RH2SN	SN3Y	TISI2_C54	YZN5
NI31SI12	RH3SN2	SN4Y5	TIZN10	Z_PHASE
NI3B_D011	RH3ZR	SN5TI6	TIZN15	ZN11Y3
NI3SI_MONOCL	RH4ZR3_HT	SN5Y2	TIZN5	ZN12Y
NI3SI_ORTHO	RH4ZR3_LT	SNTI2	V2B3	ZN22ZR
NI3SI2	RH5TI3	T_M23B6_D84	V3SN	ZN2ZR3
NI3SN4	RH5ZR3	T1_CU2TI	V4ZN5	ZN39ZR5
NI3TA_D0A	RH7B3	T1CUNITI	VSN2	ZN3ZR_HT
NI3TI_D024	RHB	T2_CU3TI2	VZN3_L12	ZNZR_B2
NI3Y	RHSN	T2CUNITI	W2B5_D8I	ZR5SI4_TP36
NI4B3	RHSN2_RT	T3_CU4TI3	W2B9	ZRB12_D2F
NI4Y	RHSN4	T4CUFETI	W3MC	ZRSI2_C49
NI5ZR_C15B	RHZR_HT	T4CUNITI	W5SI3_D8M	
NI7ZR2	RHZR_LT	T5CUFETI	WB_ALPHA	
NI8M	RHZR2	T6CUNITI	WB_BETA	
NISI_B31	RU25Y44	TA3SN	Y13ZN58	

## Models for the Phases Included in TCHEA4

LIQUID:L	
:AL AL1N1 B C CO CR CU FE HF HF1N1 IR MN MO N NB NI RH RE RU SI SN TA TI	
V W Y ZN ZR:	
> LIQUID mixture.	
FCC_A1	2 SUBL, SITES 1.00: 1.00:
:AL CO CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR:B C N	
VA:	
> If FCC_L12 is defined, this phase will be combined to it.	
FCC_L12	3 SUBL, SITES 0.75: 0.25: 1.00:
:AL CO CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR:AL CO	
CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR:B C N VA:	
> This phase has some contribution from FCC_A1.	
BCC_A2	2 SUBL, SITES 1.00: 3.00:
:AL CO CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR VA:B	
C N VA:	
> If BCC_B2 is defined, this phase will be combined to it.	
BCC_B2	3 SUBL, SITES 0.50: 0.50: 3.00:
:AL CO CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR VA:AL	
CO CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR VA:B C N VA:	
> This phase has some contribution from BCC_A2.	
HCP_A3	2 SUBL, SITES 1.00: 0.50:
:AL CO CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR:B C N	
VA:	
> Disordered HCP_A3 solution phase; Also for pure Co, Re, Ru, Ti	
CBCC_A12	2 SUBL, SITES 1.00: 1.00:
:AL CO CR CU FE IR MN MO NB NI RE RU SI SN TA TI V W Y ZN ZR:B C VA:	
CUB_A13	2 SUBL, SITES 1.00: 1.00:
:AL CO CR CU FE HF IR MN MO NB NI RE RU SI SN TA TI V W Y ZN ZR:B C VA:	
DIS_FCC_A1	2 SUBL, SITES 1.00: 1.00:

:AL CO CR CU FE HF IR MN MO NB NI RE RH RU SI SN TA TI V W Y ZN ZR: B C N VA:
DIAMOND_A4 :AL B C SI SN: > Pure C, Si or solution phases based on them
BETA_RHOMBO_B      2 SUBL, SITES 93.00: 12.00: :B: B C CU SI:
GRAPHITE :B C: > Strukturbericht A9
BCT_A5 :AL CU NI SN ZN: > Disordered BCT solution phase; Also for pure Sn
HCP_ZN              2 SUBL, SITES 1.00: 0.50: :AL CO CR CU FE IR MN NI RE RH RU SN TI ZN: VA:
T_M23B6_D84        4 SUBL, SITES 20.00: 6.00: 6.00: 3.00: :CO HF NI RE: B: B VA: AL CR HF MO RE TA TI V W ZR: > ternary borides, Prototype CR23C6, Strukturbericht D84
NI3B_D011          2 SUBL, SITES 3.00: 1.00: :CO CR FE MO NI: B: > also Co3B
RE3B_E1A          2 SUBL, SITES 3.00: 1.00: :CR MO RE TA W: B:
RE7B3_D102        3 SUBL, SITES 7.00: 3.00: 3.00: :CO CR MO NB RE RU TA W: B: B VA:
RH7B3              2 SUBL, SITES 0.70: 0.30: :RH: B:
M2B_TETR          2 SUBL, SITES 2.00: 1.00: :AL CO CR FE MN MO NB NI RE TA W: B: > also Co2B, Fe2B, Mn2B, Mo2B, Ni2B, Ta2B, W2B, Strukturbericht C16
CR2B_ORTH         2 SUBL, SITES 0.67: 0.33:

:CR FE MO RE:B:	
> Strukturbericht D1f	
MN2B_D1F	2 SUBL, SITES 0.67: 0.33:
:MN:B:	
CR5B3_D81	2 SUBL, SITES 0.62: 0.38:
:CR MO:B:	
> also Mo5B3	
M3B2_D5A	2 SUBL, SITES 3.00: 2.00:
:FE HF MO NB TA V:B:	
> also TA3B2, NB3B2, V3B2, prototype U3Si2, Strukturbericht D5a	
M3B2_T	3 SUBL, SITES 0.40: 0.20: 0.40:
:CR FE MO NI W:CR FE NI:B:	
> ternary borides, Mo2FeB2, Mo2CrB2	
IR4B3	2 SUBL, SITES 4.00: 3.00:
:IR:B:	
NI4B3	2 SUBL, SITES 0.57: 0.43:
:NI:B:	
IR5B4	2 SUBL, SITES 5.00: 4.00:
:IR:B:	
MB_B27	2 SUBL, SITES 1.00: 1.00:
:B:CO CR FE HF MN MO RE TI Y:	
> prototype FeB(beta), Strukturbericht B27, also CoB, HfB, MnB, TiB	
MB_B33	2 SUBL, SITES 1.00: 1.00:
:CR FE HF MO NB NI TA TI V:B:	
> CrB, NbB, NiB, TaB, VB	
MOB_BG	2 SUBL, SITES 1.00: 1.00:
:CR FE MO:B:	
> MoB, Strukturbericht Bg	
WB_ALPHA	2 SUBL, SITES 1.00: 1.00:
:B C VA:W:	
> Strukturbericht Bg	

RUB	2 SUBL, SITES 1.00: 1.00:
:RU:B:	
WB_BETA	2 SUBL, SITES 1.00: 1.00:
:B C VA:W:	
> Strukturbericht B33, prototype CrB	
RHB	2 SUBL, SITES 1.00: 1.10:
:RH:B:	
> Strukturbericht B81	
M5B6	2 SUBL, SITES 5.00: 6.00:
:NB V:B:	
> proptotype V5B6, also Nb5B6	
IR4B5	2 SUBL, SITES 4.00: 5.00:
:IR:B:	
M3B4_D7B	2 SUBL, SITES 4.00: 3.00:
:B:AL CR HF MN NB TA TI V:	
> Prototype: Ta3B4, also Cr3B4, Mn3B4, Nb3B4, Ti3B4, V3B4 and ternary	
RU2B3	2 SUBL, SITES 2.00: 3.00:
:RU:B:	
V2B3	2 SUBL, SITES 0.40: 0.60:
:V:B:	
MB2_C32	2 SUBL, SITES 2.00: 1.00:
:B:AL CR HF MN MO NB RU TA TI V Y ZR:	
> AlB2,CrB2,HfB2,MnB2,MoB2,NbB2,RuB2,TaB2,TiB2,VB2,YB2,ZrB2	
REB2	3 SUBL, SITES 1.00: 2.00: 2.00:
:RE:B:B VA:	
MO2B5_D8I	2 SUBL, SITES 0.32: 0.68:
:MO:B:	
> Strukturbericht D8i	
W2B5_D8I	2 SUBL, SITES 5.00: 2.00:
:B C VA:W:	
MB4	2 SUBL, SITES 0.20: 0.80:

:CR MN MO Y:B:
W2B9            2 SUBL, SITES 9.00: 2.00: :B:W:
YB6            2 SUBL, SITES 1.00: 6.00: :Y:B:
ZRB12_D2F       2 SUBL, SITES 12.00: 1.00: :B:Y ZR: > also YB12, Strukturbericht D2f
ALB12_ALPHA    2 SUBL, SITES 1.00: 12.00: :AL TI:B:
YB66           2 SUBL, SITES 1.00: 66.00: :Y:B:
M12C           3 SUBL, SITES 6.00: 6.00: 1.00: :CO NI:MO W:C: > Prototype Mo6Ni6C
M6C            4 SUBL, SITES 2.00: 2.00: 2.00: 1.00: :CO FE NI:MO NB TA W:CO CR FE MO NB NI TA V W:C: > Prototype W3Fe3C, Strukturbericht E93
ALM3C_E21       3 SUBL, SITES 1.00: 3.00: 1.00: :AL:CO FE:C: > AlCo3C, AlFe3C, Perovskite
Ti3AlC          3 SUBL, SITES 3.00: 1.00: 1.00: :Ti:Al:C VA:
W3MC           3 SUBL, SITES 3.00: 1.00: 1.00: :W:CO NI:C: > W3CoC, W3NiC
M23C6           3 SUBL, SITES 20.00: 3.00: 6.00: :CO CR FE MN NI RE V:CO CR FE MN MO NI RE V W:C: > prototype Cr23C6, Strukturbericht D84
ALCR2C          3 SUBL, SITES 1.00: 1.00: 2.00: :AL:C:CR:

CEMENTITE	2	SUBL, SITES	3.00:	1.00:
:CO CR FE MN MO NI V W:C N:				
> Strukturbericht D011				
Ti2AlC	3	SUBL, SITES	2.00:	1.00: 1.00:
:Ti:Al:C VA:				
M5C2	2	SUBL, SITES	5.00:	2.00:
:FE MN:C:				
> Mn5C2				
M7C3	2	SUBL, SITES	7.00:	3.00:
:CO CR FE MN MO NI RE V W:C:				
> proptotype Cr7C3, Strukturbericht D101				
Ti3AlC2	3	SUBL, SITES	3.00:	1.00: 2.00:
:Ti:Al Si:C VA:				
M3C2	2	SUBL, SITES	3.00:	2.00:
:CO CR MO V W:C:				
> prototype Cr3C2(beta), Strukturbericht D510				
Al4C3	2	SUBL, SITES	4.00:	3.00:
:AL Si:C:				
> Strukturbericht D71				
MC_ETA	2	SUBL, SITES	1.00:	1.00:
:MO V W:C VA:				
> MoC, prototype TiAs, Strukturbericht Bi				
MC_SHP	2	SUBL, SITES	1.00:	1.00:
:MO W:C N:				
> Prototype WC, also MoC low temperature,				
YC_GAMMA	2	SUBL, SITES	1.00:	1.00:
:Y:C C2 VA:				
Y15C19_H	2	SUBL, SITES	19.00:	15.00:
:C:Y:				
Y15C19_R	2	SUBL, SITES	19.00:	15.00:
:C:Y:				

Y2C3_H	3 SUBL, SITES 2.00: 2.00: 1.00:
:Y:C:C VA:	
Y2C3_R	3 SUBL, SITES 2.00: 2.00: 1.00:
:Y:C:C VA:	
YC2_C11A	
:C2Y1:	
PI_A13	3 SUBL, SITES 12.80: 7.20: 4.00:
:CR:FE NI:N:	
> Cr-(Fe,Ni)-nitrides	
FE4N_LP1	2 SUBL, SITES 4.00: 1.00:
:CO CR FE MN NI:C N:	
> Only stable in Co-N, Fe-N and Cr-Fe-N when gas suspended	
ALTI3N	3 SUBL, SITES 1.00: 3.00: 1.00:
:AL:TI:N:	
> Strukturbericht E21	
ALTI2N	3 SUBL, SITES 1.00: 2.00: 1.00:
:AL:TI:N:	
AL2TI3N2	3 SUBL, SITES 2.00: 3.00: 2.00:
:AL:TI:N:	
FECN_CHI	2 SUBL, SITES 2.20: 1.00:
:FE:C N:	
> Only stable in C-Fe-N when gas suspended	
TI4N3	2 SUBL, SITES 0.69: 0.32:
:TI:N:	
TI2N_C4	2 SUBL, SITES 2.00: 1.00:
:TI:N:	
Z_PHASE	3 SUBL, SITES 1.00: 1.00: 1.00:
:CR FE:MO NB V:N VA:	
HF3N2	2 SUBL, SITES 3.00: 2.00:
:HF:N:	
MN6N4	2 SUBL, SITES 6.00: 4.00:

:MN:N:	
Tl3N2	2 SUBL, SITES 0.71: 0.29:
:TI:N:	
Hf4N3	2 SUBL, SITES 4.00: 3.00:
:HF:N:	
MN6N5	2 SUBL, SITES 6.00: 5.00:
:MN:N:	
ALN_B4	2 SUBL, SITES 1.00: 1.00:
:AL:N:	
TAN_EPS	2 SUBL, SITES 1.00: 1.00:
:TA:N:	
HFN	2 SUBL, SITES 1.00: 1.00:
:HF:N VA:	
MN6SI	2 SUBL, SITES 17.00: 3.00:
:AL MN:SI ZN:	
MN9SI2	2 SUBL, SITES 33.00: 7.00:
:MN:SI:	
CR3NI5SI2	4 SUBL, SITES 3.00: 5.00: 2.00: 1.00:
:CR:NI:SI:C VA:	
CR2NI2SI	3 SUBL, SITES 5.00: 5.00: 3.00:
:CR:NI:SI:	
CR3SI_A15	3 SUBL, SITES 3.00: 1.00: 3.00:
:CR FE IR MO NB NI RE SI SN TA TI V ZR:AL CO CR IR MO NB NI RH RU SI SN	
TA TI V ZR:C VA:	
> also Cr3Ir, Cr3Rh, Cr3Ru, Mo3Al, Mo3Si, Nb3Al, Nb3Sn, Ti3Ir, V3Co, V3Ir, V3Ni, V3Si.	
Prototype Cr3Si.	
CO3SI	2 SUBL, SITES 3.00: 1.00:
:CO:SI:	
IR3SI	2 SUBL, SITES 3.00: 1.00:
:IR:SI:	
M3SI	2 SUBL, SITES 3.00: 1.00:

:HF NB TA TI ZR:SI: > also Nb3Si, Ta3Si, Ti3Si, Zr3Si
MN3SI            2 SUBL, SITES 3.00: 1.00: :MN FE:AL SI:
NI3SI_ORTHO      2 SUBL, SITES 3.00: 1.00: :NI:SI: > Strukturbericht D011
NI31SI12        2 SUBL, SITES 5.00: 2.00: :CO CR CU FE NI:SI: > Ni5Si2 and its solution
CO2SI_C23        2 SUBL, SITES 2.00: 1.00: :CO CR CU FE IR NI TI:SI: > Prototype Co2Si (alpha), also Ni2Si (delta)
CRNBSI          3 SUBL, SITES 1.00: 1.00: 1.00: :CR:NB:SI: > hP9 Fe2P
FE2SI            2 SUBL, SITES 0.67: 0.33: :FE:SI:
NI2SI_TETA       3 SUBL, SITES 1.00: 1.00: 1.00: :CU NI:NI VA:AL SI:
RE2SI            2 SUBL, SITES 2.00: 1.00: :RE:SI:
RU2SI_C37        2 SUBL, SITES 2.00: 1.00: :RU:SI:
M5SI3_D88        4 SUBL, SITES 2.00: 3.00: 3.00: 1.00: :CR CU FE HF MN MO NI NB SI TI W Y ZR:AL CR SI SN TI:CR CU FE HF MN MO NI NB TI Y ZR:C SN VA: > Fe5Si3, Hf5Si3, Mn5Si3, Ti5Cr3, Ti5Si3, Ti5Sn3, Y5Si3, Zr5Si3, Hf5Sn3, Y5Sn3, Zr5Sn3, Zr5Sn4, Mo5Si3C, Cr5Si3C, prototype Mn5Si3
TA5SI3_D8L       2 SUBL, SITES 5.00: 3.00: :HF NB TA:AL SI:

> also Nb5Si3, prototype Cr5B3, Strukturbericht D8I	
W5Si3_D8M	3 SUBL, SITES 4.00: 1.00: 3.00: :CR FE MO NB TI V W:CR FE MO NB TI V W Si:AL SI SN: > also Cr5Si3, Mo5Si3, Nb5Si3, V5Si3
IR3Si2	2 SUBL, SITES 3.00: 2.00: :IR:SI:
M3Si2_D5A	2 SUBL, SITES 3.00: 2.00: :HF NB ZR:SI: > Hf3Si2, Zr3Si2
NI3Si2	2 SUBL, SITES 3.00: 2.00: :NI:SI:
M11Si8	2 SUBL, SITES 11.00: 8.00: :CR NB:SI: > Cr11Si8, Nb11Si8
M4Si3	2 SUBL, SITES 4.00: 3.00: :CR NI RU:SI: > Cr4Si3, Nb4Si3, Ru4Si3
ZR5Si4_TP36	2 SUBL, SITES 5.00: 4.00: :HF NB TI Y ZR:AL SI: > Hf5Si4, Ti5Si4, Y5Si4, Zr5Si4(alpha), prototype Zr5Si4
M6Si5	2 SUBL, SITES 6.00: 5.00: :CR NB TI V:SI: > Cr6Si5, Nb6Si5, V6Si5
FESI_B20	2 SUBL, SITES 1.00: 1.00: :CO CR FE MN NI RE:AL SI: > also CoSi, CrSi, Mn Si, ReSi, Prototype FeSi
IRSI	2 SUBL, SITES 1.00: 1.00: :IR:SI:
MSI_B27	2 SUBL, SITES 1.00: 1.00: :HF NB TI Y ZR:AL SI: > TiSi, HfSi, YSi, ZrSi(alpha), Prototype FeB

NISI_B31	2 SUBL, SITES 1.00: 1.00:
:NI:SI:	
RUSI	2 SUBL, SITES 1.00: 1.00:
:RU:SI:	
IR4SI5	2 SUBL, SITES 4.00: 5.00:
:IR:SI:	
IR3SI4	2 SUBL, SITES 3.00: 4.00:
:IR:SI:	
RU2SI3	2 SUBL, SITES 2.00: 3.00:
:RU:SI:	
IR3SI5	2 SUBL, SITES 3.00: 5.00:
:IR:SI:	
Y3SI5_HT	2 SUBL, SITES 3.00: 5.00:
:Y:SI:	
Y3SI5_LT	2 SUBL, SITES 3.00: 5.00:
:Y:SI:	
MN11SI19	2 SUBL, SITES 11.00: 19.00:
:MN:AL SI:	
CRSI2_C40	2 SUBL, SITES 1.00: 2.00:
:CR CU HF MO NB SI TA TI V W:AL CR CU SI:	
> CrSi <sub>2</sub> , NbSi <sub>2</sub> , TaSi <sub>2</sub> , VSi <sub>2</sub> , Prototype CrSi <sub>2</sub>	
FESI2_H	2 SUBL, SITES 0.30: 0.70:
:FE:SI:	
FESI2_L	2 SUBL, SITES 0.33: 0.67:
:FE:SI:	
MSI2_C1	2 SUBL, SITES 1.00: 2.00:
:CO CU MN NI:AL CU SI:	
> CoSi <sub>2</sub> , Prototype CaF <sub>2</sub>	
MOSI2_C11B	2 SUBL, SITES 1.00: 2.00:
:AL CO CU FE MO NI RH W ZN:AL CR HF SI TI ZR:	
> also AlCr <sub>2</sub> ,CuHf <sub>2</sub> ,CuTi <sub>2</sub> ,CuZr <sub>2</sub> ,PdTi <sub>2</sub> ,RhTi <sub>2</sub> ,Ti <sub>2</sub> Zn,WSi <sub>2</sub> ,ZnZr <sub>2</sub> , Prototype	

RESI2_C11B	2	SUBL, SITES 0.36: 0.64:	:RE:SI:
TISI2_C54	2	SUBL, SITES 1.00: 2.00:	:MO NB RU TI ZR:AL SI SN: > also ZrSn2
YSI2_HT	2	SUBL, SITES 1.00: 2.00:	:Y:SI:
ZRSI2_C49	2	SUBL, SITES 1.00: 2.00:	:ZR Y HF NB:SI: > ZrSi2, YSi2, HfSi2, Prototype ZrSi2
IRSI3_LT	2	SUBL, SITES 1.00: 3.00:	:IR:SI:
IRSI3_HT	2	SUBL, SITES 1.00: 3.00:	:IR:SI:
ALFESI_ALPHA	4	SUBL, SITES 0.66: 0.19: 0.05: 0.10:	:AL:FE:SI:AL SI:
ALFESI_BETA	3	SUBL, SITES 14.00: 3.00: 3.00:	:AL:FE:SI:
ALFESI_GAMMA	3	SUBL, SITES 3.00: 1.00: 1.00:	:AL:FE:SI:
ALFESI_DELTA	3	SUBL, SITES 0.55: 0.15: 0.30:	:AL:FE:SI:
ALFESI_TAU1	3	SUBL, SITES 2.00: 2.00: 1.00:	:AL:FE:SI:
ALFESI_TAU3	3	SUBL, SITES 2.00: 1.00: 1.00:	:AL:FE:SI:
AL2MN2SI3	3	SUBL, SITES 2.00: 2.00: 3.00:	:AL:MN:SI: > the Al-Mn-Si ternary phase, tao1
AL5MN6SI7	3	SUBL, SITES 5.00: 6.00: 7.00:	:AL:MN:SI:

> the Al-Mn-Si ternary phase, tao2	
AL1MN1SI1	3 SUBL, SITES 1.00: 1.00: 1.00: :AL:MN:SI: > the Al-Mn-Si ternary phase, tao3
AL3MNSI2	3 SUBL, SITES 3.00: 1.00: 2.00: :AL:MN:SI: > the Al-Mn-Si ternary phase, tao4
AL3MN4SI2	3 SUBL, SITES 3.00: 4.00: 2.00: :AL:MN:SI: > the Al-Mn-Si ternary phase, tao5
ALMNSI_T6	2 SUBL, SITES 4.00: 1.00: :AL MN:SI: > the Al-Mn-Si ternary phase, tao6
ALMNSI_T8	5 SUBL, SITES 6.00: 2.00: 12.00: 6.00: 2.00: :MN VA:MN VA:AL:AL SI:AL SI: > the Al-Mn-Si ternary phase, tao8
AL2MNSI3	3 SUBL, SITES 2.00: 1.00: 3.00: :AL:MN:SI: > the Al-Mn-Si ternary phase, tao10
MN15NI45SI40	3 SUBL, SITES 0.15: 0.45: 0.40: :MN:NI:SI: > Mn-Ni-Si ternary phase, T1 or N
MN15NI50SI35	3 SUBL, SITES 0.15: 0.50: 0.35: :MN:NI:SI: > Mn-Ni-Si ternary phase, T2 or PHI
MN6NI16SI7	3 SUBL, SITES 0.21: 0.55: 0.24: :MN:NI:SI: > Mn-Ni-Si ternary phase, T3 or G
MN1NI1SI1	3 SUBL, SITES 1.00: 1.00: 1.00: :MN:NI:SI: > Mn-Ni-Si ternary phase, T4 or E

MNNISI_T5	2	SUBL, SITES	1.00: 2.00:
:MN:NI Si:			
> Mn-Ni-Si ternary phase, T5 or "tao 1"			
MNNISI_T6	2	SUBL, SITES	1.00: 2.00:
:MN:NI Si:			
> Mn-Ni-Si ternary phase, T6 or "tao 2"			
MN3NI2SI	3	SUBL, SITES	3.00: 2.00: 1.00:
:MN:NI:Si:			
> Mn-Ni-Si ternary phase, T7 or Omega			
MN2NISI	2	SUBL, SITES	3.00: 1.00:
:MN NI:Si:			
> Mn-Ni-Si ternary phase, T8 or S			
MN6NISI3	3	SUBL, SITES	0.61: 0.12: 0.27:
:MN:NI:Si:			
> Mn-Ni-Si ternary phase, T9 or R			
MN66NI4SI30	3	SUBL, SITES	0.66: 0.04: 0.30:
:MN:NI:Si:			
> Mn-Ni-Si ternary phase, T10 or U			
MN52NI29SI19	3	SUBL, SITES	0.52: 0.29: 0.19:
:MN:NI:Si:			
> Mn-Ni-Si ternary phase, T11 or W			
B4C	2	SUBL, SITES	1.00: 1.00:
:B11C B12:B2 C2B CB2:			
> Strukturbericht D1g			
BN_B4	2	SUBL, SITES	1.00: 1.00:
:B:N:			
B3SI	3	SUBL, SITES	6.00: 2.00: 6.00:
:B:Si:B Si:			
> Strukturbericht D1g			
B6SI	3	SUBL, SITES	210.00: 23.00: 48.00:
:B:Si:B Si:			

BNSI	3 SUBL, SITES 61.00: 1.00: 8.00:
	:B:Si:B Si:
SIC	2 SUBL, SITES 1.00: 1.00:
	:Si:C:
	> Strukturbericht B3
Si3N4	2 SUBL, SITES 3.00: 4.00:
	:Si:N:
AL4SiC4	3 SUBL, SITES 4.00: 1.00: 4.00:
	:AL:Si:C:
AL8SiC7	3 SUBL, SITES 8.00: 1.00: 7.00:
	:AL:Si:C:
FE8Si2C	3 SUBL, SITES 8.00: 2.00: 1.00:
	:FE:Si:C:
	> Prototype Mn8Si2C
Ti3SiC2	3 SUBL, SITES 3.00: 1.00: 2.00:
	:Ti:Si:C:
DIS_MU	
	:AL CO CR CU FE MN MO NB NI RE TA TI W:
	> Part of the description of MU_PHASE.
MU_PHASE	4 SUBL, SITES 1.00: 2.00: 6.00: 4.00:
	:AL CO CR CU FE MN MO NB NI RE TA TI W:AL CO CR CU FE MN MO NB NI RE TA
	TI W:AL CO CR CU FE MN MO NB NI RE TI TA W:AL CO CR CU FE MN MO NB NI RE
	TA TI W:
DIS_SIG	
	:AL CO CR FE IR MN MO NB NI RE RU SI TA TI V W:
	> Part of the description of SIGMA phase.
SIGMA	3 SUBL, SITES 10.00: 4.00: 16.00:
	:AL CO CR FE IR MN MO NB NI RE RU SI TA TI V W:AL CO CR FE IR MN MO NB
	NI RE RU SI TA TI V W:AL CO CR FE IR MN MO NB NI RE RU SI TA TI V W:
	> DIS_SIG contribution is introduced in the description of this phase
CHI_A12	3 SUBL, SITES 24.00: 10.00: 24.00:

		:CR FE NI RE:AL CR HF MO NB TA TI W ZR:CR FE MO NB NI RE TA W: > also Hf5Re24, Mo2Re8, Ta3Re7, Ti5Re24, WRe3, Zr5Re24
G_PHASE	3	SUBL, SITES 16.00: 6.00: 7.00: :AL CO FE MN NI TI:HF NB TI Y ZR:CO FE MN NI SI: > prototype Th6Mn23, Strukturbericht D8a, also Al16Ni7Ti6(Mu)
GAMMA_D83	3	SUBL, SITES 4.00: 1.00: 8.00: :AL NI SI ZN:AL CU NI SI ZN:CU FE MN NI ZN: > Cu5Zn8, Ni5Zn8, Al5Cu8 (rt), Prototype Cu9Al4 (cP52, P-43m)
GAMMA_H	3	SUBL, SITES 4.00: 1.00: 8.00: :AL ZN:AL CU ZN:CU FE MN NI: > Cu5Zn8-type Al8Cu5 (ht) phase
HEUSLER_L21	3	SUBL, SITES 0.50: 0.50: 1.00: :AL CR NI SN TI:AL HF NB NI TA TI ZR:CO FE NI RU VA: > Heusler phase, Strukturbericht L21
P_PHASE	3	SUBL, SITES 24.00: 20.00: 12.00: :CR FE NI RE:CR FE MO NI RE:MO: > Prototype : Cr9Mo21Ni20
R_PHASE	3	SUBL, SITES 27.00: 14.00: 12.00: :CO CR FE NI RE:MO W:CO CR FE MO NI RE W: > Prototype : Co5Cr2Mo3
MNNI_L10	2	SUBL, SITES 0.50: 0.50: :CU MN NI:CU MN NI:
IRMN_L10	2	SUBL, SITES 0.50: 0.50: :IR MN TI:IR MN TI: > also IrTi
ALTI_L10	2	SUBL, SITES 1.00: 1.00: :AL CO CR MN MO NB SN TA TI V W ZR:AL CO CR MN MO NB SN TA TI V W ZR: > Solution of ordered L10.
ALCU_EPS	2	SUBL, SITES 1.00: 1.00: :AL CU NI:CU FE:
ALCU_ETA	2	SUBL, SITES 1.00: 1.00:

:AL CU:CU FE NI ZN:	
ALPHA_B19	2 SUBL, SITES 1.00: 1.00: :MO NB TI V ZR:MO NB TI V ZR: > ordered HCP, prototype AuCd
ALRE	2 SUBL, SITES 1.00: 1.00: :AL:RE:
ALZR	2 SUBL, SITES 1.00: 1.00: :AL:HF Y ZR: > also AlHf, AlY
COSN_B35	2 SUBL, SITES 0.50: 0.50: :CO FE NI:SN: > This is CoSn, FeSn
COY_BF	2 SUBL, SITES 1.00: 1.00: :CO:Y:
CUTI_B11	2 SUBL, SITES 1.00: 1.00: :CO CU NI TI:CU NI TA TI:
HFIR	2 SUBL, SITES 1.00: 1.00: :HF:IR:
HFMN	2 SUBL, SITES 0.50: 0.50: :HF:MN:
HFNI_ALPHA	2 SUBL, SITES 0.50: 0.50: :HF:NI:
HFRE	2 SUBL, SITES 1.00: 1.00: :HF:RE:
HFRH	2 SUBL, SITES 1.00: 1.00: :HF RH:RH:
IRV_L10	2 SUBL, SITES 0.50: 0.50: :IR:IR V:
IRV_RT	2 SUBL, SITES 1.00: 1.00: :IR:V:
IRZR_B2	2 SUBL, SITES 1.00: 1.00:

:IR ZR:IR ZR:	
IRZR_ALPHA	2 SUBL, SITES 1.00: 1.00:
:IR ZR:ZR:	
MNTA	2 SUBL, SITES 1.00: 1.00:
:MN:TA:	
MNTI_LT	2 SUBL, SITES 1.00: 1.00:
:MN:TI:	
MOIR_HT	2 SUBL, SITES 1.00: 1.00:
:IR MO:IR MO:	
MOIR_LT	2 SUBL, SITES 1.00: 1.00:
:MO:IR:	
NBZN	2 SUBL, SITES 0.50: 0.50:
:NB:ZN:	
NIZN_TP2	2 SUBL, SITES 0.50: 0.50:
:AL CU NI ZN:AL NI ZN:	
NIZR	2 SUBL, SITES 1.00: 1.00:
:NI:TI Y ZR:	
RHSN	2 SUBL, SITES 1.00: 1.00:
:RH:SN:	
RHZR_LT	2 SUBL, SITES 1.00: 1.00:
:RH:RH ZR:	
> i.e. alpha_RhZr	
RHZR_HT	2 SUBL, SITES 1.00: 1.00:
:RH:RH ZR:	
> i.e. beta_RhZr	
ZNZR_B2	2 SUBL, SITES 1.00: 1.00:
:ZN:ZR:	
ALCU_PRIME	2 SUBL, SITES 2.00: 1.00:
:AL:CU:	
AL2FE	2 SUBL, SITES 2.00: 1.00:
:AL CU:FE MN:	

ALRE2	2 SUBL, SITES 1.00: 2.00:
:AL:RE:	
ALRH2	2 SUBL, SITES 1.00: 2.00:
:AL:RH:	
AL2TI_LT	2 SUBL, SITES 2.00: 1.00:
:AL NB TI:AL CO NB TA TI V ZR:	
AL2W	2 SUBL, SITES 2.00: 1.00:
:AL:W:	
ALZR2_B82	2 SUBL, SITES 1.00: 2.00:
:AL:TI Y ZR:	
> also AIY2	
C14_LAVES	2 SUBL, SITES 2.00: 1.00:
:AL CO CR CU FE HF MN MO NB NI SI RE RU TA TI V W Y ZN ZR:AL CO CR CU FE	
HF MN MO NB NI SI RE RU TA TI V W Y ZN ZR:	
> prototype CuZn2	
C15_LAVES	2 SUBL, SITES 2.00: 1.00:
:AL CO CR CU FE HF IR MN MO NB NI SI RE RU TA TI V W Y ZN ZR:AL CO CR CU	
FE HF MO NB NI SI RE RU TA TI V W Y ZR:	
C16_THETA	2 SUBL, SITES 2.00: 1.00:
:AL HF MN MO NB SN TA TI W ZR:AL CO CR CU FE IR MN NI RH SI:	
> Al2Cu,Hf2Al,Hf2Ni,Sn2Co,Sn2Fe,Sn2Rh,Sn2Mn,Ta2Co,Ta2Ni,Zr2Co,Zr2Ni,Zr2Ir	
C36_LAVES	2 SUBL, SITES 2.00: 1.00:
:AL CO CR CU FE HF MO NB NI TA TI W ZR:AL CO CR CU FE HF MO NB NI TA TI	
W ZR:	
CRNI2_OP6	2 SUBL, SITES 1.00: 2.00:
:CR MO W:MO NI W:	
CU2TI	2 SUBL, SITES 2.00: 1.00:
:CO CU NI:TI:	
CU2Y_H	2 SUBL, SITES 2.00: 1.00:
:CU:Y:	
CU2Y_L	2 SUBL, SITES 2.00: 1.00:

:CU:Y:	
HF2M	2 SUBL, SITES 2.00: 1.00:
:HF:IR RH:	
HFSN2_C40	2 SUBL, SITES 1.00: 2.00:
:HF:SN:	
MNNI2	2 SUBL, SITES 1.00: 2.00:
:MN NI:NI:	
NBZN2	2 SUBL, SITES 0.33: 0.67:
:NB:ZN:	
NI2V	2 SUBL, SITES 2.00: 1.00:
:MO NI:MO NB TA V:	
> Ni2V and its solutions	
NI2TA_C11B	2 SUBL, SITES 2.00: 1.00:
:CO NI:TA TI:	
NITI2	2 SUBL, SITES 1.00: 2.00:
:CO CR CU FE NI RE TI:AL CR CU HF NI TA TI ZR:	
NI2Y	2 SUBL, SITES 2.00: 1.00:
:NI:Y:	
REZR2	2 SUBL, SITES 1.00: 2.00:
:NI RE:ZR:	
RHSN2_RT	2 SUBL, SITES 0.33: 0.67:
:RH:SN:	
RH2SN	2 SUBL, SITES 2.00: 1.00:
:RH:SN:	
SNTI2	2 SUBL, SITES 1.00: 2.00:
:SN:TI:	
SN2Y_C49	2 SUBL, SITES 2.00: 1.00:
:SN:Y:	
RHZR2	2 SUBL, SITES 1.00: 2.00:
:RH:ZR:	
TASN2	2 SUBL, SITES 1.00: 2.00:

:TA:SN:	
VSN2	2 SUBL, SITES 0.60: 0.40:
:SN:V:	
YZN2_A	2 SUBL, SITES 1.00: 2.00:
:Y:ZN:	
YZN2_B	2 SUBL, SITES 1.00: 2.00:
:Y:ZN:	
AL3CO	2 SUBL, SITES 3.00: 1.00:
:AL:CO:	
ALCU_DEL	2 SUBL, SITES 2.00: 3.00:
:AL:CU FE:	
AL3IR	2 SUBL, SITES 0.75: 0.25:
:AL:IR:	
AL3NI_D011	2 SUBL, SITES 0.75: 0.25:
:AL:NI:	
AL3RH_LT	2 SUBL, SITES 3.00: 1.00:
:AL:RH:	
ALTI3_D019	2 SUBL, SITES 3.00: 1.00:
:AL CO CR CU MN MO NB NI SN TA TI V W ZR:AL C CR MO NB NI SI SN TA TI V W:	
> Ni3Sn_LT, Mn3Sn, Ti3Sn.	
AL3TI_D022	2 SUBL, SITES 3.00: 1.00:
:AL CO CR FE MO NB NI SI TI V:AL CO CR MO NB NI SI TA TI V ZR:	
> gamma double prime, prototype Al3Ti, also Al3Nb, Al3Ta, Al3V, Ni3V	
AL3Y_HT	2 SUBL, SITES 0.75: 0.25:
:AL:Y:	
AL3Y_LT	2 SUBL, SITES 0.75: 0.25:
:AL:Y:	
AL3ZR_D023	2 SUBL, SITES 3.00: 1.00:
:AL ZN:HF TI ZR:	
> also Al3Hf, Zn3Zr	
COSN3_OS32	2 SUBL, SITES 0.25: 0.75:

:CO:SN:	
CO3V	2 SUBL, SITES 3.00: 1.00:
:CO NI V:CO V:	
CO3Y	2 SUBL, SITES 3.00: 1.00:
:CO:Y:	
CU3SN	2 SUBL, SITES 3.00: 1.00:
:CU SN:CU SN:	
CUTI3	2 SUBL, SITES 1.00: 3.00:
:CU TI:TI:	
HFIR3	2 SUBL, SITES 1.00: 3.00:
:HF:IR:	
HFNI3_ALPHA	2 SUBL, SITES 0.25: 0.75:
:HF:NI:	
HFNI3_BETA	2 SUBL, SITES 0.25: 0.75:
:HF:NI:	
IRY3	2 SUBL, SITES 1.00: 3.00:
:IR:Y:	
IR3Y	2 SUBL, SITES 3.00: 1.00:
:IR:Y:	
IRZR3	2 SUBL, SITES 1.00: 3.00:
:IR:ZR:	
MZR3_E1A	2 SUBL, SITES 1.00: 3.00:
:CO FE NI:Y ZR:	
> COZR3, FEZR3, NiY3, COY3	
MN3TI	2 SUBL, SITES 3.00: 1.00:
:MN:TI:	
MOIR3	2 SUBL, SITES 1.00: 3.00:
:MO:IR MO:	
MO3IR	2 SUBL, SITES 3.06: 0.94:
:MO:IR:	
NBZN3	2 SUBL, SITES 0.25: 0.75:

:NB:ZN:	
NI3TA_D0A	2 SUBL, SITES 3.00: 1.00: :AL CO CR FE NI NB:AL FE MO NB NI TA TI V W: > also delta, Ni3Mo, Ni3Nb, Strukturbericht D0a, prototype Cu3Ti
NI3TI_D024	2 SUBL, SITES 0.75: 0.25: :AL CO CR CU FE HF NI TA TI W ZR:AL CR CU HF MO NB NI SI TA TI W ZR: > also Eta, AlNi6Ta
NI3Y	2 SUBL, SITES 3.00: 1.00: :FE NI:Y: > also Fe3Y
RH3ZR	2 SUBL, SITES 3.00: 1.00: :RH ZR:RH ZR:
RUY3	2 SUBL, SITES 0.25: 0.75: :RU:Y:
SN3Y	2 SUBL, SITES 3.00: 1.00: :SN:Y:
TA3SN	2 SUBL, SITES 3.00: 1.00: :TA:SN:
V3SN	2 SUBL, SITES 0.20: 0.80: :SN:V:
VZN3_L12	2 SUBL, SITES 1.00: 3.00: :V:ZN:
YZN3	2 SUBL, SITES 1.00: 3.00: :Y:ZN:
ZN3ZR_HT	2 SUBL, SITES 3.00: 1.00: :ZN:ZR:
AL13CO4	2 SUBL, SITES 13.00: 4.00: :AL:CO:
AL5CO2_D811	2 SUBL, SITES 5.00: 2.00: :AL:CO RH: > also Al5Rh2

AL9CO2	2	SUBL, SITES 9.00: 2.00:
:AL:CO RH:		
> also Al9Rh2		
AL11CR2	3	SUBL, SITES 10.00: 1.00: 2.00:
:AL:AL:CR:		
AL13CR2	2	SUBL, SITES 13.00: 2.00:
:AL:CR:		
AL4CR	2	SUBL, SITES 4.00: 1.00:
:AL:CR:		
AL8CR5_H	2	SUBL, SITES 8.00: 5.00:
:AL:CR:		
AL8CR5_L	2	SUBL, SITES 8.00: 5.00:
:AL:CR:		
> Strukturbericht D810		
AL9CR4_H	2	SUBL, SITES 9.00: 4.00:
:AL:CR:		
AL9CR4_L	2	SUBL, SITES 9.00: 4.00:
:AL:CR:		
ALCU_ZETA	2	SUBL, SITES 9.00: 11.00:
:AL:CU FE:		
AL23CUFE4	3	SUBL, SITES 23.00: 1.00: 4.00:
:AL:CU:FE:		
AL62CU25FE13	3	SUBL, SITES 0.12: 0.26: 0.62:
:FE:AL CU:AL:		
AL7CU2FE	3	SUBL, SITES 1.00: 2.00: 7.00:
:FE NI:CU:AL:		
> Solution phase of the ternary compound Al7Cu2Fe		
AL10CU10FE	3	SUBL, SITES 1.00: 10.00: 10.00:
:FE:AL CU:AL:		
AL7CU4NI	2	SUBL, SITES 1.00: 1.00:
:AL:FE CU NI VA:		

AL28CU4MN7	3 SUBL, SITES 28.00: 7.00: 4.00:
:AL:MN:CU:	
AL11CU5MN3	3 SUBL, SITES 11.00: 3.00: 5.00:
:AL:MN:CU:	
ALCU3MN2	3 SUBL, SITES 1.00: 2.00: 3.00:
:AL:MN:CU:	
AL13FE4	3 SUBL, SITES 0.63: 0.23: 0.14:
:AL CU:FE MN RU:AL SI VA:	
> solution phases based on Al <sub>13</sub> Fe <sub>4</sub> (aka Al <sub>3</sub> Fe) and Al <sub>13</sub> Ru <sub>4</sub>	
AL5FE2	2 SUBL, SITES 5.00: 2.00:
:AL CU:FE MN:	
AL5FE4	
:AL CU FE:	
AL16FEMN3	2 SUBL, SITES 4.00: 1.00:
:AL:FE MN:	
AL13FE2MN2	2 SUBL, SITES 4.00: 13.00:
:FE MN:AL:	
AL10FEMN2	2 SUBL, SITES 3.00: 10.00:
:FE MN:AL:	
AL9IR2	2 SUBL, SITES 0.82: 0.18:
:AL:IR:	
AL45IR13	2 SUBL, SITES 0.78: 0.22:
:AL:IR:	
AL13IR4	2 SUBL, SITES 0.77: 0.23:
:AL:IR:	
AL28IR9	2 SUBL, SITES 0.76: 0.24:
:AL:IR:	
AL2_7IR	2 SUBL, SITES 0.73: 0.27:
:AL:IR:	
AL12MN	2 SUBL, SITES 12.00: 1.00:
:AL:MN:	

AL4MN_R	2	SUBL, SITES 461.00:107.00:
:AL:MN FE:		
> AL461MN107		
AL4MN_U	2	SUBL, SITES 4.00: 1.00:
:AL:MN:		
AL11MN4_LT	2	SUBL, SITES 11.00: 4.00:
:AL:MN FE:		
AL11MN4_HT	2	SUBL, SITES 29.00: 10.00:
:AL MN:MN:		
AL8MN5	3	SUBL, SITES 12.00: 5.00: 9.00:
:AL TI ZN:MN:AL CU MN SI TI:		
AL6MN	2	SUBL, SITES 6.00: 1.00:
:AL:FE MN RE RU:		
> also Al6Re, Al6Ru		
AL31MN6Ni2	3	SUBL, SITES 31.00: 6.00: 2.00:
:AL:MN:NI:		
> Orthorhombic, ternary Al-Mn-Ni phase		
ALMNTI_L12	3	SUBL, SITES 0.25: 0.08: 0.67:
:AL MN TI:AL MN:AL MN TI:		
> Ti25MN9Al66_L12		
AL63MO37	2	SUBL, SITES 63.00: 37.00:
:AL:MO:		
AL8MO3	2	SUBL, SITES 8.00: 3.00:
:AL:MO:		
ALMO_A2	2	SUBL, SITES 1.00: 1.00:
:AL MO:AL MO:		
> Strukturbericht A2, it was badly described		
AL3Ni5	2	SUBL, SITES 0.38: 0.62:
:AL:NI:		
AL3Ni2_D513	3	SUBL, SITES 3.00: 2.00: 1.00:
:AL SI SN ZN:AL CU NI RU:RU NI VA:		

> Al <sub>3</sub> Ni <sub>2</sub> , Al <sub>3</sub> Ru <sub>2</sub>	
AL11RE4	2 SUBL, SITES 11.00: 4.00: :AL:RE:
AL4RE	2 SUBL, SITES 4.00: 1.00: :AL:RE:
AL3RH_HT	2 SUBL, SITES 2.00: 1.00: :AL:RH VA:
AL7RH3	2 SUBL, SITES 7.00: 3.00: :AL:RH:
AL15Si2M4	3 SUBL, SITES 14.00: 4.00: 5.00: :AL:FE MN:AL Si: > Solution of Al-Mn-Si ternary phase, tao 9, Al <sub>15</sub> (Mn,Fe) <sub>3</sub> Si <sub>2</sub>
AL3TI_LT	2 SUBL, SITES 3.00: 1.00: :AL TI:AL TI ZR:
AL5TI2_HT	2 SUBL, SITES 5.00: 2.00: :AL TI:AL NB TA TI V ZR:
AL5TI3	2 SUBL, SITES 5.00: 3.00: :AL:TA TI:
AL10V	2 SUBL, SITES 10.00: 1.00: :AL:V:
AL7V	2 SUBL, SITES 7.00: 1.00: :AL:V:
AL23V4	2 SUBL, SITES 23.00: 4.00: :AL:V:
AL8V5	2 SUBL, SITES 8.00: 5.00: :AL:V:
AL77W23	2 SUBL, SITES 77.00: 23.00: :AL:W:
AL7W3	2 SUBL, SITES 7.00: 3.00: :AL:W:
AL12W	2 SUBL, SITES 12.00: 1.00:

:AL:MO RE W: > also Al12Mo, Al12Re
AL4W            2 SUBL, SITES 4.00: 1.00:  :AL:MO W: > also Al4Mo
AL5W            2 SUBL, SITES 5.00: 1.00:  :AL:MO W: > also AL5MO
AL2ZR3          2 SUBL, SITES 2.00: 3.00:  :AL:HF TI Y ZR: > also Al2Hf3, Al2Y3
AL3ZR4          2 SUBL, SITES 3.00: 4.00:  :AL:HF TI ZR: > also Al3Hf4
AL3ZR2          2 SUBL, SITES 3.00: 2.00:  :AL:HF ZR: > also Al3Hf2
AL3ZR5          2 SUBL, SITES 3.00: 5.00:  :AL:TI ZR: > Strukturbericht D8m, prototype Si3W5
AL4ZR5          2 SUBL, SITES 4.00: 5.00:  :AL:ZR:
CO7HF           2 SUBL, SITES 7.00: 1.00:  :CO:HF:
CO10CU57TI33    3 SUBL, SITES 0.10: 0.57: 0.33:  :CO:CU:TI:
CO7M2           2 SUBL, SITES 7.00: 2.00:  :CO:NB TA: > Co7Nb2, Co7Ta2
CO17Y2          3 SUBL, SITES 1.00: 2.00: 15.00:  :CO2 Y:CO2 Y:CO:

CO5Y_D2D	3	SUBL, SITES	1.00: 4.00: 1.00:
:CO2 Y:CO:CO VA:			
CO3Y2	2	SUBL, SITES	3.00: 2.00:
:CO:Y:			
CO7Y6	2	SUBL, SITES	7.00: 6.00:
:CO:Y:			
CO3Y4	2	SUBL, SITES	3.00: 4.00:
:CO:Y:			
CO5Y8	2	SUBL, SITES	5.00: 8.00:
:CO:Y:			
COZN_LT	2	SUBL, SITES	1.00: 1.00:
:CO ZN:VA:			
COZN_HT	2	SUBL, SITES	1.00: 1.00:
:CO ZN:VA:			
COZN_GAMMA_D82	2	SUBL, SITES	1.00: 1.00:
:CO ZN:VA:			
> Zn11Co2 Prototype Zn9(Zn0.5Fe0.5)2Fe2 (cI52, I-43m)			
COZN_DELTA	2	SUBL, SITES	0.12: 0.88:
:CO:ZN:			
COZN_GAMMA1	2	SUBL, SITES	0.12: 0.88:
:CO:ZN:			
COZN_GAMMA2	2	SUBL, SITES	0.07: 0.93:
:CO:ZN:			
CO11ZR2	2	SUBL, SITES	11.00: 2.00:
:CO:ZR:			
CRMN3_HT_SIGMA	3	SUBL, SITES	8.00: 4.00: 18.00:
:MN:CR:CR MN:			
CR3MN5	2	SUBL, SITES	3.00: 5.00:
:CR:MN:			
CRZN13	2	SUBL, SITES	1.00: 13.00:
:CR:ZN:			

CRZN17	2	SUBL, SITES 1.00: 17.00:
:CR:ZN:		
CU51HF14	2	SUBL, SITES 51.00: 14.00:
:CU:HF:		
CU8HF3	2	SUBL, SITES 8.00: 3.00:
:CU:HF:		
CU10HF7	2	SUBL, SITES 10.00: 7.00:
:CU:HF:		
CU5MN4SI	3	SUBL, SITES 0.50: 0.37: 0.13:
:CU:MN:SI:		
CU6NISI3	2	SUBL, SITES 0.73: 0.27:
:CU NI:SI:		
CU46NI25SI29	3	SUBL, SITES 0.46: 0.25: 0.29:
:CU:NI:SI:		
CU6SN5_HT_NIAS	3	SUBL, SITES 1.00: 1.00: 1.00:
:CO CU MN NI VA:AL CU NI SN:CO CU MN NI VA:		
> Cu6Sn5_HT, Co3Sn2, Mn(2-x)Sn, Ni3Sn2		
CU33SI7_DELTA	2	SUBL, SITES 0.82: 0.17:
:CU:SI:		
CU15SI4_EPSILON	2	SUBL, SITES 0.79: 0.21:
:CU MN:AL SI:		
CU56SI11_GAMMA	2	SUBL, SITES 0.84: 0.16:
:CU MN NI SI:SI:		
CU56SI11_GAMMA	2	SUBL, SITES 0.84: 0.16:
:CU MN NI SI:SI:		
CUSI_ETA	2	SUBL, SITES 0.76: 0.24:
:CU MN NI:SI:		
CU6SN5_LT	3	SUBL, SITES 1.00: 1.00: 1.00:
:CU:CU SN:SN:		
CU10SN3	2	SUBL, SITES 0.77: 0.23:
:CU NI:SN:		

CU41SN11	2	SUBL, SITES 41.00: 11.00:
:CU SN ZN:CU SN ZN:		
> Cu41Sn11 with solubility of Zn.		
CU3TI2	2	SUBL, SITES 3.00: 2.00:
:CU NI FE:CO TI:		
CU4TI1	2	SUBL, SITES 4.00: 1.00:
:CU TI:CU TI:		
CU4TI3	2	SUBL, SITES 4.00: 3.00:
:CO CU NI:TI:		
CU2TIZR	3	SUBL, SITES 0.50: 0.25: 0.25:
:CU:TI:ZR:		
CU7Y1	2	SUBL, SITES 1.00: 5.00:
:CU2 Y:CU:		
CU4Y	2	SUBL, SITES 4.00: 1.00:
:CU:Y:		
CU7Y2	2	SUBL, SITES 7.00: 2.00:
:CU:Y:		
CU10ZR7	2	SUBL, SITES 10.00: 7.00:
:CU:ZR:		
CU51ZR14	2	SUBL, SITES 51.00: 14.00:
:CU:ZR:		
CU8ZR3	2	SUBL, SITES 8.00: 3.00:
:CU:ZR:		
FENBZR_CFC2	3	SUBL, SITES 2.00: 1.00: 3.00:
:FE NB ZR:NB ZR:NB ZR:		
FE3SN2	2	SUBL, SITES 3.00: 2.00:
:FE:SN:		
FE5SN3	2	SUBL, SITES 5.00: 3.00:
:FE:SN:		
FEZN_GAMMA_D82	4	SUBL, SITES 0.15: 0.15: 0.23: 0.46:
:FE ZN:FE ZN:FE ZN:ZN:		

FEZN_GAMMA_D81	3	SUBL, SITES 0.14: 0.12: 0.74:
:FE:FE ZN:ZN:		
FEZN_DELTA	4	SUBL, SITES 0.06: 0.18: 0.53: 0.24:
:FE:FE ZN:ZN:ZN:		
FEZN_ZETA	3	SUBL, SITES 0.07: 0.86: 0.07:
:FE VA:ZN:VA ZN:		
GAMMA_D03		
:CU MN NI SN ZN:		
> Cu3Sn solution, Prototype BiF3 (cF16, Fm-3m)		
HF5IR3	2	SUBL, SITES 5.00: 3.00:
:HF:IR:		
HF3NI7	2	SUBL, SITES 0.30: 0.70:
:HF:NI:		
HF8NI21	2	SUBL, SITES 8.00: 21.00:
:HF ZR:NI:		
> also ZR8NI21, prototype Hf8Ni21		
HF3RH4	2	SUBL, SITES 3.00: 4.00:
:HF:RH:		
HF3RH5	2	SUBL, SITES 3.00: 5.00:
:HF:RH:		
HF5SN4	2	SUBL, SITES 5.00: 4.00:
:HF:SN:		
IR2Y3	2	SUBL, SITES 2.00: 3.00:
:IR:Y:		
IR2Y5	2	SUBL, SITES 2.00: 5.00:
:IR:Y:		
IR3Y5	2	SUBL, SITES 3.00: 5.00:
:IR:Y:		
IR3ZR5	2	SUBL, SITES 3.00: 5.00:
:IR:ZR:		
MN3SN2	2	SUBL, SITES 3.00: 2.00:

:MN:SN:	
MNTI_HT	2 SUBL, SITES 0.52: 0.48:
:MN:TI:	
MN4TI	2 SUBL, SITES 0.81: 0.18:
:MN:TI:	
MN12Y	2 SUBL, SITES 12.00: 1.00:
:MN:Y:	
MNZN9	2 SUBL, SITES 1.00: 9.00:
:MN:ZN:	
MO7IR3	2 SUBL, SITES 0.70: 0.30:
:MO:IR:	
MONI4_BETA	2 SUBL, SITES 1.00: 4.00:
:MO W:NI:	
> also WNi4, Strukturbericht D1a	
MONI_DELTA	3 SUBL, SITES 24.00: 20.00: 12.00:
:CO CR FE NI RE:CO CR FE MO NI RE W:CU MO W:	
MOZN7	2 SUBL, SITES 1.00: 7.00:
:MO:ZN:	
MOZN22	2 SUBL, SITES 1.00: 22.00:
:MO:ZN:	
NB15NI56TI29	3 SUBL, SITES 0.15: 0.56: 0.29:
:NB:NI:TI:	
NB8NI9TI3	3 SUBL, SITES 0.40: 0.45: 0.15:
:NB:NI:TI:	
NB5NI75TI20	3 SUBL, SITES 0.05: 0.75: 0.20:
:NB:NI:TI:	
NB13NI75TI12	3 SUBL, SITES 0.13: 0.75: 0.12:
:NB:NI:TI:	
NB15NI80TI5	3 SUBL, SITES 0.15: 0.80: 0.05:
:NB:NI:TI:	
NB3RU5	2 SUBL, SITES 0.38: 0.62:

:NB RU:RU:	
NBSN2	2 SUBL, SITES 1.00: 2.00:
:NB SN V:NB SN:	
NBZN7	2 SUBL, SITES 0.12: 0.88:
:NB:ZN:	
NBZN15	2 SUBL, SITES 0.06: 0.94:
:NB:ZN:	
NB2ZN3	2 SUBL, SITES 0.40: 0.60:
:NB:ZN:	
NI8M	2 SUBL, SITES 8.00: 1.00:
:NI:NB TA:	
> Ni8Ta, Ni8Nb	
NI7ZR2	2 SUBL, SITES 7.00: 2.00:
:AL CO CR NI:HF Y ZR:	
> also NI7HF2, NI7Y2, CO7HF2 and CO7Y2	
NI3SN4	3 SUBL, SITES 0.25: 0.25: 0.50:
:CU NI:NI SN:SN:	
NI2Y3	2 SUBL, SITES 2.00: 3.00:
:NI:Y:	
NI4Y	2 SUBL, SITES 4.00: 1.00:
:NI:Y:	
NI17Y2	2 SUBL, SITES 1.00: 0.12:
:AL FE NI:Y:	
> also Fe17Y2	
NIZN8_DELTA	2 SUBL, SITES 0.11: 0.89:
:NI:AL ZN:	
NI11ZR9	2 SUBL, SITES 11.00: 9.00:
:NI:HF ZR:	
> also Ni11Hf9	
NI10ZR7	2 SUBL, SITES 23.00: 17.00:
:NI:HF ZR:	

> also Ni10Hf7	
Ni5ZR_C15B	2 SUBL, SITES 5.00: 1.00: :AL CU NI:HF Y ZR: > also Ni5Y, Ni5Hf, Cu5Hf and Cu5Zr, Strukturbericht C15b
RHSN4	2 SUBL, SITES 1.00: 4.00: :RH:SN:
RH3SN2	3 SUBL, SITES 0.12: 0.50: 0.38: :RH:RH:SN:
RH5TI3	2 SUBL, SITES 5.00: 3.00: :RH:TI:
RH4ZR3_LT	2 SUBL, SITES 4.00: 3.00: :RH ZR:ZR: > alpha_Rh4Zr3
RH4ZR3_HT	2 SUBL, SITES 4.00: 3.00: :RH ZR:ZR: > beta_Rh4Zr3
RH5ZR3	2 SUBL, SITES 5.00: 3.00: :RH:RH ZR:
RU3SN7	2 SUBL, SITES 0.30: 0.70: :RU:SN:
RU2SN3	2 SUBL, SITES 0.40: 0.60: :RU:SN:
RU2Y3	2 SUBL, SITES 0.40: 0.60: :RU:Y:
RU25Y44	2 SUBL, SITES 0.36: 0.64: :RU:Y:
RU2Y5	2 SUBL, SITES 0.29: 0.71: :RU:Y:
SN5TI6	2 SUBL, SITES 5.00: 6.00: :AL SN:NB TI: > also Sn5Nb6

SN5Y2	2 SUBL, SITES 5.00: 2.00:
:SN:Y:	
SN10Y11	2 SUBL, SITES 10.00: 11.00:
:SN:Y:	
SN4Y5	2 SUBL, SITES 4.00: 5.00:
:SN:Y:	
TAAL	2 SUBL, SITES 0.52: 0.48:
:TA:AL:	
TAAL2	2 SUBL, SITES 0.35: 0.65:
:TA:AL:	
TA41IR59	2 SUBL, SITES 0.41: 0.59:
:TA:IR:	
TA43IR57	2 SUBL, SITES 0.43: 0.57:
:TA:IR TA:	
T1CUNITI	2 SUBL, SITES 2.00: 1.00:
:CU NI:TI:	
> the Cu-Ni-Ti ternary phase, Tau1	
T2CUNITI	3 SUBL, SITES 0.17: 2.83: 2.00:
:CU:NI:TI:	
> the Cu-Ni-Ti ternary phase, Tau2	
T4CUNITI	3 SUBL, SITES 0.05: 0.70: 0.25:
:CU:NI:TI:	
> the Cu-Ni-Ti ternary phase, Tau4	
T6CUNITI	3 SUBL, SITES 0.25: 0.50: 0.25:
:CU:NI:TI:	
> the Cu-Ni-Ti ternary phase, Tau6	
T1_CU2TI	2 SUBL, SITES 2.00: 1.00:
:CU FE:TI:	
> the Cu-Fe-Ti ternary phase, Tau1	
T2_CU3TI2	2 SUBL, SITES 3.00: 2.00:
:CU FE:TI:	

> the Cu-Fe-Ti ternary phase, Tau2	
T3_CU4TI3	2 SUBL, SITES 4.00: 3.00: :CU FE:TI: > the Cu-Fe-Ti ternary phase, Tau3
T4CUFETI	2 SUBL, SITES 0.63: 0.37: :CU FE:TI: > the Cu-Fe-Ti ternary phase, Tau4
T5CUFETI	2 SUBL, SITES 0.55: 0.45: :CU FE:TI: > the Cu-Fe-Ti ternary phase, Tau5
TIZN5	2 SUBL, SITES 1.00: 5.00: :TI:ZN:
TIZN10	2 SUBL, SITES 1.00: 10.00: :TI:ZN:
TIZN15	2 SUBL, SITES 1.00: 15.00: :TI:ZN:
V4ZN5	2 SUBL, SITES 4.00: 5.00: :V:ZN:
Y2ZN17	2 SUBL, SITES 2.00: 17.00: :Y:ZN:
Y13ZN58	2 SUBL, SITES 13.00: 58.00: :Y:ZN:
YZN5	2 SUBL, SITES 1.00: 5.00: :Y:ZN:
ZN12Y	2 SUBL, SITES 12.00: 1.00: :ZN:Y:
ZN11Y3	2 SUBL, SITES 11.00: 3.00: :ZN:Y:
ZN22ZR	2 SUBL, SITES 22.00: 1.00: :ZN:ZR:
ZN39ZR5	2 SUBL, SITES 39.00: 5.00:

:ZN:ZR:	
ZN2ZR3	2 SUBL, SITES 2.00: 3.00:
:ZN:ZR:	
GAS:G	
:N2:	
IRNB_L10	2 SUBL, SITES 1.00: 1.00:
:IR NB:IR NB:	
NI3SI_MONOCL	2 SUBL, SITES 3.00: 1.00:
:NI:SI:	
GPHASE_ALCOTI	4 SUBL, SITES 1.00: 16.00: 7.00: 6.00:
:AL CO:AL TI:CO:TI:	
AL5CU4ZN	4 SUBL, SITES 1.00: 4.00: 4.00: 1.00:
:AL CU:AL:CU:ZN:	
AL24MN5ZN	3 SUBL, SITES 5.00: 1.00: 24.00:
:MN ZN:ZN:AL:	
AL9MN2ZN	3 SUBL, SITES 2.00: 1.00: 9.00:
:MN:ZN:AL:	
AL11MN3ZN2	3 SUBL, SITES 3.00: 2.00: 11.00:
:MN:ZN:AL:	
O_PHASE	3 SUBL, SITES 0.50: 0.25: 0.25:
:NB TA TI:AL NB TA TI:NB TA TI:	
> The O phase	
O1_DIS	2 SUBL, SITES 0.75: 0.25:
:AL NB TI:AL NB TI:	
> The disordered O phase	
ALNI2ZN	3 SUBL, SITES 0.25: 0.50: 0.25:
:AL:NI:ZN:	
AL13NI38ZN49	3 SUBL, SITES 0.13: 0.38: 0.49:
:AL:NI:ZN:	
> Al-Ni-Zn ternary phase	
ALSI3TI2	3 SUBL, SITES 0.17: 0.50: 0.33:

---

:AL:SI:TI:	
B82_OMEGA	3 SUBL, SITES 1.00: 1.00: 1.00:
:AL SN:CO NB SN TA TI:TI:	
TAU_COSNTI	3 SUBL, SITES 1.00: 1.00: 1.00:
:CO NI:SN:TI:	
CU4MNSN	3 SUBL, SITES 0.67: 0.17: 0.17:
:CU:SN:MN:	
CUMNZN	3 SUBL, SITES 0.33: 0.33: 0.33:
:CU:MN:ZN:	