

TCAL4 - TCS Al-based alloy database, Version 4.0

TCAL4 is a thermodynamic database for Al-based alloys for use with the Thermo-Calc, DICTRA and TC-PRISMA software packages. TCAL4 is based on the critical evaluation of binary, ternary and important higher-order systems which enables predictions to be made for multi-component systems and alloys of industrial importance. A hybrid approach of experiments, first-principles calculations and CALPHAD modeling has been used to obtain thermodynamic descriptions of the constituent binary and ternary systems over the whole composition and temperature ranges. The database has been validated where possible against higher-order systems and is the result of a long-term collaboration with academia that has involved extensive experimental work, theoretical calculations and critical assessments of the published literature. The database has been developed in a 34-element framework:

Al	Cu	Fe	Li	Mg	Mn	Ni	Si	Zn
B	C	Cr	Ge	Sn	Sr	Ti	V	Zr
Ag	Ca	H	Hf	K	La	Na	Sc	
Be	Bi	Cd	Ce	Co	Ga	In	Pb	

In total, 240 of the binary systems in this 34-element framework have been assessed, mostly to their full range of composition, and can be calculated with the BINARY Module in Thermo-Calc. TCAL4 also contains assessments of 62 ternaries, and these can be calculated with the TERNARY Module in Thermo-Calc. Many quaternaries with Al as one element have also been assessed.

TCAL4 contains 528 solution and intermetallic phases in total. The GAS phase is rejected by default when retrieving the data from the database. One has to manually restore it when it is required for a calculation. Phase diagrams of assessed binary and ternary systems can be conducted using the BINARY and TERNARY modules, respectively. The complete list of phases is given at the end of this document. First there is a list of all phases and then a detailed description of their models, e.g. number of sublattices and constituents on each sublattice.

TCAL4 includes nearly all stable phases in the assessed systems and most important metastable phases (or precipitates) that may form in as-cast and aged Al-based alloys. The database can be used to calculate various phase diagrams and property diagrams in the assessed systems or even extrapolated higher-order systems. The extrapolation to higher-order systems helps to understand the phase equilibria in multi-component industrial aluminum alloys, so as to predict the phase formation, phase fractions and phase compositions or to calculate the driving force of forming a phase. Note that the extrapolation to higher-order systems might not be valid beyond the Al-rich region. The database can also be used for predicting solidification behaviour of Al-alloys with the SCHEIL_GULLIVER module in Thermo-Calc and simulating multi-particle precipitation during aging treatment with the TC-PRISMA software.

Some ordered compounds with the structure of L1₂ or B2 were modeled as stoichiometric phases, and named as L1₂_FCC and B2_BCC, respectively. In general, however, the ordered B2 and L1₂ phases, together with bcc_A2 and fcc_A1, respectively, have been modeled with the so-called partition model, which describes an ordered phase and its disordered counterpart using a single Gibbs energy curve. This type of description is of particular importance to be able to predict second order transformations between a disordered phase and its ordered structures. Also note that there

may be several possible composition sets for the phases named FCC_L12 and BCC_B2, due to the co-existence of disordered and ordered structures or the presence of miscibility gap.

MAJOR UPDATES FROM TCAL1.0 TO TCAL1.1

The description of the Al-Zn-Mg-Cu-Fe core system has been systematically refined and validated in order to give more accurate predictions for commercial Al-based alloys, especially the 7xxx series alloys. More specifically, crucial corrections or modifications have been made for the following related ternary systems, Al-Cu-Fe, Al-Cu-Mg, Al-Cu-Zn, and Al-Mg-Zn.

Another major enhancement is that users can now get the conventional phase names in Al-based alloys for a general name used in the database by using the command LIST_SYSTEM CONSTITUENT in the TDB module.

MAJOR UPDATES FROM TCAL1.1 TO TCAL1.2

The Cu-Li, Li-Mg, Al-Cu-Li and Al-Li-Mg systems have been assessed and/or implemented in order to be able to predict the phase formation in Al-Cu-Li-Mg(-Zn) alloys (i.e. some of the 2xxx and 8xxx series alloys). The descriptions of the Al-Cu-Mg-Si and Al-Fe-Mn-Si core systems have been refined and validated, in order to give more accurate predictions for commercial Al-based alloys, including wrought alloys from series 2xxx to series 7xxx and foundry alloys series 3xx.x. The Al-Cr-Si system was tentatively assessed to include the Cr-bearing phase $\text{Al}_{13}\text{Cr}_4\text{Si}_4$.

The two compounds, AL8FEMNSI2 and AL5CU2MN3, were removed from the database since their existences were disputed. The VSI2 phase was merged into CRSI2_C40, and the AB3_L12 phase into L12_FCC. Thermodynamic models were reviewed for most phases, and many un-assessed parameters were reasonably estimated. Some phases were renamed to use their conventional names.

Additionally, the C-Mg binary description was reassessed. Now the two Mg carbides, MgC_2 and Mg_2C_3 , are metastable and the C solubility in liquid Mg is greatly reduced to accord with the mostly published experimental data.

MAJOR UPDATES FROM TCAL1.2 TO TCAL2.0

Since TCAL2, all necessary volume data (including molar volume and thermal expansion) had been added for most of the solution phases and intermetallic phases. This allows for the calculation of volume fraction of phases, as well as density, thermal expansivity and lattice parameters using Thermo-Calc. However, it should be noted that the molar volume data incorporated has no pressure dependence.

21 more binary systems have been implemented: Al-Be, Al-Bi, Al-Cd, Al-Ce, Al-Co, Al-Ga, Al-In, Al-Pb, Bi-Cu, Cd-Cu, Ce-Cu, Co-Li, Cr-Li, Cu-Co, Cu-Ga, Cu-In, Cu-Pb, Cu-Sc, Li-Ni, Li-Si and Li-Zr. Some of them were reassessed in this project. Additionally, the Al-Ca and Al-Sc descriptions have been updated. The AlLi_2 phase was considered in Al-Li.

The three ternary systems, Al-C-Si, Al-Cu-Sc, and Al-Li-Si, have been newly implemented. The previous provisional description of the Al-Cr-Si system has been replaced by a much more reliable description, which is derived from a thorough thermodynamic modeling over the entire compositional range and a wide temperature range. The Mn-Ni-Si description is also updated.

MAJOR UPDATES FROM TCAL2.0 TO TCAL2.1.1

The Al-Fe-Mn-Si quaternary description had been systematically refined, including a deep revision of the Al-Fe-Si description and adjustments of the Al-Fe-Mn and Al-Mn-Si descriptions. It has been validated that this refinement improved the phase formation in a wide range of casting and wrought aluminum alloys, since Fe, Mn and Si are the most common additives and/or impurities in aluminum alloys.

A new Al-Ni description had been adopted and adjustments were subsequently made on the Al-Ni-based ternary systems.

The BCC_B2 description in the Ni-Zn binary system was reassessed. The Al-Ni-Zn ternary was reassessed. The Al-Mn-Ni description was improved by solving some known issues.

Both the Al-Cr and Al-V binary systems were improved in the Al-Rich corner. The Al-Si molar volume data were refined.

MAJOR UPDATES FROM TCAL2.1.1 TO TCAL3.0

19 binary systems were added, Ag-Ca, Ca-Cu, Ca-Fe, Ca-H, Ca-La, Ca-Li, Ca-Mn, Ca-Na, Ca-Ni, Ca-Sc, Ca-Si, Ca-Sr, Ca-Zn, Ag-Sc, Fe-Sc, Mn-Sc, Ni-Sc, Sc-Si and Sc-Zr.

Modeling of Al-Cu metastable precipitates: GPI Zones (described as the miscibility gap of fcc_A1), θ'' -Al₃Cu (i.e. GPII Zones) and θ' -Al₂Cu.

Modeling of Al-Cu-Mg-Zn metastable phases: S'-Al₂CuMg, T'-Al_{0.3}Mg_{0.4}Zn_{0.3} and η' -Al₃Mg_{2.5}Zn_{3.5}.

Modeling of Al-Mg-Si metastable precipitates: β'' -Mg₅Si₆ (GPII zones), Al-containing β'' -Al₂Mg₅Si₄, β' -Mg₉Si₅, U1-Al₂MgSi₂, U2-Al₄Mg₄Si₄ and B'-Al₃Mg₉Si₇.

Modeling of the metastable Al_mFe phase (modeled as Al₄Fe), which has been observed in some as-cast aluminum alloys such as AA1xxx, AA5128 and A206.

Necessary volume data were assessed for the new phases and newly introduced end-members. The Sn-Zn and Cu-Fe-Ni descriptions were updated. Some known issues were solved.

MAJOR UPDATES FROM TCAL3.0 TO TCAL4.0

The metastable Al-Cu precipitate Ω was modeled as the Al₂Cu_OMEGA phase. In the Al-Cu-Mg-Zn system, the descriptions of the metastable precipitates ETA_PRIME (η') and T_PRIME (T') were refined. In the Al-Mg-Si system, the BETA_AL_DPRIME (Al-containing β'') phase was merged into BETA_DPRIME (β'') and treated as the same phase.

53 Ag-, H-, Hf-, K-, La-, Li-, Na- and/or Sc-containing binary systems were added, Ag-Cu, Ag-Fe, Ag-La, Ag-Li, Ag-Mg, Ag-Mn, Ag-Na, Ag-Ni, Ag-Si, Ag-Zn, Cu-H, Cu-Hf, Cu-La, Cu-Na, Fe-H, Fe-Hf, Fe-K, Fe-La, Fe-Li, Fe-Na, H-K, H-La, H-Li, H-Mg, H-Mn, H-Na, H-Ni, H-Zn, Hf-K, Hf-Li, Hf-Mg, Hf-Mn, Hf-La, Hf-Ni, Hf-Sc, Hf-Si, K-Li, K-Mg, K-Na, K-Zn, La-Mn, La-Ni, La-Sc, La-Si, La-Zn, Li-Mn, Li-Na, Li-Sc, Li-Zn, Na-Sc, Na-Si, Na-Zn, and Sc-Zn. The previous Ag-Al binary description was replaced.

HCP_ZN was merged into HCP_A3. Necessary adjustments were made for the descriptions of Zn-containing systems in order to reproduce the phase equilibria.

Zr was introduced to Al₃Ti_D022 and Ti to Al₃Zr_D023. Al-Ti was updated in the Al-rich region. A preliminary assessment of the Al-Ti-Zr system was conducted. The description of Al₄Mn_R in the Al-Fe-Mn system was refined.

Molar volumes and thermal expansivities were evaluated for all the newly added phases and end-members. Some existing volume data were updated as well.

Assessed binary systems in full range of composition and temperature:

	Ag	Al	B	Bc	Bi	C	Ca	Ce	Cd	Co	Cr	Cu	Fe	Ga	Ge	H	Hf	In	K	La	Li	Mg	Mn	Na	Ni	Pb	Sc	Si	Sr	Ti	V	Zn	Zr
Al	2																																
B		2																															
Be		2																															
Bi		2																															
C		2	2																														
Ca	1	2																															
Ce		2																															
Cd		2																															
Co		2																															
Cr		2	2																														
Cu	2	2	2			2	2	2	2	2	2	2	2																				
Fe	2	2	2				2	2					2	2																			
Ga		2												2																			
Ge		2	2			2								2	2	2																	
H		2					2							2	2																		
Hf		2												2	2																		
In		2								2																							
K		2													2			2	2														
La	2	2						2						2	2					2													
Li	2	2						2					2	2	2	2				2													
Mg	2	2	2			2							2	2	2				2	2	2												
Mn	2	2	2			2	2						2	2	2				2	2	2												
Na	2	2					2						2	2					2	2		2											
Ni	2	2	2			2	2						2	2	2				2	2	2	2											
Pb		2											2																				
Sc	2	2						2						2	2				2				2	2	2								
Si	2	2	2			2	2						2	2	2			2			2	2	2	2		2							
Sn		2	2			2							2	2	2						2	2		2			2						
Sr	2	2			2	2							2	2	2						2	2		2			2		2				
Ti	2	2			2								2	2	2						2	2		2			2	2	2				
V	2	2			2								2	2	2						2	2		2			2	2	2	2			
Zn	2	2	2			2	2						2	2	2			2			2	2	2	2		2		2	2	2			
Zr		2	2			2							2	2	2						2	2		2			2	2	2	2	2		

Assessed ternary systems, mostly in full compositional ranges

Al-C-Si	Al-Cr-Si	Al-Cu-Fe	Al-Cu-Li	Al-Cu-Mg	Al-Cu-Mn
Al-Cu-Ni	Al-Cu-Sc	Al-Cu-Si	Al-Cu-Zn	Al-Fe-Mg	Al-Fe-Mn
Al-Fe-Ni	Al-Fe-Si	Al-Fe-Zn	Al-Li-Mg	Al-Li-Si	Al-Mg-Mn
Al-Mg-Ni	Al-Mg-Si	Al-Mg-Zn	Al-Mn-Ni	Al-Mn-Si	Al-Mn-Zn
Al-Ni-Si	Al-Ni-Zn	Al-Si-Zn	Cu-Fe-Mg	Cu-Fe-Mn	Cu-Fe-N
iCu-Fe-Si	Cu-Fe-Zn	Cu-Mg-Mn	Cu-Mg-Ni	Cu-Mg-Si	Cu-Mg-Zn
Cu-Mn-Ni	Cu-Mn-Si	Cu-Mn-Zn	Cu-Ni-Si	Cu-Ni-Zn	Cu-Si-Zn
Fe-Mg-Mn	Fe-Mg-Ni	Fe-Mg-Si	Fe-Mg-Zn	Fe-Mn-Ni	Fe-Mn-Si
Fe-Mn-Zn	Fe-Ni-Si	Fe-Ni-Zn	Fe-Si-Zn	Mg-Mn-Ni	Mg-Mn-Si
Mg-Mn-Zn	Mg-Ni-Si	Mg-Ni-Zn	Mg-Si-Zn	Mn-Ni-Si	Mn-Ni-Zn
Mn-Si-Zn	Ni-Si-Zn				

Assessed quaternary systems

Al-Cu-Mg-Zn	Al-Cu-Mg-Si	Al-Fe-Mn-Si	Al-Cu-Fe-Mn	Al-Cu-Fe-Ni	Al-Cu-Mg-Ni
Al-Cu-Mn-Si	Al-Cu-Ni-Si	Al-Fe-Mg-Mn	Al-Fe-Mg-Si	Al-Fe-Ni-Si	Al-Mg-Mn-Si

Examples of calculations¹

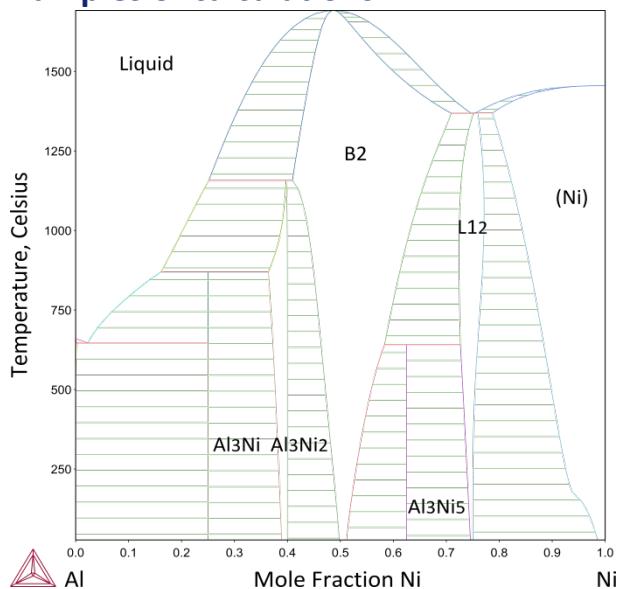


Fig.1: Calculated Al-Ni phase diagram^[1]

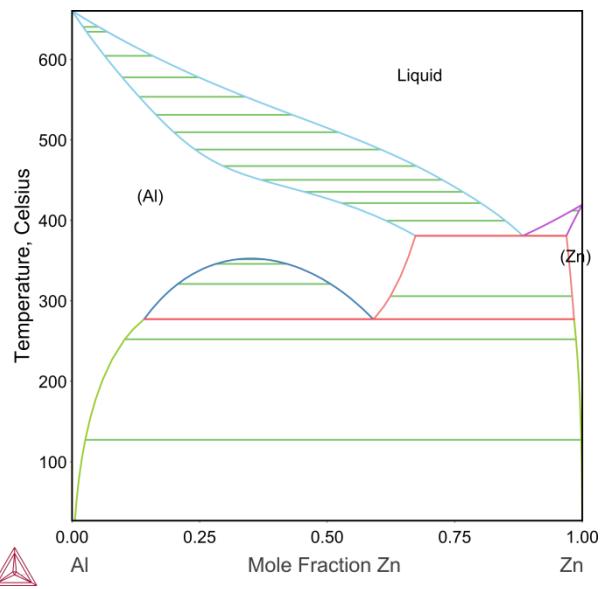


Fig.2: Calculated Al-Zn phase diagram^[2]

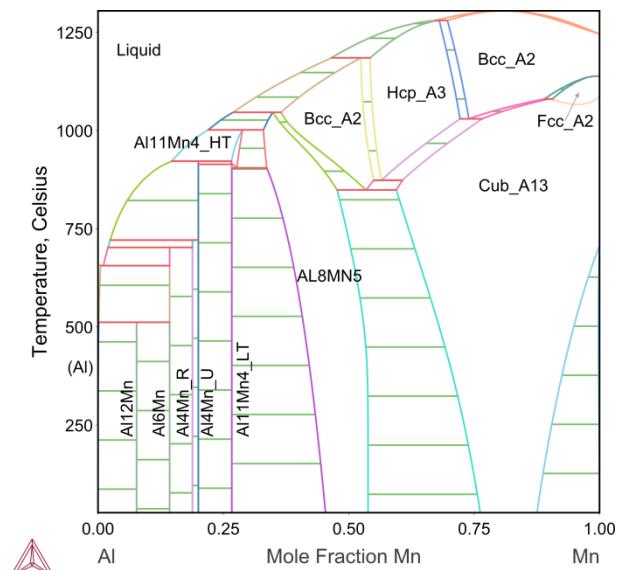


Fig.3: Calculated Al-Mn phase diagram^[3]

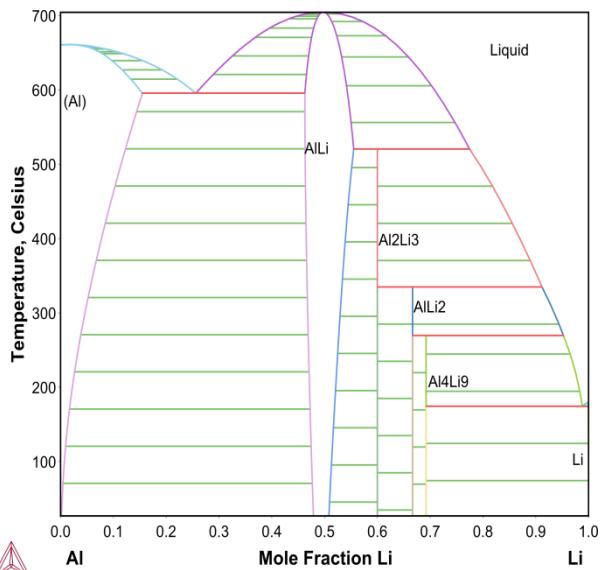


Fig.4: Calculated Al-Li phase diagram^[4]

¹ Some phase diagrams had been calculated with older versions of the TCAL database, so small differences might be observed if they are recalculated with TCAL4. For the systems, which had been considerably or significantly improved, their phase diagrams were recalculated with TCAL4.

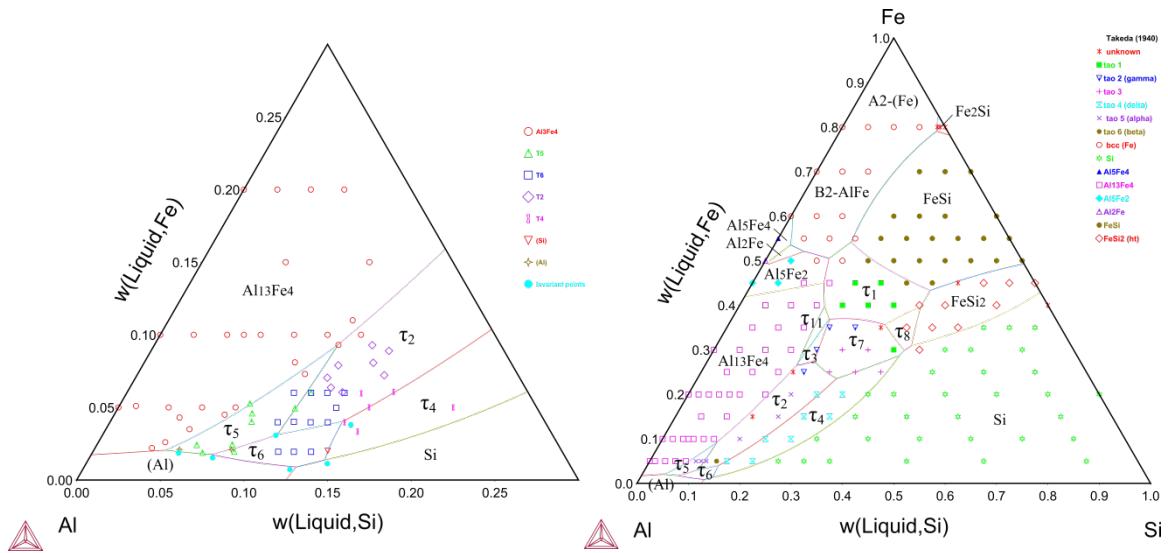


Fig.5: Calculated Al-Fe-Si liquidus projection (τ_5 : α -AlFeSi; τ_6 : β -AlFeSi), (a) in the Al-rich corner (the invariant points are from Pontevichi et al. [5, 6] and the remaining data from Takeda and Mutuzaki [7], Munson [8] and Zakharov et al. [9]); (b) over the entire compositional range, with the data from Takeda and Mutuzaki [7].

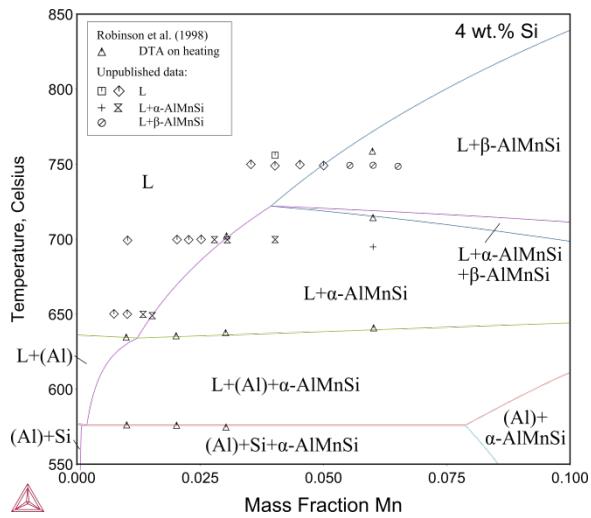


Fig.6: Al-Mn-Si vertical section at 4 wt. % Si (β -AlMnSi: τ_8 ; α -AlMnSi: τ_9)

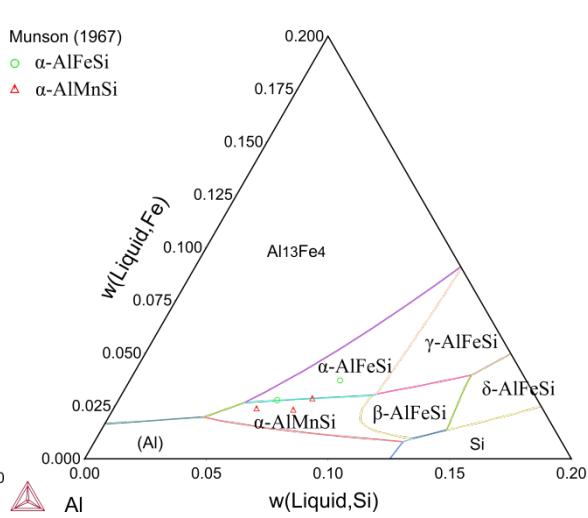


Fig.7: Al-Fe-Mn-Si liquidus surface at 0.3 wt. % Mn. (α -AlFeSi: τ_5 ; β -AlFeSi: τ_6 ; γ -AlFeSi: τ_7 ; δ -AlFeSi: τ_4 ; α -AlMnSi: τ_9)

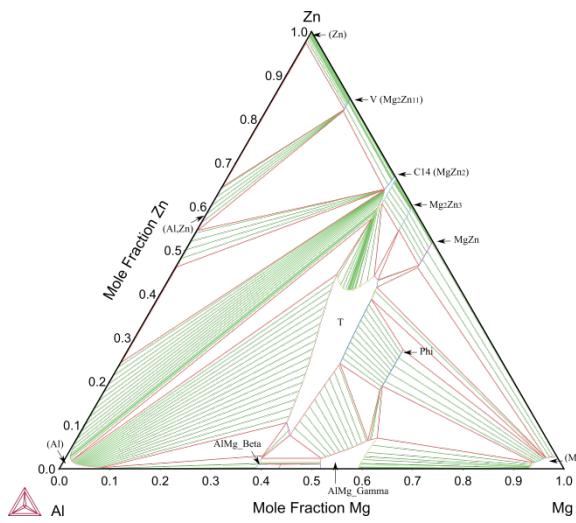


Fig.8: Isothermal section of Al-Mg-Zn^[10] at 335°C

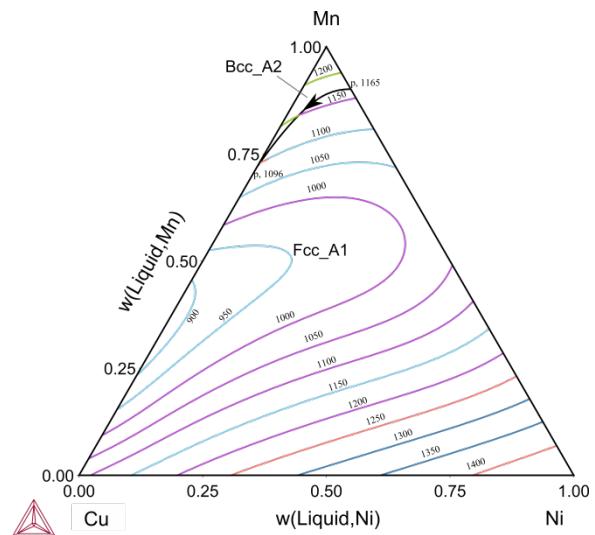


Fig.9: Liquidus projection with isothermal lines in the Cu–Mn–Ni system^[11]

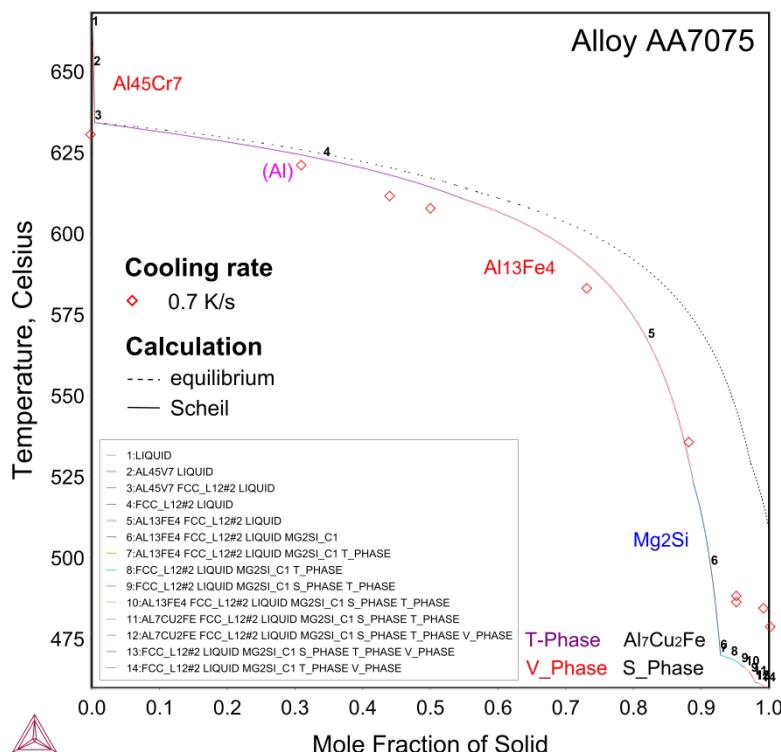


Fig.10: Equilibrium solidification and Scheil solidification simulations of alloy AA7075, compared with experimental result^[12]. (Al), Al₁₃Fe₄, Mg₂Si, T-Phase and V-Phase (MgZn₂) are found in the microstructure as predicted from the calculation. Al₄₅Cr₇ forms primarily as a Cr-bearing phase, which may have been overlooked in experimental investigation due to its small amount. S-Phase was shown at the late stage of the Scheil solidification and its amount is small. Al₂Cu was experimentally observed but not shown in the calculation.

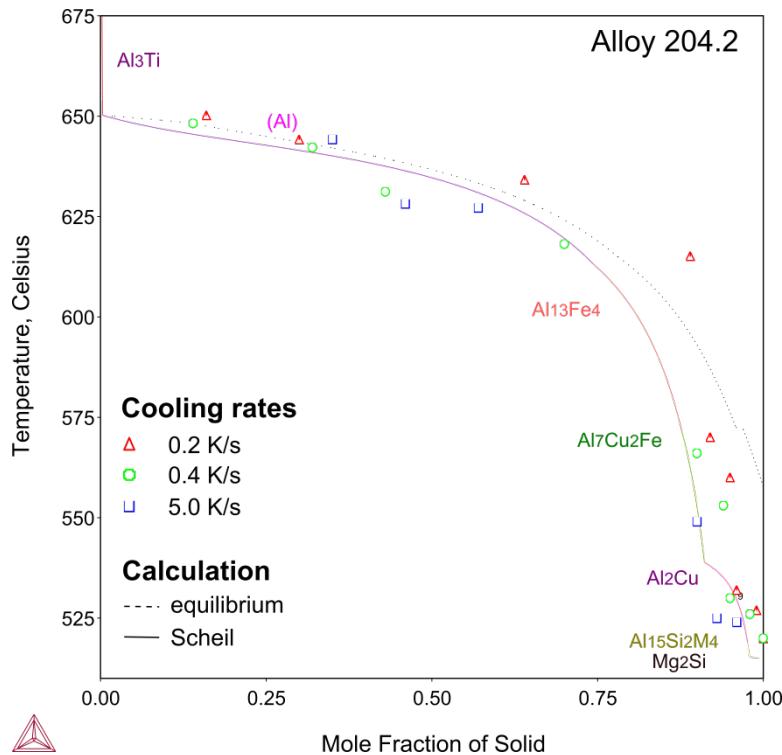


Fig.11: Equilibrium solidification and Scheil solidification simulations of alloy 204.2, compared with experimental result [12]. (Al), Al₁₃Fe₄, Al₇Cu₂Fe and Al₂Cu are found in the microstructure as predicted from the calculations. Al₃Ti appeared as the primary phase since this alloy contains 0.22 wt. % Ti.

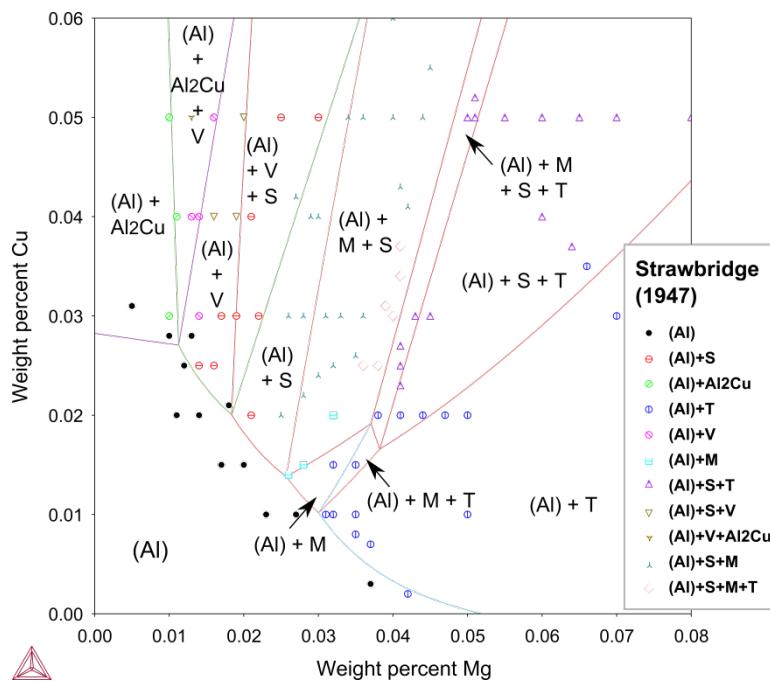


Fig.12: Calculated isothermal sections of the Al-Cu-Mg-Zn system at 460°C: at 8 wt. % Zn (experimental data are from Strawbridge et al. [13]).

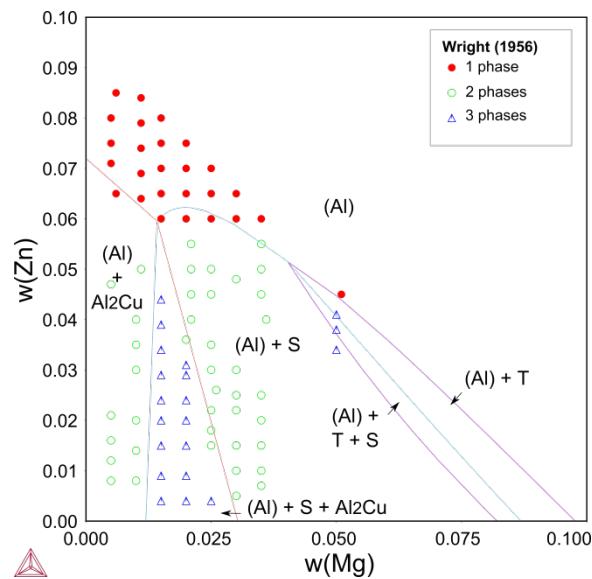


Fig.13: Calculated isothermal sections of the Al-Cu-Mg-Zn system at 90 wt.% Al (experimental data are from Wright [14]).

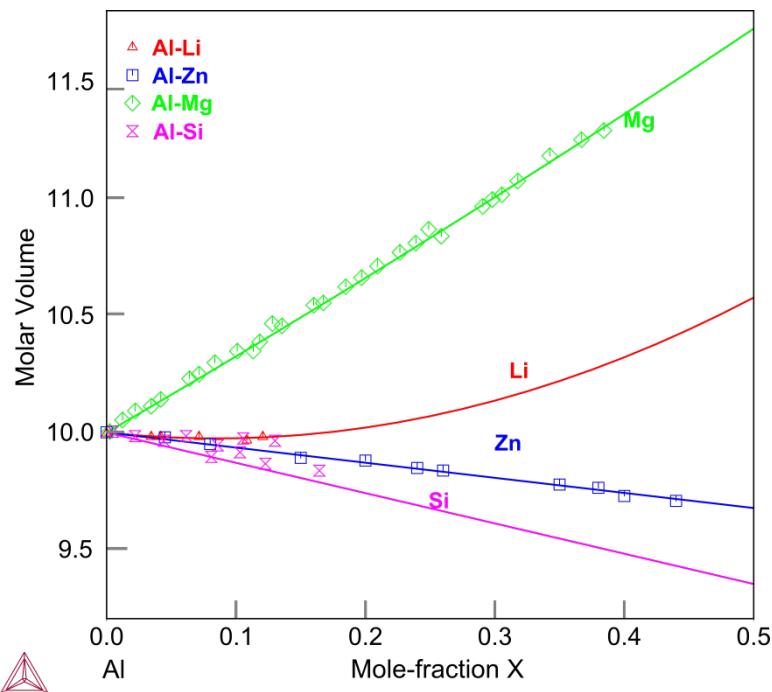


Fig.14: Calculated molar volumes of the Al-X ($X=\text{Li}, \text{Mg}, \text{Si}, \text{Zn}$) fcc_A1 phase

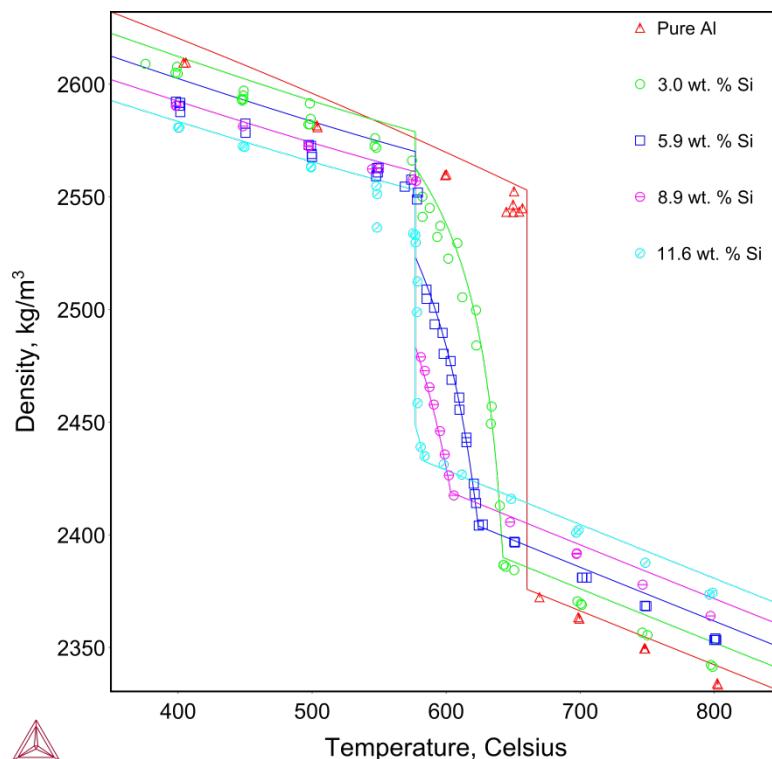


Fig.15: Calculated densities of pure Al and Al-Si alloys versus the temperature, in comparison with experimental data from Magnusson and Arnberg^[15]

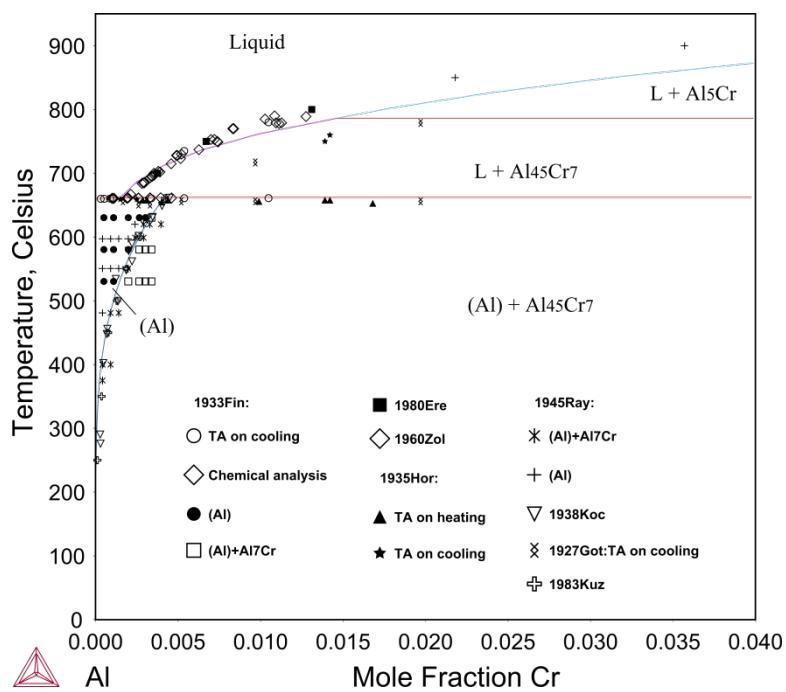


Fig.16: Calculated Al-rich Al-Cr binary phase diagram. Experimental data are from refs. [16- 18]

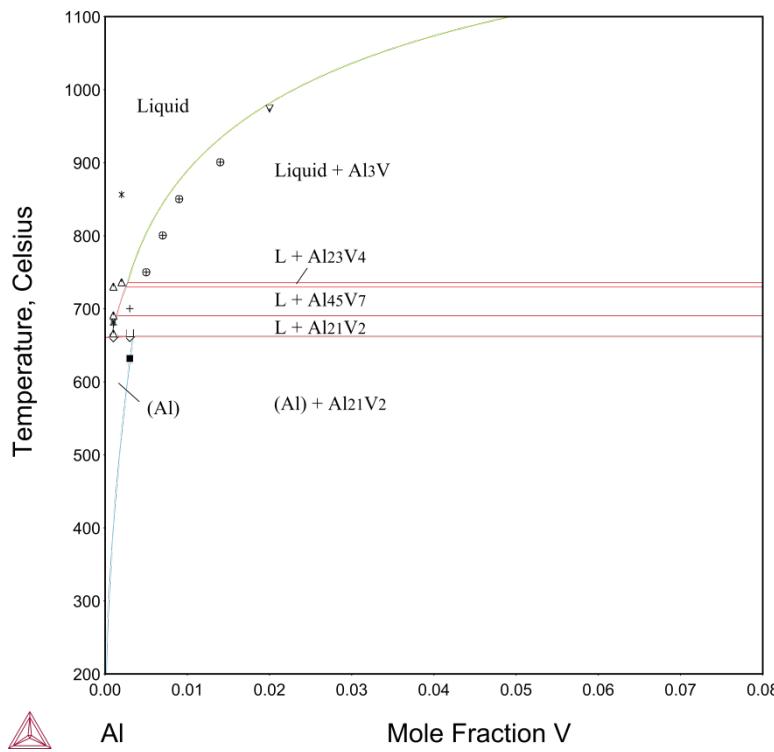


Fig.17: Calculated Al-rich Al-V binary phase diagram. Experimental data were read from ref. [19]

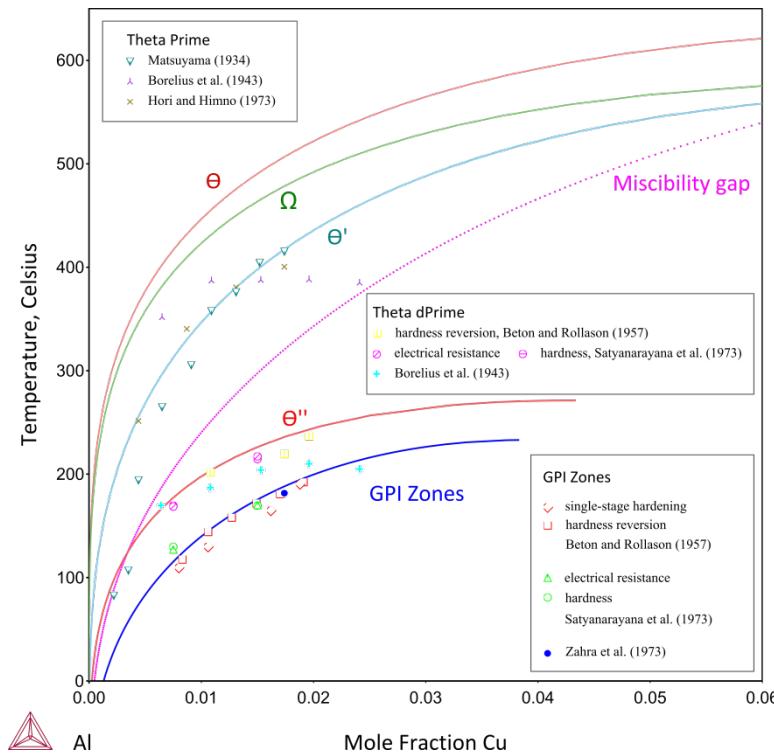


Fig.18: Calculated Al-Cu (Al) solvus curves equilibrated with θ , Ω , θ' , θ'' (or GPII zones) and GPI zones, respectively. A metastable miscibility gap of fcc_A1 is shown. The GPI zones were modeled as the second composition set of fcc_A1, i.e. fcc_A1#2. It is assumed that experimentally observed GPI zones are usually tiny (say < 3 nm), so the interfacial and elastic energy has to be considered. The line for GPI zones was calculated with adding +800 J/mole-atoms to the energy of fcc_A1#2.

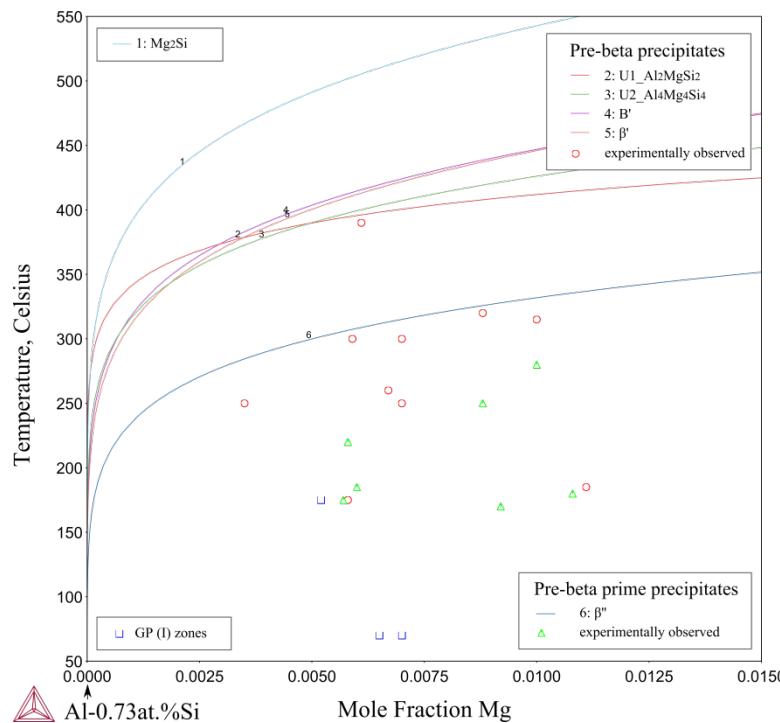


Fig.19: Calculated fcc_A1 solvi in alloys at 0.73 at.% Si and varying Mg content, relative to different Al-Mg-Si precipitates, including the stable β -Mg₂Si phase, pre- β precipitates (β' , U1, U2 and B')and the pre- β' precipitate, i.e. β'' . The symbols indicate certain precipitates have been experimentally observed in alloys of given compositions and aged at given temperatures.

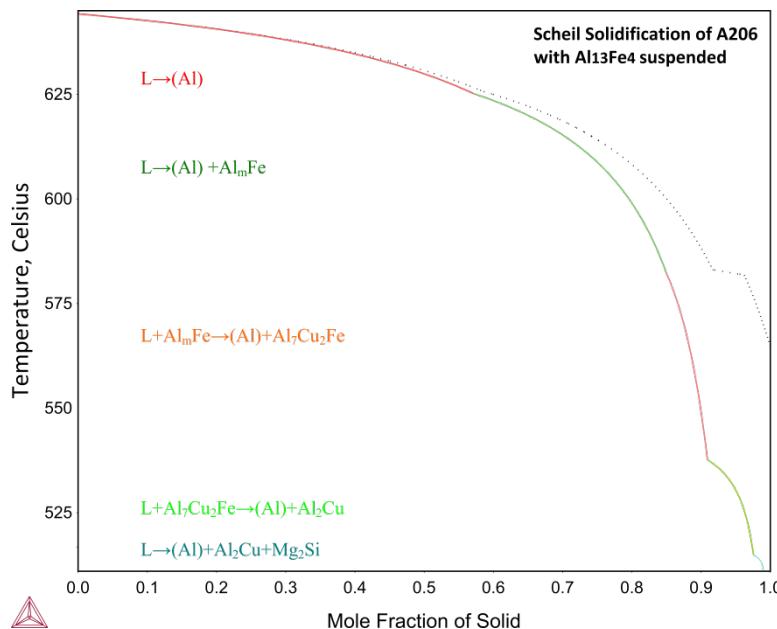


Fig.20: Scheil solidification of an A206 alloy (Al-4.58Cu-0.28Mg-0.51Fe-0.07Si-0.003Mn, wt.%) with the Al₁₃Fe₄ phase suspended. According Liu et al. [20], the metastable Al_mFe phase formed after (Al) during the solidification. The phase formation sequence and phase transformation temperatures can be well accounted for with this calculation.

References

1. N. Dupin, Thesis, LTPCM, France, 1995.
I. Ansara, N. Dupin, H.L. Lukas, B. Sundman, *J. Alloys Compd.*, 247 (1997) 20-30.
2. S. anMey, *Z. Metallkd.*, 84 (1993) 451-455.
3. Y. Du, et al., Reassessment of the Al-Mn system and a thermodynamic description of the Al-Mg-Mn system, *Int. J. Mater. Res.* 98 (9) (2007) 855-871.
4. N. Saunders, Calculated stable and metastable phase equilibria in Al-Li-Zr alloys, *Z. Metallkd.* 80 (1989) 894-903. Modified by H.-L. Chen in order to implement the new compound AlLi₂.
5. S. Pontevichi, F. Bosselet, F. Barbeau, M. Peronnet, J.C. Viala, Solid-liquid phase equilibria in the Al-Fe-Si system at 727 °C. *J. Phase Equilb. Diffus.* 25 (2004) 528–537.
6. F. Bosselet, S. Pontevichi, M. Sacerdote-Peronnet, J.C. Viala, Experimental measurement of the Al-Fe-Si isothermal section at 1000 K. *J. Phys. IV France* 122 (2004) 41–46.
7. H.P. Takeda, K. Mutuzaki, The equilibrium diagram of the Iron-Aluminum-Silicon system, *Tetsu to hagane* 26 (1940) 335–361.
8. D. Munson, A clarification of the phases occurring in aluminum-rich Aluminum-Iron-Silicon alloys with particular reference to the ternary phase α -AlFeSi. *J. Inst. Met.* 95 (1967) 217–219.
9. A.M. Zakharov, I.T. Gu'din, A.A. Arnol'd, Y.A. Matsenko, Phase diagram of the Aluminium-Silicon-Iron system in the concentration range of 10-14% Si and 0-3% Fe. *Russ. Metall.* 3 (1988) 177–180.
10. P. Liang, *Thermochim. Acta*, 314 (1998) 87-110, Al-Mg-Zn. Updated by H.-L. Chen in 2012 during the database construction.
11. W.H. Sun, *CALPHAD*, 33 (2009) 642-649.
12. L. Bäckerud, G.C. Chai, J. Tamminen, *Solidification Characteristics of Aluminium Alloys*, Vol. 1 and 2. Seden (1990).
13. D.J. Strawbridge, W. Hume-Rothery, A.T. Little, The constitution of aluminum-copper-magnesium-zinc alloys at 460°C, *J. Inst. Metals (London)*, 74 (1947) 191-225.
14. E.H. Wright, Equilibrium relations at 460°C in aluminum-copper-magnesium-zinc alloys of high purity, Internal Report 13-56-EC2, Aluminum Research Laboratories, Aluminum Company of America.
15. T. Magnusson, L. Arnberg, Density and solidification shrinkage of hypoeutectic Aluminum-Silicon alloys, *Metall. Mater. Trans. A* 32 (2001) 2605-2613.
16. W.L. Fink, H.R. Freche, *Trans. AIME* 104 (1933) 325.
17. G. V. Raynor, K. Little, *J. Inst. Met.* 71 (1945) 481.
18. V.N. Eremenko, Y.A.V. Natanzon, V.P. Titov, *Russ. Metall.* 6 (1980) 193.
19. W.P. Gong, Y. Du, et al., Thermodynamic reassessment of the Al-V system, *Int. J. Mater. Res.* 95 (11) (2004) 978-986.
20. K. Liu, X. Cao, X. G. Chen, A new Iron-rich intermetallic-AlmFe phase in Al-4.6Cu-0.5Fe cast alloy, *MMTA* 43 (2012) 1097-1101.

List of all phases included in TCAL4

The elements in phase names are rearranged in alphabetical order except for carbides and borides, where B and C are at the end. In order to designate the high- and low-temperature modifications of a phase, respectively, the suffixes “_HT” and “_LT” are used instead of “alpha-”, “beta-” or “gamma”.

One can list phases and constituents in the Database module and the GES module. For some phases, supplementary information has been included in their definitions. In order to show the information, it is recommended to use “List-System” or “List-Database” with the option of “Constituents” in the Database module.

LIQUID	M11GE8_OP76	ALMG_GAMMA
FCC_A1	SIZR3_TP32	ALMGZN_PHI
FCC_L12	SI4ZR5_TP36	AL12MN
BCC_A2	SI2ZR_C49	AL6MN
BCC_B2	SI2TI_C54	AL4MN_R
CBCC_A12	ZRM5_C15B	AL4MN_U
CUB_A13	FEM_B35	AL11MN4_LT
DHCP	SN5TI6_OI44	AL11MN4_HT
HCP_A3	M7C3_D101	AL8MN5
BETA_RHOMBO_B	M23C6_D84	AL3NI_D011
GRAPHITE	V_PHASE	AL3NI2
RHOMBO_A7	AGLA	AL3NI5
TETRA_A6	AG2LA	CA2CU
C14_LAVES	AG51LA14	CACU
C15_LAVES	AG5LA	HCP_CA
C36_LAVES	AGMG3	CAH2_LT
DIAMOND_A4	AGMG4	CAH2_HT
BCT_A5	AG2NA	CALI2
GAS	AGZN3	CA2NI7
ORTHORHOMBIC_GA	AGZN	CANI3
SIGMA	AG5ZN8	CA3SI4
AL45V7	ALCU_DEL	CA14SI19
CUZR2_C11B	ALCU_EPS	CASI2
B2_BCC	ALCU_ETA	CA3ZN
ALB2_C32	AL2CU_C16	CA5ZN3
FEB_B27	AL2CU_OMEGA	CAZN
AL3TI_D022	THETA_PRIME	CAZN2
ALZR2_B82	THETA_DPRIME	CAZN3
CAZN13_CF112	ALCU_ZETA	CAZN11
NI3SN_D019	GAMMA_D83	CU9GA4_0
CRB_B33	GAMMA_H	CU9GA4_1
AL3ZR5_D8M	AL2FE1	CU9GA4_2
AL2SR_OI12	AL5FE2	CU9GA4_3
CACU5_D2D	AL5FE4	CUGA2
M3B4_D7B	AL13FE4	CUGA_THETA
SI2ZR3_D5A	AL4FE	CU5HF
CO2SI_C23	ALMG_BETA	CU51HF14
CR5B3_D81	ALMG_EPS	CU8HF3

CU10HF7	LA7NI16	NI7HF3
CU1HF2	LANI3	NI10HF7
CUIN_GAMMA	LA2NI7_LT	NI11HF9
CUIN_THETA	LA2NI7_HT	NIHF_LT
CU2IN_HT	LANI5	NIHF_HT
CU2IN_LT	LA3SI2	NIHF2
CU11IN9	LA5SI3	HF2SI
CU37LA3	LA5SI4	HF3SI2
CU6LA1_LT	LASI	HF5SI3
CU6LA1_HT	LASI2_A1	HFSI
CU5LA1	LASI2_A2	HF5SI4
CU4LA1	LAZN2	HFSI2
CU2LA1	LAZN4	NI2SI_LT
CU1LA1	LAZN5	NI3SI2
CUMG2	LA3ZN22	NI3SI_LT
CU4SC	LAZN11	L12_FCC
CU2SC_C11B	LAZN13	NI3SI_MT
CUSC	LA2ZN17	NI5SI2
CU33SI7_DELTA	LI2ZN3_L	NISI_B31
CU15SI4_EPSILON	LI2ZN3_H	NISI2
CU56SI11_GAMMA	LI2ZN5_L	NIZN_LT
CUSI_ETA	LI2ZN5_H	NIZN8
EPSILON	LIZN4_L	ALB12
C14_FE2HF	LIZN4_H	AL4C3
C36_FE2HF	LIZN2	AL5CR
C15_FE2HF	BCC_B32	AL4CR
FE1HF2	MG2NI	ALCR_GAMMA1
FE2SC_C15	MG2SI_C1	GAMMA_D810
FE6SC29	MG7ZN3	AL7SR8
FE2SI	MGZN	AL4M_D13
FESI2_H	MG2ZN3	AL2TI
FESI2_L	FCC_L10	AL18TI7
FESI_B20	MNNI2	ALTI3_D019
MN5SI3_D88	MNSC4	AL21V2
GAMMA_D82	MN11SI19	AL23V4
FEZN_GAMMA1	MN3SI	AL8V5
FEZN_DELTA	MN6SI	AL2ZR
FEZN_ZETA	MN9SI2	AL4ZR5
LAH3	MNZN9	AL3ZR2_OF40
MGH2	NASI_LT	AL2ZR3_TP20
HFMN	NASI_HT	AL3ZR_D023
KNA2	NAZN13	CU6SN5_LT
KZN13	NI5HF	CUSN_GAMMA
LA3NI	NI7HF2	CU10SN3
LA7NI3	NI3HF_LT	CU3SN
LANI	NI3HF_HT	CU41SN11
LA2NI3	NI21HF8	CU6SN5_LT

CUSR	GE2NI5	ZN22ZR
CU2TI	GE3NI5_LT	ZN39ZR5
CU3TI2	GE12NI19	ZN3ZR_LT
CU4TI1	NI7SC2	ZN3ZR_HT
CU4TI3	NISC_B2	B4C
CUTI_B11	NISC2	CR2B_ORTH
CUTI3	NI3SN2_LT	CRB4
CU10ZR7	NI3SNHT	SRB6
CU51ZR14	NI3SN2HT	V2B3
CU8ZR3	NI3SN4	V5B6
FE2GE1	NISR	ZRB12
FEGE_ETA	NI3TI_D024	CU6CE
FE6GE5	NITI2	CU5CE
FE2GE3	NI2V7	CU4CE
FE5SN3_D82	NI2V1	CU2CE
FE3SN2	NI10ZR7	CUCE
FEZR3	NI12ZR8	CR3C2
MGB4	NI11ZR9	V3C2
MGB7	NI7ZR2	CDCU2
MG2C3	SC5SI3	CD3CU4
MGC2	SCSI	CD8CU5
MG2SR	SC3SI5_LT	CD10CU3
MG38SR9	SC3SI5_HT	CR3GE
MG23SR6	SCZN	CR5GE3_HT
MG17SR2	SCZN2	CR5GE3_LT
MN2B_D1F	SC13ZN58	CR11GE19
MNB4	SC3ZN17	GE2SR
M5C2	SCZN12	GE3TI5
CR3MN5	SIB3	V3GE
HIGH_SIGMA	SIB6	V17GE31
GEMN3_HT	SIBX	SR3SN5
GE2MN3	SIC	SRSN3
GE2MN5	CR3SI	SRSN4
GE3MN7	CRSI2_C40	SN3TI2
MN2SN	SI2SRHT	V3SN
MN3SN2	SI2SR_LT	VSN2
MNTI_LT	SI3TI5	SN3ZR5
MNTI_HT	V3SI	SNZR3_A15
MN3TI	CRZN13	V2ZR
MN4TI	CRZN17	AL14CA13
MNV_SIGMA	SRZN5_LT	AL3CA8
NI4B3_ORTH	TIZN2	AL4CE
NI4B3_MONO	TIZN5	AL11CE3
GENI2	TIZN10	AL3CE_H
GE3NI5_HT	TIZN15	AL3CE_L
GE2NI3	V4ZN5	ALCE
GENI3_GAMMA	ZN2ZR	ALCE2

ALCE3_H
ALCE3_L
AL13CO4
AL3CO
AL5CO2
AL9CO2
AL3LA
ALLA
AL53LA22
AL11LA3_LT
AL11LA3_HT
AL2LI3
ALLI2
AL4LI9
ALLI_B32
AL3SC
AL2SC
ALSC
ALSC2
AL3HF4
LI22SI5
LI13SI4
LI7SI3
LI12SI7
AG9CA2
AG7CA2
AG2CA
AGCA3
AG4SC
AGSC
AL4C4SI
AL8C7SI
AL13CR4SI4
AL9CR3SI
ALCRSI_T3
ALCRSI_T4
AL23CUFE4
AL62CU25FE13
AL7CU2FE
AL10CU10FE
AL2CULI
ALCULI_T2
ALCULI_R
ALCULI_B

Q_PHASE
S_PHASE
S_PRIME
T_PHASE
AL28CU4MN7
AL11CU5MN3
ALCU3MN2
AL7CU4NI
ALCUSC_TAU
AL9FENI
AL10FE3NI
AL71FE5NI24
ALFESI_T1
ALFESI_T2
ALFESI_T3
ALFESI_T4
AL8FE2SI
AL9FE2SI2
ALFESI_T7
ALFESI_T8
ALFESI_T10
ALFESI_T11
ALFEZN_GAMMA
ALLIMG_T
ALLISI
AL3LI8SI5
ALLI5SI2
AL18MG3MN2
ALMG3NI2
BETA_PRIME
B_PRIME
U1_AL2MGSI2
U2_AL4MG4SI4
BETA_DPRIME
ETA_PRIME
T_PRIME
AL31MN6NI2
AL2MN2SI3
AL5MN6SI7
AL1MN1SI1
AL3MNSI2
AL3MN4SI2
ALMNSI_T6
ALMNSI_T8

AL15SI2M4
AL2MNSI3
AL24MN5ZN
AL9MN2ZN
AL11MN3ZN2
ALNI2SI
AL6NI3SI
ALNI16SI9
ALNI2ZN
AL13NI38ZN49
CU16MG6SI7
CU3MG2SI
CU5MN4SI
CUMNZN
CU6NISI3
CU46NI25SI29
FE5NI3SI2
ZN13M2
MG6MN3NI
MGNI6SI6
MG2NI3SI
MG10NI55SI35
MG2NI16SI11
MG5NI9SI
MG9NI29SI16
MGNI6ZN6
MN15NI45SI40
MN15NI50SI35
MN6NI16SI7
MN1NI1SI1
MNNISI_T5
MNNISI_T6
MN3NI2SI
MN2NISI
MN6NISI3
MN66NI4SI30
MN52NI29SI19
MN7NI7ZN86
NI2SIZN_T1
NI9SI2ZN_T2
NI2SIZN3_T3
NISIZN_T4
Q_ALCUMIGSI
AL18FE2MG7SI10

List of models for all the phases included in TCAL4

LIQUID	:AG AL CU FE K LI MG MN NA NI SI ZN GE CR SN SR TI V ZR HF LA SC CO PB BE BI CA CE CD GA IN: B C H VA: > Metallic LIQUID solution phase	:AG AL CU FE K LI MG MN NA NI SI ZN GE CR SN SR TI V ZR HF LA SC CO PB BE BI CA CE CD GA IN: B C H VA: > Metallic HCP_A3 solution, alpha_Mg/Hf/Sc/Ti/Zr, epsilon_CuZn, etc.
FCC_A1 2 SUBL 1 1	:AL CU FE MG MN NI SI ZN CR GE SN SR TI V ZR AG CA HF K LA LI NA SC CO PB BE BI CE CD GA IN: B C H VA: > Metallic FCC_A1 solution, e.g. (Al), (Cu), and MC carbides	BETA_RHOMBO_B 2 SUBL 93 12 :B: CU SI B C:
FCC_L12 3 SUBL 0.75 0.25 1	:AL CU FE MG MN NI SI ZN CR GE SN SR TI V ZR AG CA HF K LA LI NA SC CO PB BE BI CE CD GA IN: AL CU FE MG MN NI SI ZN CR GE SN SR TI V ZR AG CA HF K LA LI NA SC CO PB BE BI CE CD GA IN: B C H VA: > Solution of ordered FCC_L12, having Gibbs energy contribution from FCC_A1	GRAPHITE :B C:
BCC_A2 2 SUBL 1 3	:AL CU FE MG MN NI SI ZN VA GE CR SN SR TI V ZR AG CA HF K LA LI NA SC CO PB BE BI CE CD GA IN: B C H VA: > Metallic BCC_A2 solution	RHOMBO_A7 :BI IN:
BCC_B2 3 SUBL 0.5 0.5 3	:AL CU FE MG MN NI SI TI ZN VA GE CR SN SR V ZR AG CA HF K LA LI NA SC CO PB BE BI CE CD GA IN: AL CU FE MG MN NI SI TI ZN VA GE CR SN SR V ZR AG CA HF K LA LI NA SC CO PB BE BI CE CD GA IN: B C H VA: > Solution of ordered BCC_B2, having Gibbs energy contribution from BCC_A2	TETRA_A6 :IN:
CBCC_A12 2 SUBL 1 1	:AL CO CU FE LI MG MN NI SI ZN CR GE SN SR TI V ZR: B C H VA:	C14_LAVES 2 SUBL 2 1 :AL CU CR FE HF MG MN NI TI ZN ZR: AL CU CR FE HF MG MN SC TI ZN ZR: > Solution of MgZn2-type phases, including MgZn2 (Eta, aka M or sigma)
CUB_A13 2 SUBL 1 1	:AL CU FE HF LI MG MN NI SI ZN CR GE SN SR TI V ZR AG: B C H VA:	C15_LAVES 2 SUBL 2 1 :AL CA CU FE MG NI SC SI ZN CR TI ZR LA HF: AL CA CE CU FE MG NI SC SI ZN CR TI ZR LA HF: > Solution of Cu2Mg-type phases, cf24, Fd-3m
DHCP	:CU LA NI SC:	C36_LAVES 2 SUBL 2 1 :AL CU CR FE HF MG MN NI ZN ZR: AL CU CR FE HF MG NI SC ZN ZR: > Solution of MgNi2-type phases, hP24, P63/mmc
		DIAMOND_A4 :AL SI ZN B C GA GE SR SN TI: > Pure C, Ge, Si or solution phases based on them
		BCT_A5 :AL B CU GA GE IN ZN SN TI: > Pure Sn or its solution

HCP_A3 2 SUBL 1 0.5

GAS	
:AL AL2 B B2 C C2 C3 C4 C5 C60 CA CA2 CO CO2 CR CR2 CU CU2 FE FE2 GE GE2 LI LI2 MG MG2 MN NI NI2 SC SI SI2 SI3 SN SN2 SR SR2 TI TI2 V ZN K K2 NA NA2 H H2 AL1H1 AL1H2 AL1H3 CA1H1 CU1H1 H1LI1 H1NA1 K1LI1:	
ORTHORHOMBIC_GA	
:GA:	
SIGMA 3 SUBL 8 4 18	
:FE MN NI: CR V: FE CR MN NI V:	
AL45V7 2 SUBL 45 7	
:AL: CR V: > Al45Cr7, Al45V7	
CUZR2_C11B 2 SUBL 1 2	
:AL CR CU SI ZN: AL CR TI ZR: > AlCr2, CuTi2, CuZr2, Ti2Zn, ZnZr2	
B2_BCC 2 SUBL 1 1	
:AL CO FE CU ZN: CO MN TI VA ZR: > CuZr, FeTi, TiZn, ZnZr, MnZn(rt), AlCo	
ALB2_C32 2 SUBL 1 2	
:MG AL MN CR ZR TI V: B: > AlB2, B2Cr, B2Mg, B2Mn, B2Ti, B2V, B2Zr	
FEB_B27 2 SUBL 1 1	
:FE MN TI ZR SR: B SI ZN GE: > BFe, BMn, BTi, GeZr, SiTi, SrZn, SiZr	
AL3TI_D022 2 SUBL 3 1	
:AL MN NI: GE MN TI V ZR: > Al3Ti, Ni3V, GeMn3, Al3V	
ALZR2_B82 2 SUBL 1 2	
:AL GE MN SN VA: ZR MN TI VA: > SnTi2, GeMn2, AlZr2	
CAZN13_CF112 2 SUBL 1 13	
:CA SR: ZN: > CaZn13, SrZn13	
NI3SN_D019 2 SUBL 1 3	
:AL SN GE VA: LA MN FE NI TI: > Ni3Sn, SnTi3, SnMn3, AlLa3	
CRB_B33 2 SUBL 1 1	
:AL CA CR NI SR V: AG B SI GE SN ZR HF: > AgCa, AlHf, AlZr, BNi, BV, GeSr, NiZr, SiSr, SnSr	
AL3ZR5_D8M 2 SUBL 3 5	
:AL SI GE: ZR CR V: > Al3Zr5, Cr3Si5, Ge3V5, Si3V5	
AL2SR_OI12 2 SUBL 2 1	
:AL ZN: SR:	

> Al2Sr, SrZn2	
CACU5_D2D 2 SUBL 1 5	
:CA SC SR: CU NI ZN: > CaCu5, CaNi5, CaZn5, Cu5Sr, ScNi5, SrZn5	
M3B4_D7B 2 SUBL 4 3	
:B: MN CR TI V: > V3B4, Ti3B4, Mn3B4, Cr3B4	
SI2ZR3_D5A 2 SUBL 2 3	
:SI B: ZR V: > Si2Zr3, B2V3	
CO2SI_C23 2 SUBL 2 1	
:CA CU FE NI SR: AL SI ZN GE SN: > Ca2Si, Ni2Si, SiSr2, SnSr2, GeSr2	
CR5B3_D81 2 SUBL 0.625 0.375	
:CA CR SR: AG SI B GE SN: > Ca5Si3, Sn3Sr5, Si4Sr5, Ge3Sr5, B3Cr5	
M11GE8_OP76 2 SUBL 0.579 0.421	
:CR V: GE: > Cr11Ge8, V11Ge8	
SIZR3_TP32 2 SUBL 1 3	
:SI GE: ZR TI: > SiZr3, SiTi3, GeZr3	
SI4ZR5_TP36 2 SUBL 4 5	
:SI GE: TI ZR: > Si4Zr5, Si4Ti5, Ge4Zr5	
SI2ZR_C49 2 SUBL 2 1	
:SI GE: ZR: > Si2Zr, Ge2Zr	
SI2TI_C54 2 SUBL 2 1	
:GE SI SN: TI ZR: > Ge2Ti, Si2Ti, Sn2Zr	
ZRM5_C15B 2 SUBL 5 1	
:CU NI: ZR: > Cu5Zr, Ni5Zr	
FEM_B35 2 SUBL 1 1	
:FE: SN GE: > FeSn, FeGe	
SN5TI6_OI44 2 SUBL 5 6	
:GE SN SI: V TI: > Sn5Ti6, Si5V6, Ge5Ti6	
M7C3_D101 2 SUBL 7 3	
:MN CR: C: > Cr7C3, Mn7C3	
M23C6_D84 2 SUBL 23 6	
:MN CR: C SC: > Cr23C6, Mn23C6, Mn23SC6	

V_PHASE	3 SUBL 5 6 2
:AL SI ZN: CU ZN: MG:	> solution of Mg ₂ Zn ₁₁ , Al ₅ Cu ₆ Mg ₂ ; aka Z
AGLA	2 SUBL 1 1
:LA: AG:	
AG2LA	2 SUBL 1 2
:LA: AG:	
AG51LA14	2 SUBL 14 51
:LA: AG:	
AG5LA	2 SUBL 1 5
:LA: AG:	
AGMG3	2 SUBL 0.23 0.77
:AG: MG:	> Ag ₁₇ Mg ₅₄
AGMG4	2 SUBL 0.2 0.8
:AG: MG:	> Ag ₉ Mg ₃₇
AG2NA	2 SUBL 2 1
:AG: NA:	
AGZN3	
:AG ZN:	
AGZN	2 SUBL 1 2
:ZN: AG ZN:	
AG5ZN8	4 SUBL 2 2 3 6
:AG ZN: AG: AG ZN: AG ZN:	
ALCU_DEL	2 SUBL 2 3
:AL ZN: CU FE:	
ALCU_EPS	2 SUBL 1 1
:AL CU ZN NI: CU FE:	
ALCU_ETA	2 SUBL 1 1
:AL CU: CU FE ZN NI:	
AL2CU_C16	2 SUBL 2 1
:AL FE GE SN ZR MN NI HF: AL CU FE NI B MN SI:	> Al ₂ Cu, AlHf ₂ , Fe ₂ B, FeGe ₂ , FeZr ₂ , FeSn ₂ , Mn ₂ B, MnSn ₂ , NiB ₂ , NiZr ₂ , SiZr ₂
AL2CU_OMEGA	2 SUBL 2 1
:AL: CU:	> Al ₂ Cu-OMEGA metastable precipitate
THETA_PRIME	2 SUBL 2 1
:AL: CU:	
THETA_DPRIME	2 SUBL 3 1
:AL: CU:	
ALCU_ZETA	2 SUBL 9 11
:AL: CU FE:	
GAMMA_D83	3 SUBL 4 1 8
:AL FE NI SI ZN: AL CU NI SI ZN: CU MN FE NI ZN:	

> solution between Al ₈ Cu ₅ (rt) and Cu ₅ Zn ₈	
GAMMA_H	3 SUBL 4 1 8
:AL ZN: AL CU ZN: CU MN FE NI:	> Cu ₅ Zn ₈ -type Al ₈ Cu ₅ (ht) phase
AL2FE1	2 SUBL 2 1
:AL CU SI ZN: FE MN NI:	
AL5FE2	2 SUBL 5 2
:AL CU SI ZN: FE MN NI:	
AL5FE4	
:AL CU FE MN:	
AL13FE4	3 SUBL 0.6275 0.235 0.1375
:AL CU: FE MN NI ZN: AL SI VA ZN:	> solution phases based on Al ₁₃ Fe ₄ , aka Al ₃ Fe
AL4FE	2 SUBL 4.2 1
:AL: FE:	
ALMG_BETA	2 SUBL 89 140
:MG LI: AL ZN:	
ALMG_EPS	2 SUBL 23 30
:MG: AL ZN:	
ALMG_GAMMA	3 SUBL 5 12 12
:LI MG: AL MG ZN: AL MG ZN:	
ALMGZN_PHI	2 SUBL 6 5
:MG: AL ZN:	> a Al-Mg-Zn ternary phase know as PHI
AL12MN	2 SUBL 12 1
:AL: MN:	
AL6MN	2 SUBL 6 1
:AL ZN: MN FE:	
AL4MN_R	2 SUBL 0.81162 0.18838
:AL: MN FE:	> AL461MN107
AL4MN_U	2 SUBL 4 1
:AL ZN: MN:	
AL11MN4_LT	2 SUBL 11 4
:AL ZN: MN FE:	
AL11MN4_HT	2 SUBL 29 10
:AL MN: MN:	
AL8MN5	3 SUBL 12 5 9
:AL ZN: MN: AL MN SI CU:	
AL3NI_D011	2 SUBL 0.75 0.25
:AL MN NI: FE NI B C:	
AL3NI2	3 SUBL 3 2 1
:AL SI ZN: AL CU FE MG NI:NI VA:	
AL3NI5	2 SUBL 0.375 0.625

:AL: NI:	
CA2CU	2 SUBL 2 1
:CA: CU:	
CACU	2 SUBL 1 1
:CA: CU:	
HCP_CA	2 SUBL 1 0.5
:CA: H VA:	
CAH2_LT	2 SUBL 1 2
:CA: H:	
CAH2_HT	2 SUBL 1 2
:CA: H:	
CALI2	2 SUBL 2 1
:LI: CA:	
CA2NI7	2 SUBL 2 7
:CA: NI:	
CANI3	2 SUBL 0.25 0.75
:CA: NI:	
CA3SI4	2 SUBL 0.428571 0.571429
:CA: SI:	
CA14SI19	2 SUBL 0.424242 0.575758
:CA: SI:	
CASI2	2 SUBL 0.333333 0.666667
:CA: SI:	
CA3ZN	2 SUBL 3 1
:CA: ZN:	
CA5ZN3	2 SUBL 5 3
:CA: ZN:	
CAZN	2 SUBL 1 1
:CA: ZN:	
CAZN2	2 SUBL 1 2
:CA: ZN:	
CAZN3	2 SUBL 1 3
:CA: ZN:	
CAZN11	2 SUBL 1 11
:CA: ZN:	
CU9GA4_0	3 SUBL 6 6 1
:CU: CU GA: GA:	
CU9GA4_1	4 SUBL 6 3 3 1
:CU: CU GA: CU GA: GA:	
CU9GA4_2	4 SUBL 3 3 3 4
:CU: CU VA: CU GA: GA:	
CU9GA4_3	3 SUBL 6 3 4
:CU VA: CU GA: GA:	
CUGA2	2 SUBL 1 2
:CU: GA:	
CUGA_THETA	2 SUBL 0.778 0.222
:CU: GA:	

CU5HF	2 SUBL 5 1
:CU: HF:	
CU51HF14	2 SUBL 51 14
:CU: HF:	
CU8HF3	2 SUBL 8 3
:CU: HF:	
CU10HF7	2 SUBL 10 7
:CU: HF:	
CU1HF2	2 SUBL 1 2
:CU: HF:	
CUIN_GAM	4 SUBL 2 2 3 6
:CU: CU IN: CU: CU IN:	
CUIN_THETE	2 SUBL 0.7
0.3	
:CU: IN:	
CU2IN_HT	3 SUBL 0.545 0.122 0.333
:CU: CU IN: IN:	
CU2IN_LT	2 SUBL 0.64 0.36
:CU: IN:	
CU11IN9	2 SUBL 0.55 0.45
:CU: IN:	
CU37LA3	2 SUBL 37 3
:CU: LA:	
CU6LA1_LT	2 SUBL 6 1
:CU: LA:	
CU6LA1_HT	2 SUBL 6 1
:CU: LA:	
CU5LA1	2 SUBL 5 1
:CU: LA:	
CU4LA1	2 SUBL 4 1
:CU: LA:	
CU2LA1	2 SUBL 2 1
:CU: LA:	
CU1LA1	2 SUBL 1 1
:CU: LA:	
CUMG2	2 SUBL 1 2
:CU NI: MG:	
CU4SC	2 SUBL 4 1
:CU: SC:	
CU2SC_C11B	2 SUBL 2 1
:AG CU: SC:	
CUSC	2 SUBL 1 1
:CU: SC:	
CU33SI7_DELTA	2 SUBL 0.825 0.175
:CU ZN: SI:	
CU15SI4_EPSILON	2 SUBL 0.789474
0.210526	
:CU MG MN ZN: AL SI:	

CU56SI11_GAMMA	2 SUBL 0.835821
0.164179	
:CU MG MN NI SI ZN: SI:	
CUSI_ETA	2 SUBL 0.76 0.24
:CU MN NI ZN: SI:	
EPSILON	
:CU MN ZN:	
C14_FE2HF	2 SUBL 0.6667
0.3333	
:FE: HF FE:	
C36_FE2HF	2 SUBL 0.6667
0.3333	
:FE: HF:	
C15_FE2HF	2 SUBL 0.6667
0.3333	
:FE: HF:	
FE1HF2	2 SUBL 0.3333
0.6667	
:FE: HF:	
FE2SC_C15	2 SUBL 0.64 0.36
:FE: SC:	
FE6SC29	2 SUBL 0.17 0.83
:FE: SC:	
FE2SI	2 SUBL 2 1
:FE NI: AL SI:	
FESI2_H	2 SUBL 3 7
:FE NI: AL MG SI:	
FESI2_L	2 SUBL 1 2
:FE NI: AL SI:	
FESI_B20	2 SUBL 1 1
:FE MN NI CR: AL MG SI GE:	
> FeSi, MnSi, CrSi, CrGe	
MN5Si3_D88	2 SUBL 5 3
:CU FE MN NI CR ZR TI: AL CR SI GE SN:	
> Mn5Si3, Cr3Si5, Fe5Si3, Ge3Mn5,	
Ge3Zr5, Si3Zr5, Sn3Ti5	
GAMMA_D82	4 SUBL 2 2 3 6
:FE MN ZN: FE MN NI ZN: AL CU FE MN NI SI	
ZN: AL ZN:	
FEZN_GAMMA1	
3 SUBL 0.137 0.118 0.745	
:FE: AL CU FE NI SI ZN: MN ZN:	
FEZN_DELTA	
4 SUBL 5.8E-2 0.18 0.525 0.237	
:FE: AL CU FE MN NI SI ZN: ZN: ZN:	
FEZN_ZETA	3 SUBL 7.2E-2 0.856 7.2E-2
:FE MN NI VA: AL ZN: AL CU SI VA ZN:	
LAH3	3 SUBL 0.25 0.5 0.25

:LA: H	VA: H	VA:
MGH2	2 SUBL	1 2
:MG: H:		
HFMN	2 SUBL	0.5 0.5
:HF: MN:		
KNA2	2 SUBL	1 2
:K: NA:		
KZN13	2 SUBL	1 13
:K: ZN:		
LA3NI	2 SUBL	3 1
:LA: NI:		
LA7NI3	2 SUBL	7 3
:LA: NI:		
LANI	2 SUBL	1 1
:LA: NI:		
LA2NI3	2 SUBL	2 3
:LA: NI:		
LA7NI16	2 SUBL	7 16
:LA: NI:		
LANI3	2 SUBL	1 3
:LA: NI:		
LA2NI7_LT	2 SUBL	2 7
:LA: NI:		
LA2NI7_HT	2 SUBL	2 7
:LA: NI:		
LANI5	2 SUBL	1 5
:LA: NI:		
LA3SI2	2 SUBL	0.6 0.4
:LA: SI:		
LA5SI3	2 SUBL	0.625 0.375
:LA: SI:		
LA5SI4	2 SUBL	0.5556 0.4444
:LA: SI:		
LASI	2 SUBL	0.5 0.5
:LA: SI:		
LASI2_A1	2 SUBL	0.36 0.64
:LA: SI:		
LASI2_A2	2 SUBL	0.3333 0.6667
:LA: SI:		
LAZN2	2 SUBL	0.333 0.667
:LA: ZN:		
LAZN4	2 SUBL	0.2 0.8
:LA: ZN:		
LAZN5	2 SUBL	0.1667 0.8333
:LA: ZN:		
LA3ZN22	2 SUBL	0.12 0.88
:LA: ZN:		
LAZN112	SUBL	8.3E-2 0.917

:LA: ZN:	
LAZN132 SUBL 7.1E-2 0.929	
:LA: ZN:	
LA2ZN17 2 SUBL 0.105 0.895	
:LA: ZN:	
LI2ZN3_L 2 SUBL 2 3	
:LI: LI ZN:	
LI2ZN3_H 2 SUBL 2 3	
:LI ZN: LI ZN:	
LI2ZN5_L 2 SUBL 2 5	
:LI ZN: ZN:	
LI2ZN5_H 2 SUBL 2 5	
:LI ZN: ZN:	
LIZN4_L 2 SUBL 1 4	
:LI ZN: LI ZN:	
LIZN4_H 2 SUBL 0.2 0.8	
:LI ZN: LI ZN:	
LIZN2 2 SUBL 1 2	
:LI: ZN:	
BCC_B32 2 SUBL 1 1	
:LI ZN: LI ZN:	
MG2NI 2 SUBL 2 1	
:MG ZN: CU NI ZN:	
MG2SI_C1 2 SUBL 2 1	
:MG: GE SI SN: > solution phase of Mg2Si, GeMg2, Mg2Sn	
MG7ZN3 2 SUBL 51 20	
:MG: ZN:	
MGZN 2 SUBL 12 13	
:MG: AL CU ZN:	
MG2ZN3 2 SUBL 2 3	
:MG: AL CU ZN:	
FCC_L10 2 SUBL 0.5 0.5	
:AL CU MN NI TI: AL CU MN NI TI:	
MNNI2 2 SUBL 1 2	
:MN NI: NI:	
MNSC4 2 SUBL 0.2 0.8	
:MN:SC:	
MN11SI19 2 SUBL 11 19	
:MN: AL SI:	
MN3SI 2 SUBL 3 1	
:FE MN NI: AL SI:	
MN6SI 2 SUBL 17 3	
:AL MN: SI ZN:	
MN9SI2 2 SUBL 33 7	
:MN: SI:	
MNZN92 SUBL 0.1 0.9	
:MN: ZN:	

NASI_LT	2 SUBL 1 1
:NA: SI:	
NASI_HT	2 SUBL 1 1
:NA: SI:	
NAZN13 2 SUBL 1 13	
:NA: ZN:	
NI5HF 2 SUBL 0.833 0.167	
:NI: HF:	
NI7HF2 2 SUBL 0.778 0.222	
:NI: HF NI:	
NI3HF_LT 2 SUBL 0.75 0.25	
:NI: HF:	
NI3HF_HT 2 SUBL 0.75 0.25	
:NI: HF:	
NI21HF8 2 SUBL 0.724 0.276	
:NI: HF:	
NI7HF3 2 SUBL 0.7 0.3	
:NI: HF:	
NI10HF7 2 SUBL 0.588 0.412	
:NI: HF:	
NI11HF9 2 SUBL 0.55 0.45	
:NI: HF:	
NIHF_LT 2 SUBL 0.5 0.5	
:NI: HF:	
NIHF_HT 2 SUBL 0.5 0.5	
:NI: HF:	
NIHF2 2 SUBL 1 2	
:NI VA: HF:	
HF2SI 2 SUBL 0.6666667 0.3333333 :HF: SI:	
HF3SI2 2 SUBL 0.6 0.4 :HF: SI:	
HF5SI3 2 SUBL 0.625 0.375 :HF: SI:	
HFSI 2 SUBL 0.5 0.5 :HF: SI:	
HF5SI4 2 SUBL 0.555556 0.444444 :HF: SI:	
HFSI2 2 SUBL 0.333333 0.666667 :HF: SI:	
NI2SI_HT 3 SUBL 1 1 1 :CU NI: NI VA: AL SI:	
NI3SI2 2 SUBL 3 2 :FE NI: SI:	
NI3SI_HT 2 SUBL 3 1 :FE NI: AL SI:	

L12_FCC	2 SUBL 1 3
:AL GE NI SI TI V: AL FE ZR NI ZN: > L12 phase, Ni3Si_rt, AlZr3, GeNi3, TiZn3, VZn3	
NI3SI_MT 2 SUBL 1 3 :SI: NI:	
NI5SI2	2 SUBL 5 2
:CU FE NI: AL SI:	

NISI_B31	2 SUBL 1 1
:FE NI: GE SI ZN:	
> GeNi, NiSi	
NISI2	2 SUBL 2 1
:AL CU SI ZN: CU FE MN NI:	
NIZN_LT	2 SUBL 0.5 0.5
:AL FE MN NI SI ZN: AL FE MG MN NI SI ZN:	
NIZN8	2 SUBL 0.1111111
0.8888889	
:NI: AL MN ZN:	
ALB12	2 SUBL 1 12
:AL: B:	
AL4C3	2 SUBL 4 3
:AL SI: C:	
AL5CR	2 SUBL 5 1
:AL SI: CR:	
AL4CR	2 SUBL 1 4
:CR: AL SI VA:	
ALCR_GAMMA1	4 SUBL 2 2 3 6
:AL CR SI: CR: AL CR: AL SI:	
GAMMA_D810	3 SUBL 12 5 9
:AL SI: CR: AL CR SI:	
AL7SR8	2 SUBL 7 8
:AL: SR:	
AL4M_D13	2 SUBL 4 1
:AL: CA SR:	
> Al4Ca, Al4Sr	
AL2TI	2 SUBL 2 1
:AL: TI:	
AL18TI7	2 SUBL 0.72 0.28
:AL: TI:	
ALTI3_DO19	2 SUBL 3 1
:AL TI: AL TI:	
AL21V2	2 SUBL 21 2
:AL: V:	
AL23V4	2 SUBL 23 4
:AL: V:	
AL8V5	4 SUBL 2 2 3 6
:AL V: V: AL V: AL:	
AL2ZR	2 SUBL 2 1
:AL: ZR:	
AL4ZR5	2 SUBL 4 5
:AL: ZR:	
AL3ZR2_OF40	2 SUBL 3 2
:AL: HF ZR:	
> Al3Zr2, Al3Hf2	
AL2ZR3_TP20	2 SUBL 2 3
:AL ZN: HF ZR:	

> Al2Zr3, Al2Hf3, ZN2Zr3	
AL3ZR_D023	2 SUBL 3 1
:AL: HF TI ZR:	
> Al3Zr, Al3Hf	
CU6SN5_HT	3 SUBL 1 1 1
:CU: CU SN: SN:	
CUSN_GAMMA	
:CU SN:	
CU10SN3	
:CU SN:	
CU3SN	2 SUBL 3 1
:CU SN: CU SN:	
CU41SN11	2 SUBL 41 11
:CU SN: CU SN:	
CU6SN5_LT	3 SUBL 1 1 1
:CU: CU SN: SN:	
CUSR	2 SUBL 1 1
:SR: CU:	
CU2TI	2 SUBL 2 1
:CU: TI:	
CU3TI2	2 SUBL 3 2
:CU: TI:	
CU4TI1	2 SUBL 4 1
:CU TI: CU TI:	
CU4TI3	2 SUBL 4 3
:CU: TI:	
CUTI_B11	2 SUBL 1 1
:CU TI: CU TI:	
CUTI3	2 SUBL 1 3
:CU TI: TI:	
CU10ZR7	2 SUBL 10 7
:CU: ZR:	
CU51ZR14	2 SUBL 51 14
:CU: ZR:	
CU8ZR3	2 SUBL 8 3
:CU: ZR:	
FE2GE1	3 SUBL 1 1 1
:FE: FE VA: GE:	
FEGE_ETA	2 SUBL 13 9
:FE: GE:	
FE6GE52 SUBL 6 5	
:FE: GE:	
FE2GE32 SUBL 2 3	
:FE: GE:	
FE5SN3_D82	2 SUBL 5 3
:FE: SN:	
FE3SN22 SUBL 3 2	
:FE: SN:	

FEZR3	2 SUBL 1 3	MN4TI	2 SUBL 0.815 0.185
:FE: ZR:		:MN: TI:	
MGB4	2 SUBL 1 4	MNV_SIGMA	3 SUBL 10 4 16
:MG: B:		:MN V: V: MN V:	
MGB7	2 SUBL 1 7	NI4B3_ORTH	2 SUBL 0.586 0.414
:MG: B:		:NI: B:	
MG2C3	2 SUBL 2 3	NI4B3_MONO	2 SUBL 0.564 0.436
:MG: C:		:NI: B:	
MGC2	2 SUBL 1 2	GENI2	2 SUBL 0.665 0.335
:MG: C:		:NI: GE:	
MG2SR	2 SUBL 2 1	GE3NI5_HT	2 SUBL 0.625 0.375
:MG: SR:		:GE NI: GE NI:	
MG38SR9	2 SUBL 38 9	GE2NI3_2	SUBL 0.6 0.4
:MG: SR:		:GE NI: GE:	
MG23SR6	2 SUBL 23 6	GENI3_GAMMA	2 SUBL 0.744 0.256
:MG: SR:		:NI: GE:	
MG17SR2	2 SUBL 17 2	GE2NI5	2 SUBL 0.72 0.28
:MG: SR:		:NI: GE:	
MN2B_D1F	2 SUBL 0.6707 0.3293	GE3NI5_LT	2 SUBL 0.63 0.37
:MN: B:		:NI: GE:	
MNB4	2 SUBL 0.2 0.8	GE12NI19	2 SUBL 0.613 0.387
:MN: B:		:GE NI: GE NI:	
M5C2	2 SUBL 5 2	NI7SC2	2 SUBL 0.222222 0.777778
:MN: C:		:SC: NI:	
CR3MN5	2 SUBL 3 5	NISC_B2	2 SUBL 1 1
:CR: MN:		:SC: NI:	
HIGH_SIGMA	3 SUBL 8 4 18	NISC2	2 SUBL 0.72 0.28
:MN: CR: CR MN:		:SC: NI:	
GEMN3_HT	2 SUBL 1 3	NI3SN2_LT	3 SUBL 0.2 0.4 0.4
:GE MN: MN:		:SN: NI SN: NI:	
GE2MN3	2 SUBL 2 3	NI3SN_HT	3 SUBL 0.25 0.25 0.5
:GE: MN:		:NI SN: NI SN: NI:	
GE2MN5	2 SUBL 2 5	NI3SN2_HT	3 SUBL 0.33333 0.33334
:GE MN: MN:		0.33333	
GE3MN7	2 SUBL 3 7	:NI: NI SN: SN:	
:GE: MN:		NI3SN4	3 SUBL 0.25 0.25 0.5
MN2SN2	SUBL 0.643 0.357	:NI: NI SN: SN:	
:MN: SN:		NISR	2 SUBL 0.5 0.5
> Mn(2-x)Sn		:NI: SR:	
MN3SN2	2 SUBL 3 2	NI3TI_D024	2 SUBL 0.75 0.25
:MN: SN:		:NI TI: NI TI:	
MNTI_LT	2 SUBL 1 1	NITI2	2 SUBL 1 2
:MN: TI:		:NI TI: NI TI:	
MNTI_HT	2 SUBL 0.515 0.485	NI2V7	2 SUBL 2 7
:MN: TI:		:NI: V:	
MN3TI	2 SUBL 3 1	NI2V1	2 SUBL 2 1
:MN: TI:		:NI: V:	

NI10ZR7	2 SUBL 23 17
:NI: ZR:	
NI21ZR8	2 SUBL 8 21
:ZR: NI:	
NI11ZR9	2 SUBL 11 9
:NI: ZR:	
NI7ZR2	2 SUBL 7 2
:NI: ZR:	
SC5SI3	2 SUBL 0.625 0.375
:SC: SI:	
SCSI	2 SUBL 0.5 0.5
:SC: SI:	
SC3SI5_LT	2 SUBL 0.375 0.625
:SC: SI:	
SC3SI5_HT	2 SUBL 0.375 0.625
:SC: SI:	
SCZN	2 SUBL 0.5 0.5
:SC: ZN:	
SCZN2	2 SUBL 0.3333 0.6667
:SC: ZN:	
SC13ZN58	2 SUBL 0.1831 0.8169
:SC: ZN:	
SC3ZN17	2 SUBL 0.15 0.85
:SC: ZN:	
SCZN122	SUBL 7.7E-2 0.923
:SC: ZN:	
SIB3	3 SUBL 6 2 6
:B: SI: B SI:	
SIB6	3 SUBL 210 23 48
:B: SI: B SI:	
SIBX	3 SUBL 61 1 8
:B: SI: B SI:	
SIC	2 SUBL 1 1
:C: SI:	
CR3SI	2 SUBL 3 1
:CR SI: AL CR SI:	
CRSI2_C40	2 SUBL 1 2
:CR SI V: AL CR SI:	
SI2SR_HT	2 SUBL 2 1
:SI VA: SR:	
SI2SR_LT	2 SUBL 2 1
:SI: SR:	
SI3TI5	4 SUBL 2 3 3 1
:SI TI: SI TI: TI: VA:	
V3SI	2 SUBL 0.75 0.25
:V SI: SI V:	
CRZN13	2 SUBL 1 13

:CR: ZN:	
CRZN17	2 SUBL 1 17
:CR: ZN:	
SRZN5_LT	2 SUBL 1 5
:SR: ZN:	
TIZN2	2 SUBL 1 2
:TI: ZN:	
TIZN5	2 SUBL 1 5
:TI: ZN:	
TIZN10	2 SUBL 1 10
:TI: ZN:	
TIZN15	2 SUBL 1 15
:TI: ZN:	
V4ZN5	2 SUBL 4 5
:V: ZN:	
ZN2ZR	2 SUBL 2 1
:ZN: ZR:	
ZN22ZR	2 SUBL 22 1
:ZN: ZR:	
ZN39ZR5	2 SUBL 39 5
:ZN: ZR:	
ZN3ZR_LT	2 SUBL 3 1
:ZN: ZR:	
ZN3ZR_HT	2 SUBL 3 1
:ZN: ZR:	
B4C	2 SUBL 1 1
:B11C B12: B2 C2B CB2:	
CR2B_ORTH	2 SUBL 0.666667 0.333333
:CR: B:	
CRB4	2 SUBL 0.2 0.8
:CR: B:	
SRB6	2 SUBL 1 6
:SR: B:	
V2B3	2 SUBL 0.4 0.6
:V: B:	
V5B6	2 SUBL 0.454545 0.545455
:V: B:	
ZRB12	2 SUBL 12 1
:B: ZR:	
CU6CE	2 SUBL 6 1
:CU: CE:	
CU5CE	2 SUBL 5 1
:CU: CE:	
CU4CE	2 SUBL 4 1
:CU: CE:	
CU2CE	2 SUBL 2 1
:CU: CE:	
CUCE	2 SUBL 1 1

:CU: CE:	
CR3C2	2 SUBL 3 2
:CR: C:	
V3C2	2 SUBL 3 2
:V: C:	
CDCU2	2 SUBL 1 2
:CD: CU:	
CD3CU4	2 SUBL 0.4286 0.5714
:CD: CU:	
CD8CU5	4 SUBL 2 3 2 6
:CU: CD CU: CU: CU CD:	
CD10CU3	2 SUBL 0.7692 0.2308
:CD: CU:	
CR3GE	2 SUBL 0.75 0.25
:CR GE: CR GE:	
CR5GE3_HT	2 SUBL 0.625 0.375
:CR GE: GE CR:	
CR5GE3_LT	2 SUBL 0.625 0.375
:CR GE: GE CR:	
CR11GE19	2 SUBL 0.367 0.633
:CR: GE:	
GE2SR	2 SUBL 2 1
:GE: SR:	
GE3TI5	2 SUBL 5 3
:TI: GE:	
V3GE	2 SUBL 0.75 0.25
:V: GE:	
V17GE31	2 SUBL 0.354 0.646
:V: GE:	
SR3SN52	SUBL 0.375 0.625
:SR: SN:	
SRSN3	2 SUBL 0.25 0.75
:SR: SN:	
SRSN4	2 SUBL 0.2 0.8
:SR: SN:	
SN3TI2	2 SUBL 3 2
:SN: TI:	
V3SN	2 SUBL 0.205 0.795
:SN: V:	
VSN2	2 SUBL 0.6 0.4
:SN: V:	
SN3ZR5	3 SUBL 5 3 1
:ZR: SN: SN VA:	
> aka eta	
SNZR3_A15	2 SUBL 3 1
:SN ZR: SN ZR:	
V2ZR	2 SUBL 2 1
:V: ZR:	

AL14CA13	2 SUBL 14 13
:AL MG ZN: CA:	
AL3CA8	2 SUBL 3 8
:AL: CA MG:	
AL4CE	2 SUBL 0.8 0.2
:AL: CE:	
AL11CE3	2 SUBL 0.7857 0.2143
:AL: CE:	
AL3CE_H	2 SUBL 0.75 0.25
:AL: CE:	
AL3CE_L	2 SUBL 0.75 0.25
:AL: CE:	
ALCE	2 SUBL 0.5 0.5
:AL: CE:	
ALCE2	2 SUBL 0.3333 0.6667
:AL: CE:	
ALCE3_H	2 SUBL 0.25 0.75
:AL: CE:	
ALCE3_L	2 SUBL 0.25 0.75
:AL: CE:	
AL13CO4	2 SUBL 13 4
:AL: CO:	
AL3CO	2 SUBL 3 1
:AL: CO:	
AL5CO2	2 SUBL 5 2
:AL: CO:	
AL9CO2	2 SUBL 9 2
:AL: CO:	
AL3LA	2 SUBL 3 1
:AL: LA:	
ALLA	2 SUBL 1 1
:AL: LA:	
AL53LA22	2 SUBL 0.707 0.293
:AL: LA:	
AL11LA3_LT	2 SUBL 11 3
:AL: LA:	
AL11LA3_HT	2 SUBL 11 3
:AL: LA:	
AL2LI3	2 SUBL 2 3
:AL MG: LI:	
ALLI2	2 SUBL 1 2
:AL: LI:	
AL4LI9	2 SUBL 4 9
:AL: LI:	
ALLI_B32	2 SUBL 1 1
:AL LI MG: LI MG VA:	
AL3SC	2 SUBL 1 3

:SC: AL:	
AL2SC	2 SUBL 1 2
:SC: AL:	
ALSC	2 SUBL 1 1
:SC: AL:	
ALSC2	2 SUBL 2 1
:SC: AL:	
AL3HF4	2 SUBL 3 4
:AL: HF:	
LI22SI5	2 SUBL 22 5
:LI: SI:	
LI13SI4	2 SUBL 13 4
:LI: SI:	
LI7SI3	2 SUBL 7 3
:LI: SI:	
LI12SI7	2 SUBL 12 7
:LI: SI:	
AG9CA2	2 SUBL 0.818182 0.181818
:AG: CA:	
AG7CA2	2 SUBL 0.777778 0.222222
:AG: CA:	
AG2CA	2 SUBL 0.666667 0.333333
:AG: CA:	
AGCA3	2 SUBL 0.25 0.75
:AG: CA:	
AG4SC	2 SUBL 0.8 0.2
:AG: SC:	
AGSC	2 SUBL 0.5 0.5
:AG: SC:	
AL4C4SI	3 SUBL 4 1 4
:AL: SI: C:	
AL8C7SI	3 SUBL 8 1 7
:AL: SI: C:	
AL13CR4SI4	3 SUBL 13 4 4
:AL: CR: SI:	
> Al-Cr-Si, tao 1	
AL9CR3SI	3 SUBL 9 3 1
:AL: CR: SI:	
> Al-Cr-Si, tao 2	
ALCRSI_T3	2 SUBL 11 4
:AL SI: CR:	
> Al-Cr-Si, tao 3	
ALCRSI_T4	3 SUBL 58 31.5 10.5
:AL: CR: SI:	
> Al-Cr-Si, tao 4, AL58CR32SI11	
AL23CUFE4	3 SUBL 23 1 4
:AL: CU: FE:	

AL62CU25FE13	
3 SUBL 0.125 0.255 0.62	
:FE: AL CU: AL:	
AL7CU2FE	3 SUBL 1 2 7
:FE NI: CU: AL:	
> Solution phase of the ternary compound	
Al7Cu2Fe	
AL10CU10FE	3 SUBL 1 10 10
:FE: AL CU: AL:	
AL2CULI	3 SUBL 0.5 0.25 0.25
:AL: CU: LI:	
> Al-Cu-Li ternary phase, i.e. T1	
ALCULI_T2	3 SUBL 0.57 0.11
0.32	
:AL: CU: LI:	
> Al-Cu-Li ternary phase, T2	
ALCULI_R	3 SUBL 0.55 0.117 0.333
:AL: CU: LI:	
> Al-Cu-Li ternary phase, R	
ALCULI_B	3 SUBL 0.6 0.32 8E-2
:AL: CU: LI:	
> Al-Cu-Li ternary phase, TB	
Q_PHASE	3 SUBL 7 3 6
:AL: CU: MG:	
> Al7Cu3Mg6, Al-Cu-Mg ternary phase	
S_PHASE	3 SUBL 2 1 1
:AL SI: CU: MG:	
> Solution phase based on Al2CuMg	
S_PRIME	3 SUBL 2 1 1
:AL: CU: MG:	
> metastable precipitate, related to	
S_PHASE	
T_PHASE	4 SUBL 26 6 48 1
:MG: AL MG: AL CU MG ZN: AL:	
> Solution (Al,Cu,Zn)49Mg32, stable in Al-Mg-Zn, Al-Cu-Mg, Al-Cu-Mg-Zn	
AL28CU4MN7	3 SUBL 28 7 4
:AL: MN: CU:	
AL11CU5MN3	3 SUBL 11 3 5
:AL: MN: CU:	
ALCU3MN2	3 SUBL 1 2 3
:AL: MN: CU:	
AL7CU4NI	2 SUBL 1 1
:AL: FE CU NI VA:	
ALCUSC_TAU	3 SUBL 0.6154 0.3077
7.69E-2	
:AL CU: AL CU: SC:	

AL9FENI	2 SUBL 9 2
:AL: FE NI:	
AL10FE3NI	2 SUBL 5 2
:AL: FE NI:	
AL71FE5NI24	3 SUBL 0.71 5E-2
0.24	
:AL: FE: NI:	
ALFESI_T1	2 SUBL 5 3
:AL SI: FE:	
> Al-Fe-Si ternary phase, tao 1 / tao 9	
ALFESI_T2	4 SUBL 0.5 0.2 0.1
0.2	
:AL: FE: SI: AL SI:	
> Al-Fe-Si ternary phase, tao 2,	
gamma_AlFeSi	
ALFESI_T3	3 SUBL 0.56 0.24 0.2
:AL: FE: SI:	
> Al-Fe-Si ternary phase, AL56FE24SI10,	
tao 3	
ALFESI_T4	4 SUBL 0.4166 0.1667 0.25 0.1667
:AL: FE: SI: AL SI:	
> Al-Fe-Si ternary phase, tao 4,	
delta_AlFeSi	
AL8FE2SI	4 SUBL 0.6612 0.19 4.96E-2 9.92E-2
:AL: FE: SI: AL SI:	
> solution of the Al-Fe-Si ternary phase,	
tao 5, alpha_AlFeSi	
AL9FE2SI2	4 SUBL 0.598 0.152 0.1 0.15
:AL: FE: SI: AL SI:	
> Al-Fe-Si ternary phase, tao 6, aka	
Al5FeSi, beta_AlFeSi	
ALFESI_T7	2 SUBL 3 1
:AL SI: FE:	
> Al-Fe-Si ternary phase, AL9FE5SI6, tao 7	
ALFESI_T8	2 SUBL 2 1
:AL SI: FE:	
> Al-Fe-Si ternary phase, AL2FE3SI4, tao 8	
ALFESI_T10	3 SUBL 0.6 0.25 0.15
:AL: FE: SI:	
> Al-Fe-Si ternary phase, AL60FE25SI15, tao	
10	
ALFESI_T11	3 SUBL 0.65 0.25 0.1
:AL: FE: SI:	
> Al-Fe-Si ternary phase, AL85FE30SI15, tao	
11	
ALFEZN_GAMMA	2 SUBL 0.255 0.745

:AL FE ZN: ZN:	
> Al-Fe-Zn ternary phase, aka gamma 2, no	
detailed structure	
ALLIMG_T	3 SUBL 0.53 0.33 0.14
:AL: LI: MG:	
ALLISI	3 SUBL 0.333333 0.333333
0.333334	
:LI: AL: SI:	
AL3LI8Si5	3 SUBL 0.5 0.1875 0.3125
:LI: AL: SI:	
ALLI5Si2	3 SUBL 0.6625 8.75E-2 0.25
:LI: AL: SI:	
AL18MG3MN2	3 SUBL 18 3 2
:AL: MG: MN:	
ALMG3Ni2	3 SUBL 1 2 3
:AL: NI: MG:	
> Ternary phase AlMg3Ni2, cF96, Fd-3m,	
Ti2Ni type	
BETA_PRIME	2 SUBL 1.8 1
:MG: SI:	
> metastable precipitate, Mg9Si5/Mg1.8Si,	
related to Mg2Si	
B_PRIME	3 SUBL 3 9 7
:AL: MG: SI:	
> metastable precipitate, B_Prime, Al-	
containing Pre-beta phase	
U1_AL2MGSi2	3 SUBL 2 1 2
:AL: MG: SI:	
> metastable precipitate, U1_Al2MgSi2,	
Al-containing Pre-beta phase	
U2_AL4MG4Si4	3 SUBL 1 1 1
:AL: MG: SI:	
> metastable precipitate, U2_Al4Mg4Si4,	
Al-containing Pre-beta phase	
BETA_DPRIME	3 SUBL 2 5 4
:AL SI: MG: SI:	
> metastable beta double prime, related	
to Mg2Si, Mg5Si6, Al2Mg5Si4	
ETA_PRIME	3 SUBL 3 2.5 3.5
:AL: MG: ZN:	
> metastable precipitate, related to	
MgZn2-based Eta phase	
T_PRIME	3 SUBL 0.3 0.4 0.3
:AL: MG: ZN:	
> metastable precipitate, related to	
T_PHASE	
AL31MN6Ni2	3 SUBL 31 6 2

:AL: MN: NI:	> Orthorhombic, ternary Al-Mn-Ni phase
AL2MN2SI3	3 SUBL 2 2 3
:AL: MN: SI:	> the Al-Mn-Si ternary phase, tao1
AL5MN6SI7	3 SUBL 5 6 7
:AL: MN: SI:	> the Al-Mn-Si ternary phase, tao2

AL1MN1SI1	3 SUBL 1 1 1
:AL: MN: SI:	
> the Al-Mn-Si ternary phase, tao3	
AL3MNSI2	3 SUBL 3 1 2
:AL: MN: SI:	
> the Al-Mn-Si ternary phase, tao4	
AL3MN4SI2	3 SUBL 3 4 2
:AL: MN: SI:	
> the Al-Mn-Si ternary phase, tao5	
ALMNSI_T6	2 SUBL 4 1
:AL MN: SI:	
> the Al-Mn-Si ternary phase, tao6	
ALMNSI_T8	4 SUBL 6 3 3 1
:AL: MN: AL MN SI: AL SI:	
> the Al-Mn-Si ternary phase, tao 8	
AL15SI2M4	4 SUBL 16 4 1 2
:AL: FE MN: SI: AL SI:	
> Solution of Al-Mn-Si ternary phase, tao 9, Al15(Mn,Fe)3Si2	
AL2MNSI3	3 SUBL 2 1 3
:AL: MN: SI:	
> the Al-Mn-Si ternary phase, tao10	
AL24MN5ZN	3 SUBL 5 1 2 4
:MN ZN: ZN: AL:	
AL9MN2ZN	3 SUBL 2 1 9
:MN: ZN: AL:	
AL11MN3ZN2	3 SUBL 3 2 1 1
:MN: ZN: AL:	
ALNI2SI	2 SUBL 1 1
:AL SI VA: NI:	
AL6NI3SI	3 SUBL 6 3 1
:AL: NI: SI:	
ALNI16SI9	3 SUBL 1 16 9
:AL: NI: SI:	
ALNI2ZN	3 SUBL 0.25 0.5 0.25
:AL: NI: ZN:	
AL13NI38ZN49	3 SUBL 0.13 0.38 0.49
:AL: NI: ZN:	
> Al-Ni-Zn ternary phase	
CU16MG6SI7	3 SUBL 16 6 7
:CU: MG: SI:	
CU3MG2SI	3 SUBL 2.74 2 1.26
:CU: MG: SI:	
CU5MN4SI	3 SUBL 0.5 0.37 0.13
:CU: MN: SI:	
CUMNZN	3 SUBL 0.334 0.333 0.333
:CU: MN: ZN:	

CU6NISI3	2 SUBL 0.732 0.268
:CU NI: SI:	
CU46NI25SI29	3 SUBL 0.458 0.25 0.292
:CU: NI: SI:	
FE5NI3SI2	2 SUBL 4 1
:FE NI: SI:	
ZN13M2	2 SUBL 1 6.5
:FE NI: ZN:	
MG6MN3NI	
3 SUBL 0.5 0.1666667 0.3333333	
:MG: MN: NI:	
MGNI6SI6	3 SUBL 1 6 6
:MG: NI: SI:	
MG2NI3SI	3 SUBL 2 3 1
:MG: NI: SI:	
MG10NI55SI35	3 SUBL 2 11 7
:MG: NI: SI:	
MG2NI16SI11	3 SUBL 1 8 5.5
:MG: NI: SI:	
MG5NI9SI	3 SUBL 1 1.8 0.2
:MG: NI: SI:	
MG9NI29SI16	3 SUBL 9 29 16
:MG: NI: SI:	
MGNI6ZN6	4 SUBL 3 4 1 2
:MG ZN: MG NI ZN: NI: ZN:	
MN15NI45SI40	3 SUBL 0.15 0.45 0.4
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T1 or N	
MN15NI50SI35	3 SUBL 0.15 0.5 0.35
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T2 or PHI	
MN6NI16SI7	
3 SUBL 0.206897 0.551724 0.241379	
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T3 or G	
MN1NI1SI1	3 SUBL 1 1 1
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T4 or E	
MNNISI_T5	2 SUBL 1 2
:MN: NI SI:	
> Mn-Ni-Si ternary phase, T5 or "tao 1"	
MNNISI_T6	2 SUBL 1 2
:MN: NI SI:	
> Mn-Ni-Si ternary phase, T6 or "tao 2"	
MN3NI2SI	3 SUBL 3 2 1
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T7 or Omega	

MN2NISI	2 SUBL 3 1
:MN NI: SI:	
> Mn-Ni-Si ternary phase, T8 or S	
MN6NISI3	3 SUBL 0.61 0.12
0.27	
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T9 or R	
MN66NI4SI30	3 SUBL 0.66 4E-2 0.3
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T10 or U	
MN52NI29SI19	3 SUBL 0.52 0.29
0.19	
:MN: NI: SI:	
> Mn-Ni-Si ternary phase, T11 or W	
MN7NI7ZN86	3 SUBL 7E-2 7E-2
0.86	
:MN: NI: ZN:	
NI2SIZN_T1	3 SUBL 0.5 0.25 0.25
:NI: SI: ZN:	
NI9SI2ZN_T2	3 SUBL 0.75 0.1675
8.25E-2	
:NI: SI: ZN:	
NI2SIZN3_T3	3 SUBL 2 1 3
:NI: SI: ZN:	
NISIZN_T4	3 SUBL 3 2 1
:NI: SI: ZN:	
> Ni-Si-Zn tao 4, Ni ₃ Si ₂ Zn1	
Q_ALCUMGSI	4 SUBL 5 2 8 6
:AL: CU: MG: SI:	
> Quaternary phase, aka Q,	
Al ₅ Cu ₂ Mg ₈ Si ₆ , Al ₃ Cu ₂ Mg ₉ Si ₇ &	
Al ₄ Cu ₂ Mg ₈ Si ₇	
AL18FE2MG7SI10	4 SUBL 18 2 7 10
:AL: FE: MG: SI:	
> Quaternary phase, aka Al ₈ FeMg ₃ Si ₆ and	
Q_PHI/H_PHASE	



TCAL4

(This page is intended to be blank)

Thermo-Calc Software AB
Norra Stationsgatan 93
SE-113 64 Stockholm
SWEDEN

info@thermocalc.com
Phone: +46 8 545 959 30
Fax: +46 8 673 37 18
www.thermocalc.com

Org.No: 556540-6138
VAT No: SE556540613801